



Carbon Management and CO₂ Conversion

Josh Schaidle
Whole Value Chain CCUS
September 23, 2025

NREL at a Glance

4,026 Workforce, including:

- 2,968 regular/limited term
- 508 contingent workers
- 228 postdoctoral researchers
- 155 graduate student interns
- 167 undergraduate student interns

—as of 6/9/2025

World-class research expertise in:

- Energy Systems Integration
- Transportation and Fuels
- Buildings and Industry

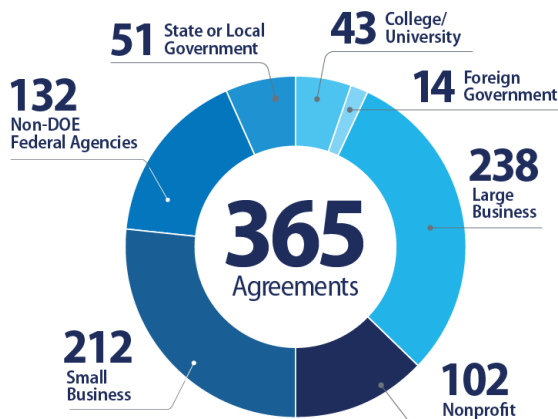
Partnerships with:

- Industry
- Academia
- Government

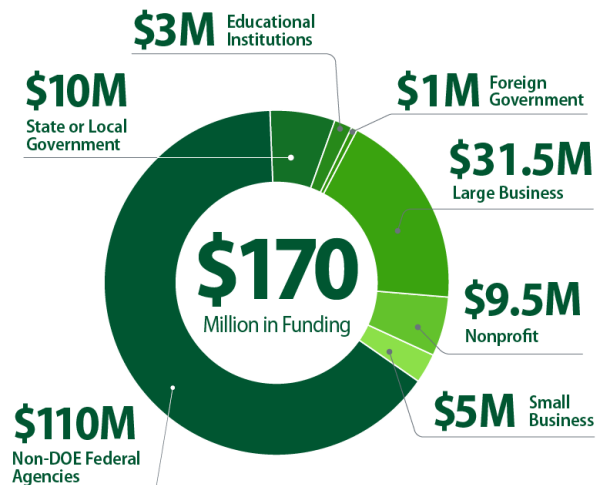
4 Campuses operate as living laboratories



More Than 1,100 Active Partnerships in FY 2024



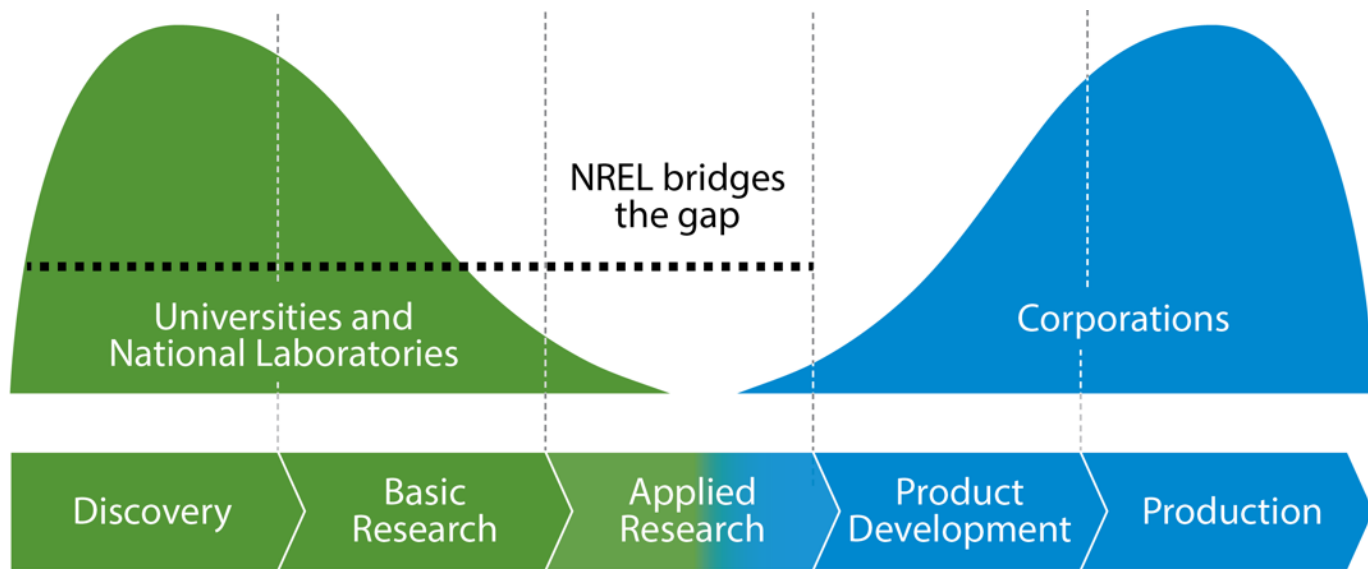
Agreements by Business Type



Funding by Business Type

*Due to agreements involving multiple partners, the number of partners exceeds the 365 new agreements executed in FY24.

We Reduce Risk in Bringing Innovations to Market



NREL helps bridge the gap from basic science to commercial application

Forward-thinking innovation yields disruptive and impactful results to benefit the entire U.S. economy

Accelerated time to market delivers advantages to American businesses and consumers

NREL Brings Distinct Capabilities

Foundational Science

Bench-scale- discovery



Solar Energy Research Facility
Science and Technology Facility
Field Test Laboratory Building



Accelerated Technology Scale-Up

Scaling R&D and Process Engineering



Energy Materials and Processing
at Scale (Completion 2027)



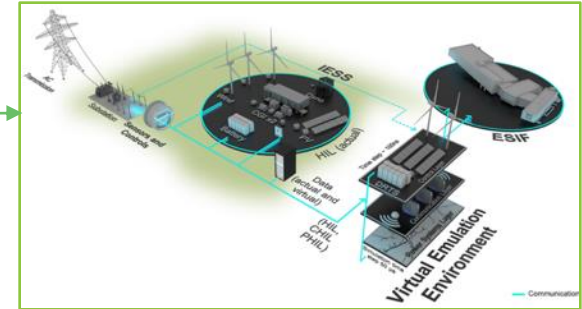
Energy Systems
Integration Facility



- Carbon-free H₂
- Grid and security tech
- Products from electrochemical processes and CO₂
- Advanced Batteries
- PV, Wind, Water Power, Geothermal
- New Buildings and Industrial Materials, Manufacturing and Systems

Systems

R&D with Industry Partners



**Advanced Research on
Integrated Energy Systems
(ARIES)**

Markets

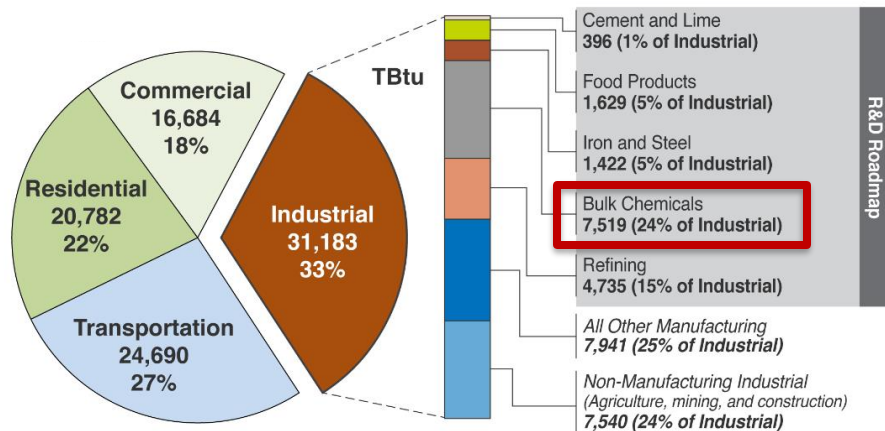
High-Performance Computing, Simulation, and Visualization



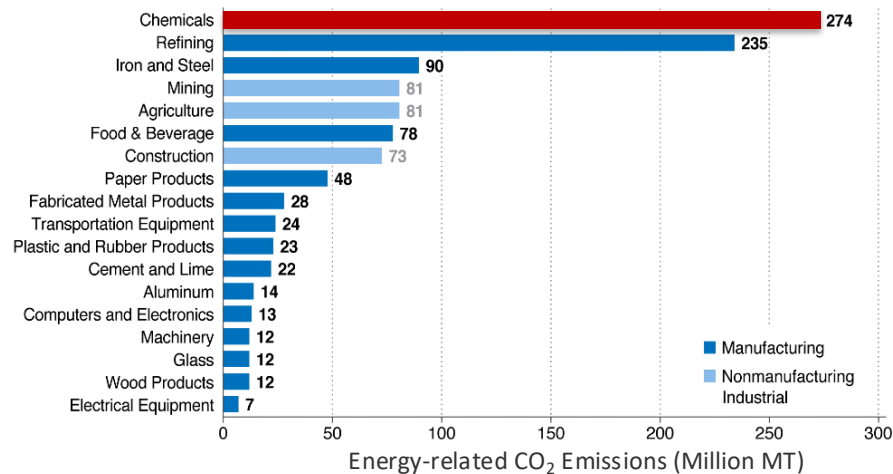
CO₂ Conversion

Energy Demand and Emissions in Industry

Energy Consumption



CO₂ Emissions

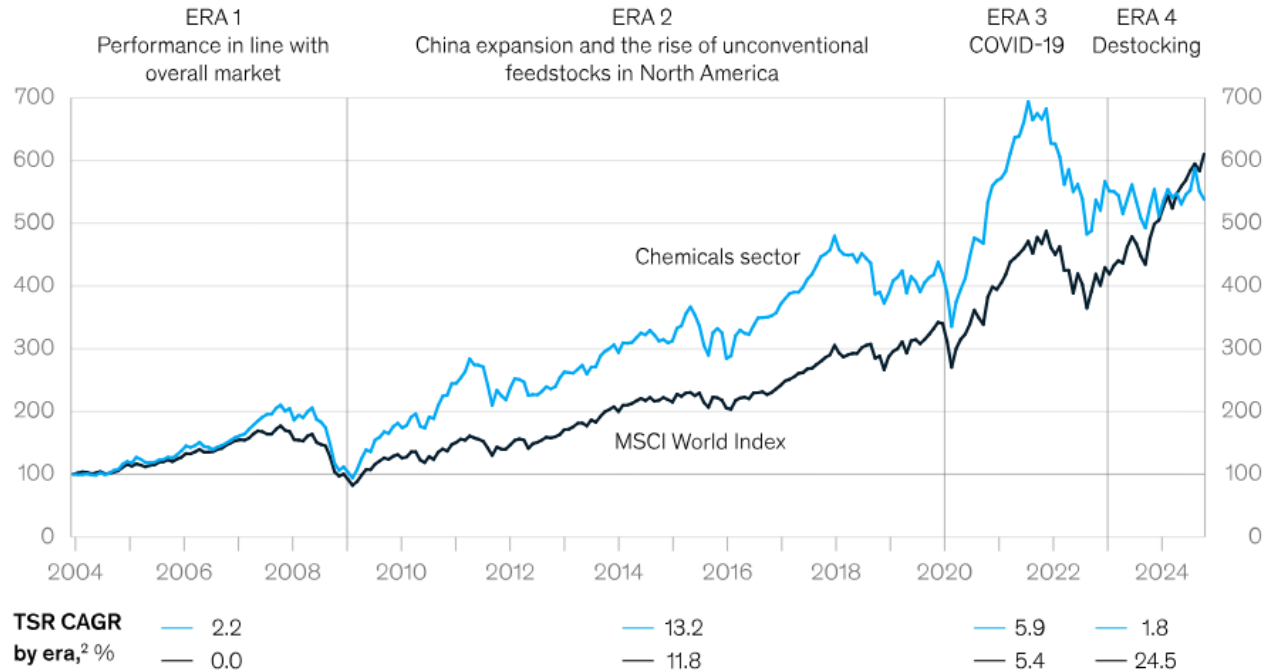


The US industrial sector consumes one third of US primary energy use and emits 30% of the total U.S. energy-related CO₂ emissions to deliver 10% of our GDP

Challenge

Lagging Returns in the Chemicals Industry

Chemicals sector TSR compared with MSCI World Index, 2003–24,¹ index (100 = Dec 2003)



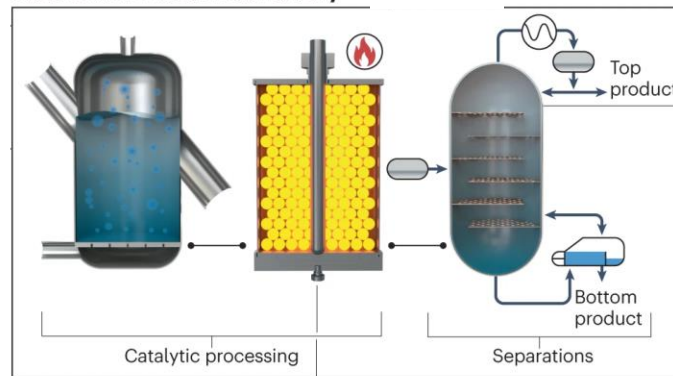
¹Weighted mean of TSR, year over year. Chemicals sector sample size (n = 673 companies).
Source: Corporate Performance Analytics by McKinsey; S&P Global Market Intelligence

Key drivers:

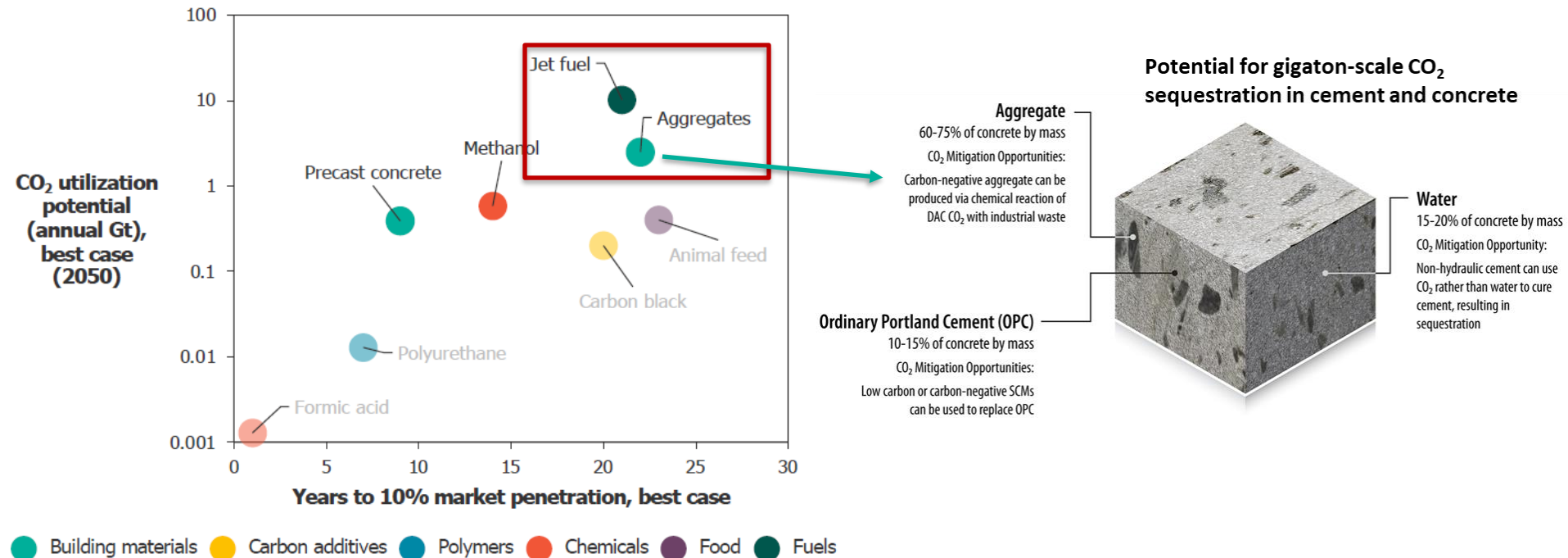
- High energy prices
- Supply chain disruptions
- Interest rates and inflation
- Capacity addition exceeding demand growth
- Stalled innovation

Opportunity Chemical Manufacturing Innovation

State of current chemical industry

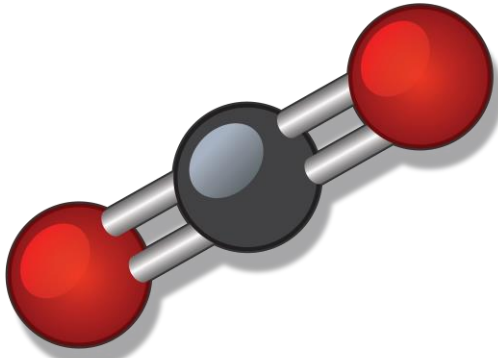


Utilizing CO₂ could be a \$1 Trillion Industry



Brutal Reality of CO₂ Conversion to Fuel

High Oxygen
Content



And neither free
nor pure

Requires Lots of
Energy



Limited Pipeline
Infrastructure

American Carbon Alliance



Innovation Opportunity: Reactive CO₂ Capture

Emerging Approach: Reactive Capture of CO₂

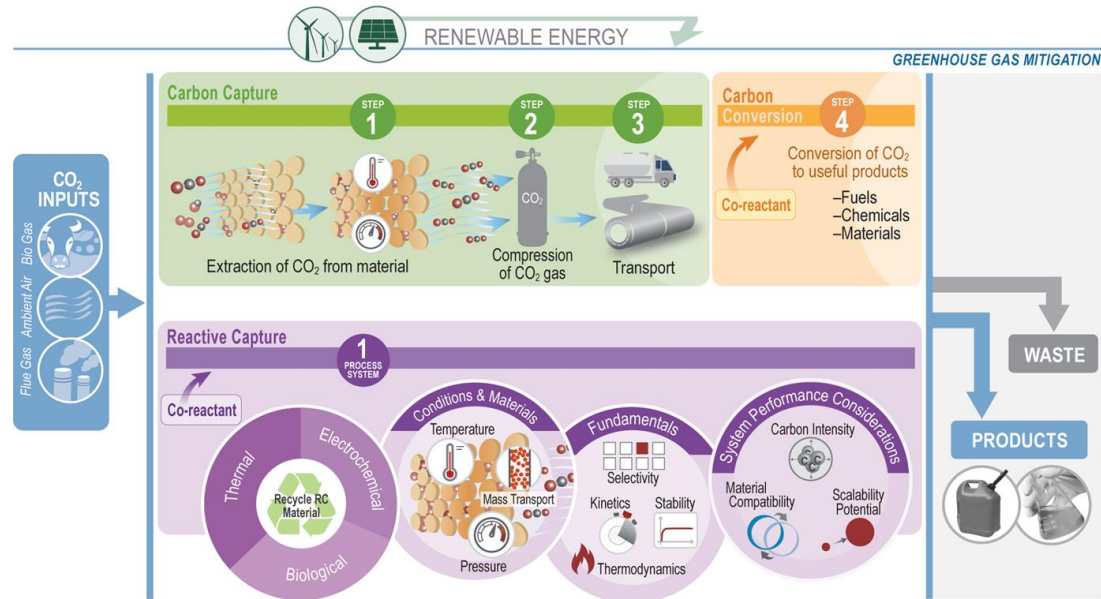
Reactive Capture Definition: The coupled process of capturing CO₂ from a mixed gas stream and converting it into a valuable product *without* going through a purified CO₂ intermediate

Can Include:

- Integration of CO₂ separation and conversion in one step
- Integration of separation and conversion in one unit
- Process intensification

Product Targets:

Form a valuable product, or mixture of products, in a more reduced state than CO₂

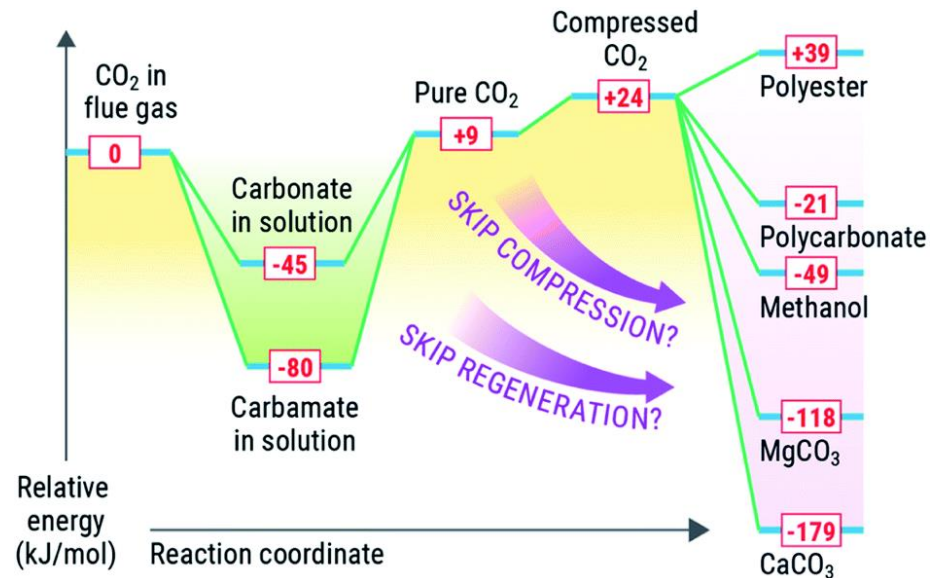


Proposed Value Proposition

Our proposed value proposition for reactive CO₂ capture is:

1. Lower energy intensity
2. Increased capital-expense-normalized productivity (i.e., capital utilization)

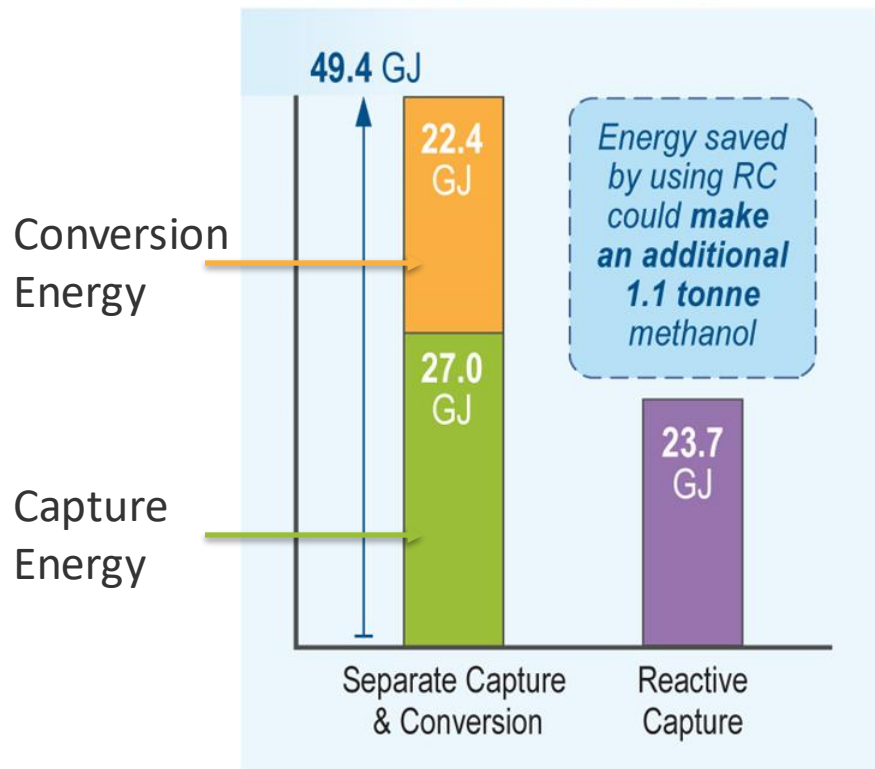
relative to the separate CO₂ capture and conversion processes that require a purified CO₂ intermediate.



D. Heldebrant, et al., *Chem. Sci.* 13 (2022) 2445-6456

Value Proposition: Potential Savings

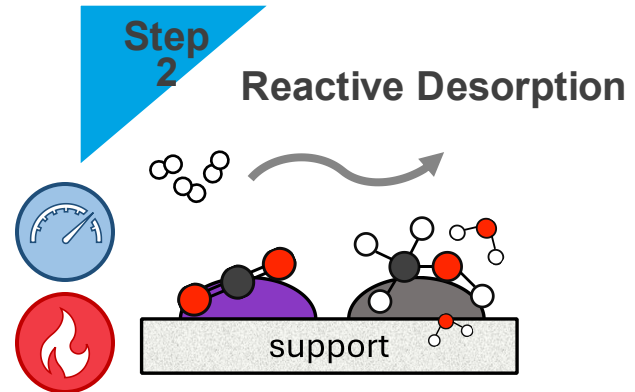
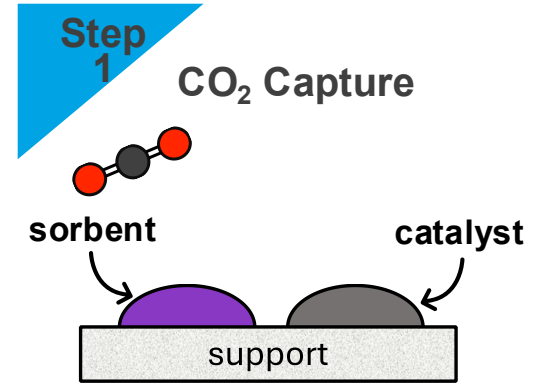
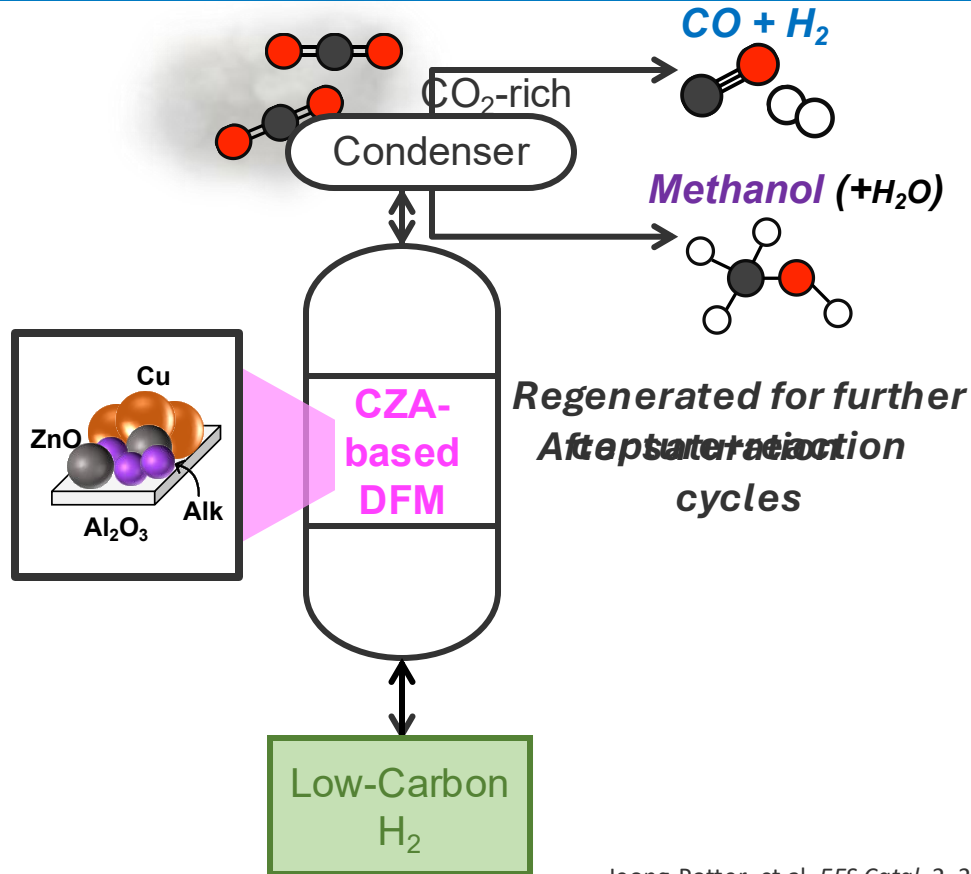
ENERGY REQUIRED TO MAKE
1 TONNE METHANOL PRODUCT



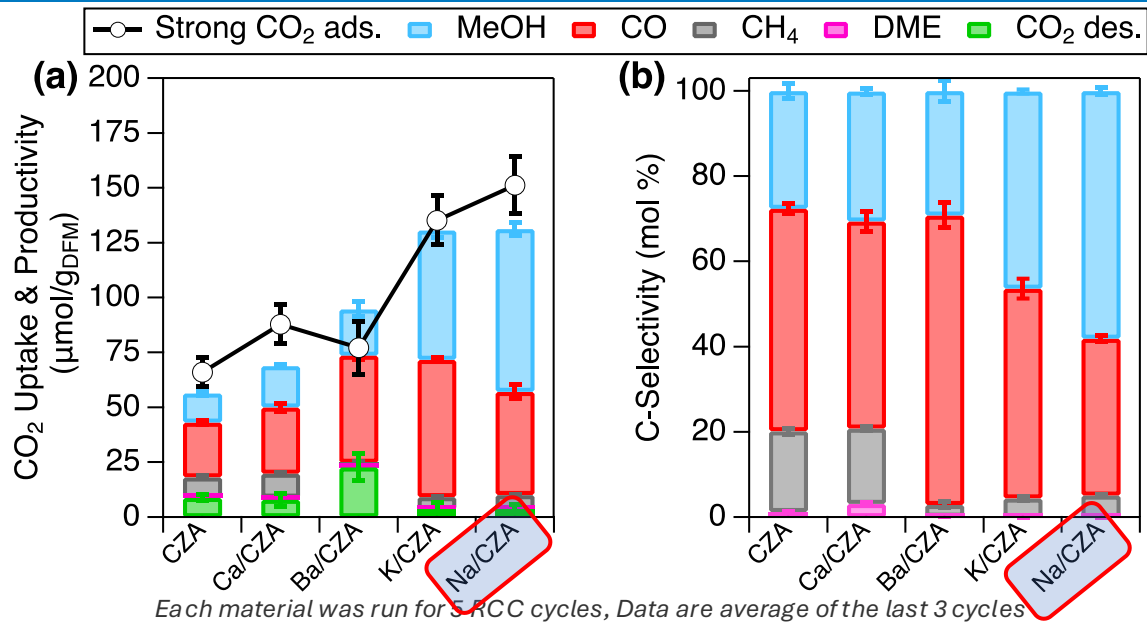
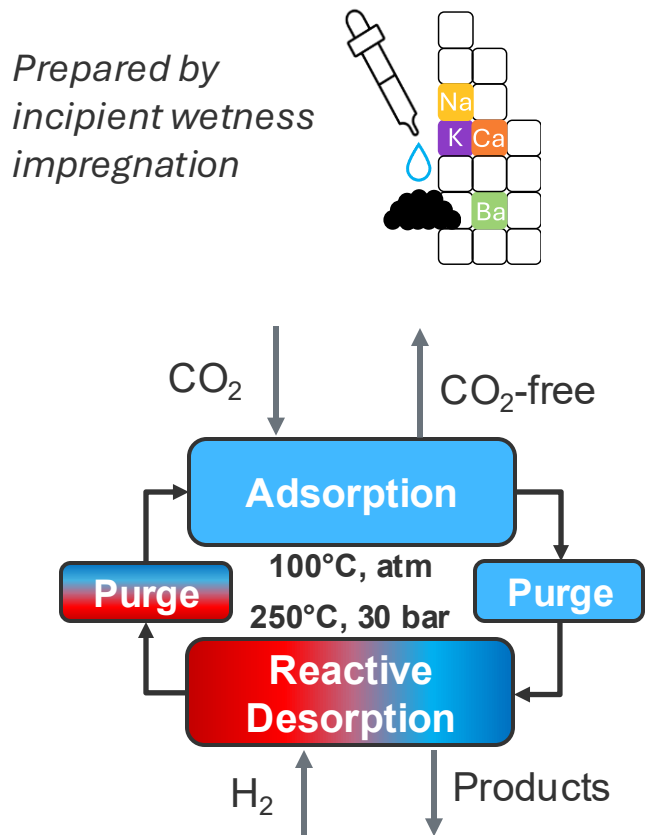
*Calculation assumes same methanol synthesis yield/efficiency regardless of CO₂ purity/captured state

Thermochemical Reactive CO₂ Capture

Reactive capture of CO₂ to MeOH – process concept

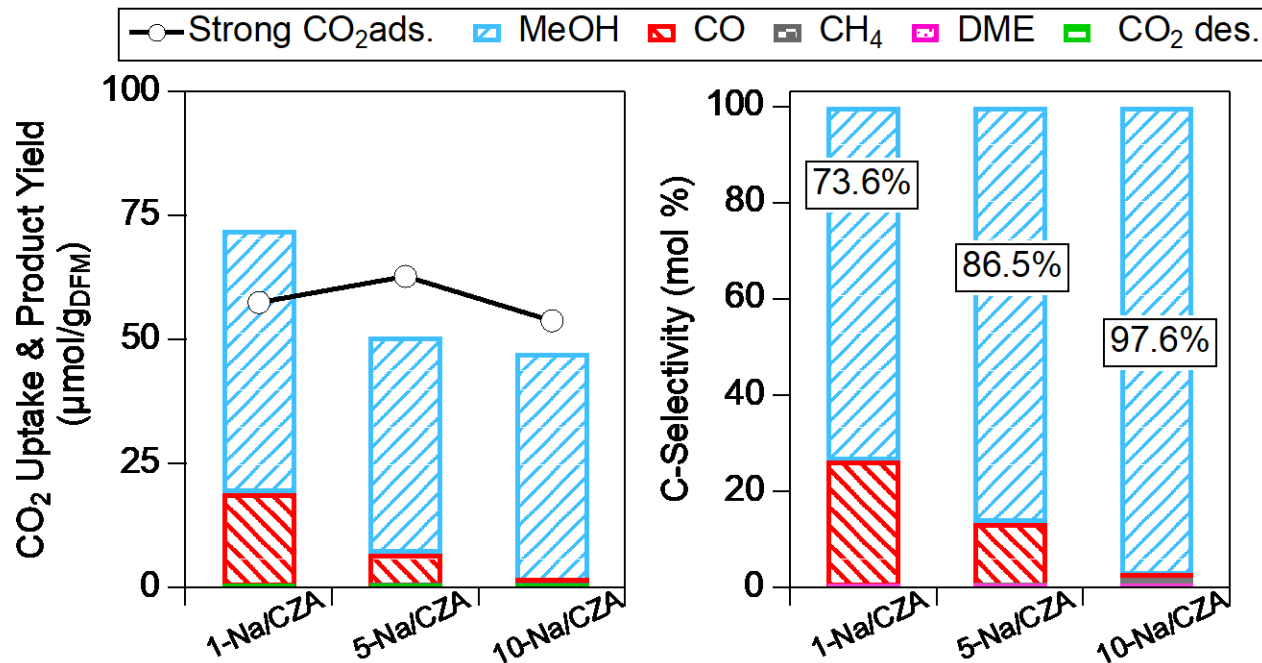


Group 1 Alk-modification provides greatest benefit



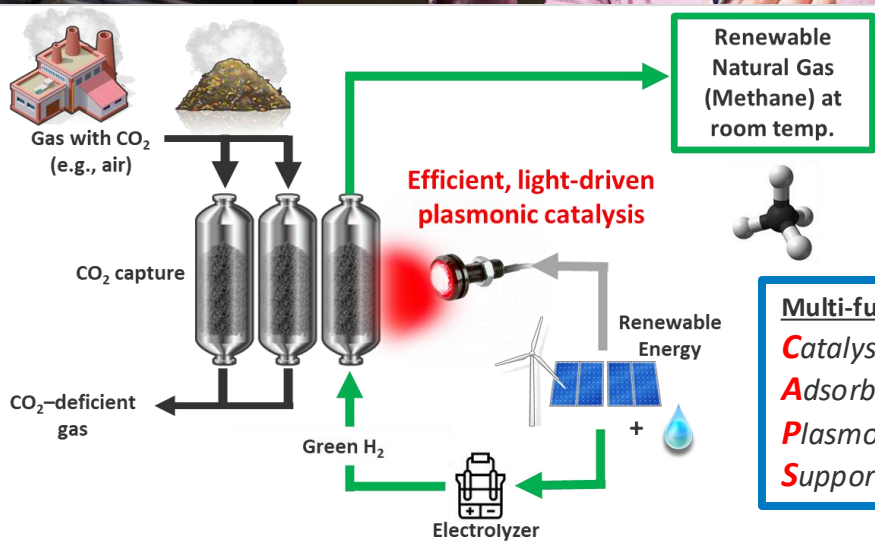
With highest capture capacity, MeOH selectivity and productivity, and low CH₄ selectivity, **K/CZA and Na/CZA are the most promising materials**

Effect of sorbent loading at “best conditions” (200°C, 30 bar)



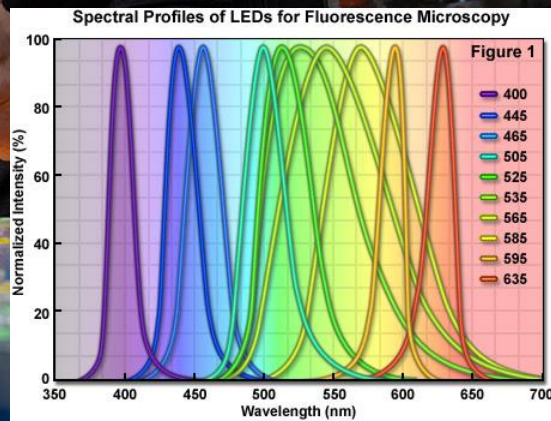
*MeOH selectivity increased from **53% to 97%** with **5% increase** in MeOH productivity compared to starting point (5% Na/CZA, 250°C, 30 bar)*

Photo-induced Reactive CO₂ Capture from the Atmosphere



Multi-functional material

Catalyst
Adsorbent
Plasmonic-antennae
Support



Background

Key Challenges for Reactive DAC

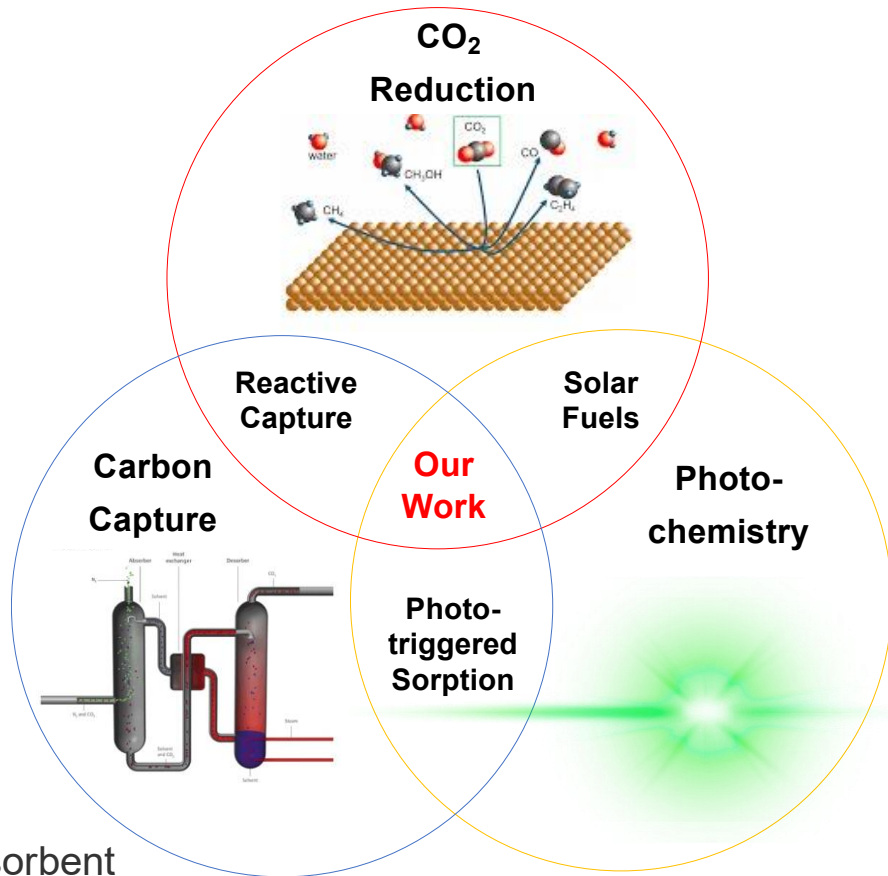
Mismatch of capture vs. catalysis conditions,
sorbent degradation, component oxidation

Hypothesis

Targeted radiation will eliminate need for bulk heating
AND will enable catalysis at lower temperatures
compatible with amine-based capture

Objectives

- (1) Efficient photo-CO₂ release from commercial DAC sorbent
- (2) Integrate CO₂ capture and conversion into a first-of-its-kind photoreactive amine-based DAC system



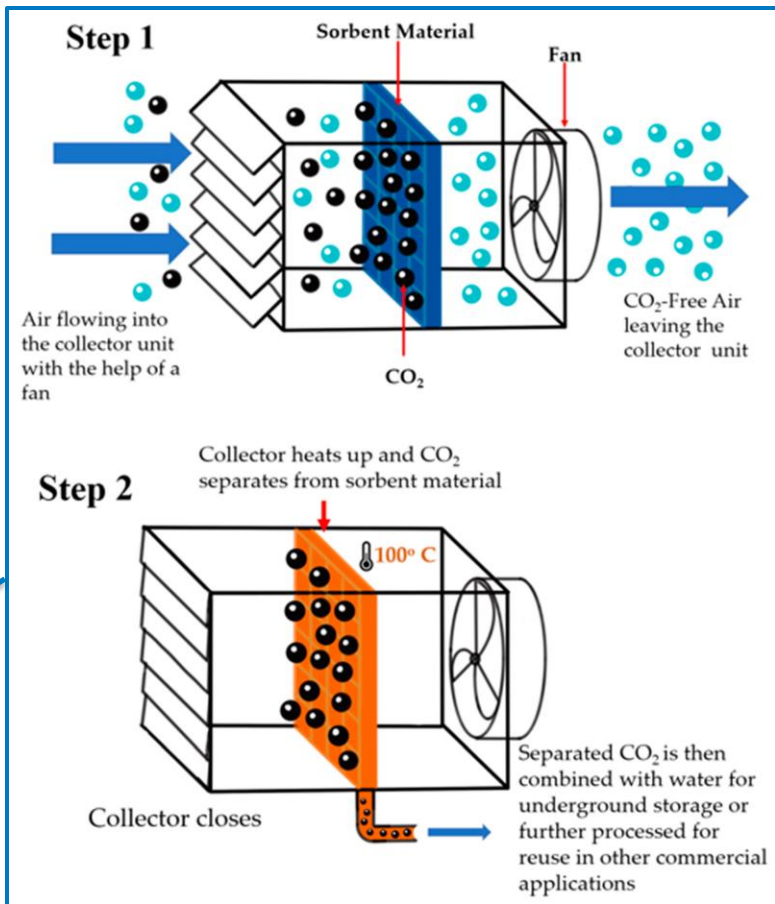
Our work is at the **NEXUS** of CO₂ reduction,
Carbon Capture, and PhotoChemistry

Direct Air Capture (DAC)

Remove dilute streams of CO₂ (400 ppm) from the atmosphere.

The two mainstream technologies are based on:

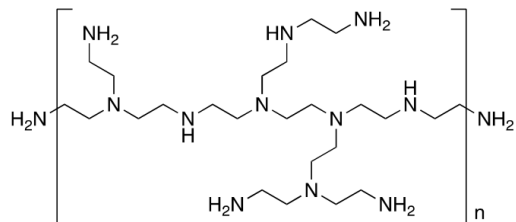
- liquid solvents
- **solid sorbents**



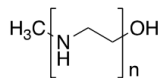
Garza, D.; Dargusch, P.; Wadley, D. A Technological Review of Direct Air Carbon Capture and Storage (DACCS): Global Standing and Potential Application in Australia. *Energies* 2023, 16, 4090. <https://doi.org/10.3390/en16104090>



Polyethyleneimine (PEI) as CO₂ DAC sorbent

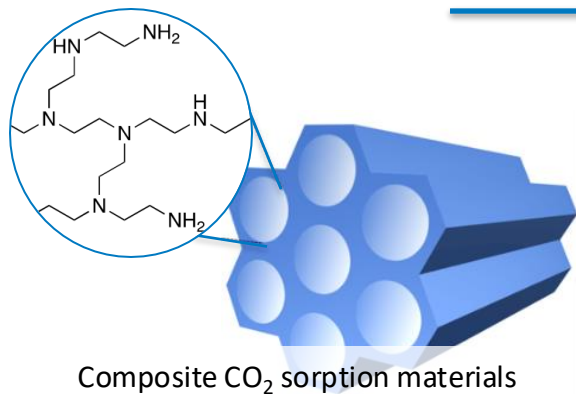


Branched PEI

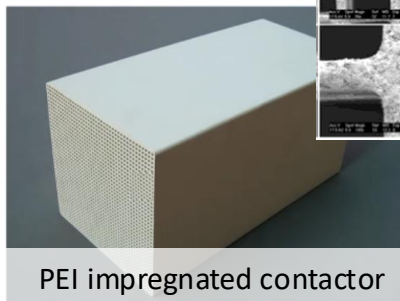


Linear PEI

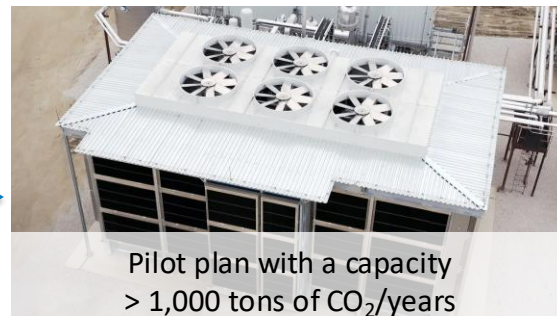
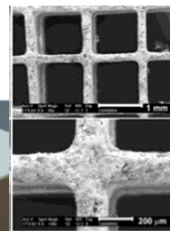
- Adsorption enthalpies ideal for dilute CO₂ streams
- Commercially available
- Low volatility



Composite CO₂ sorption materials



PEI impregnated contactor



Pilot plant with a capacity
> 1,000 tons of CO₂/years

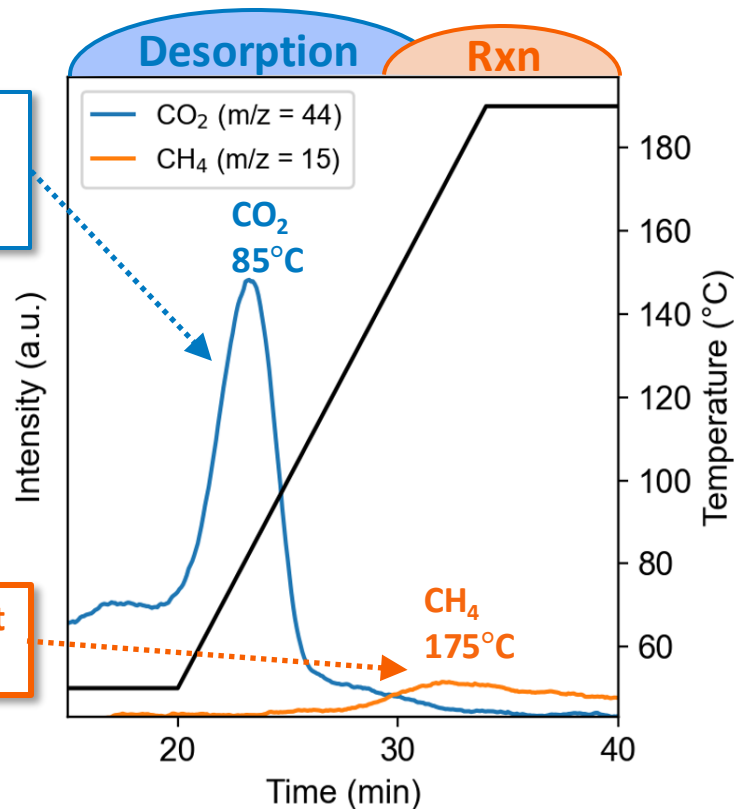
CO₂ Slip in Thermally Driven Reactive Capture with Amines

To date, temperature-driven conversion requires bulk temps. ~175 °C, posing two challenges:

- CO₂ "slip": adsorbed CO₂ desorbs ~85 °C
- -NH_x degradation: amino-polymers start degrading ~120 °C

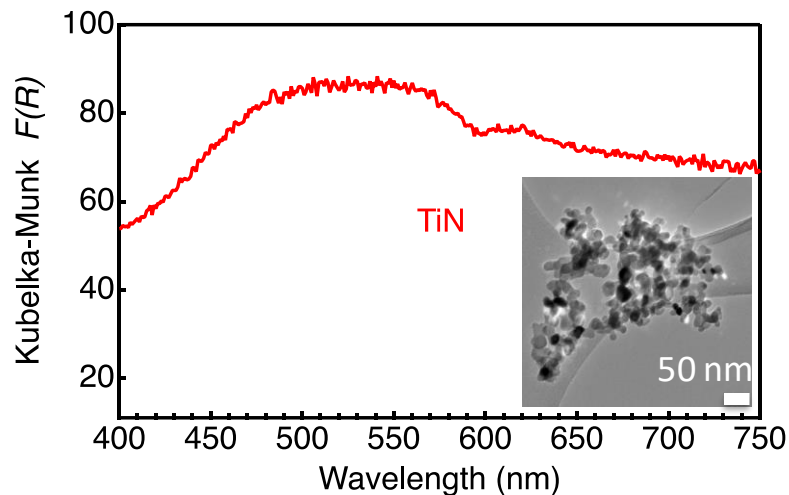
Lots of unreacted CO₂ desorption (95% "CO₂ slip")

Minimal CH₄ product formation (5%)



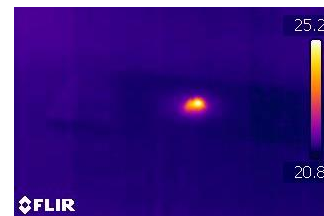
Bulk TiN particles – plasmonic heating

Bulk TiN particles –
UV-Vis absorbance and TEM picture

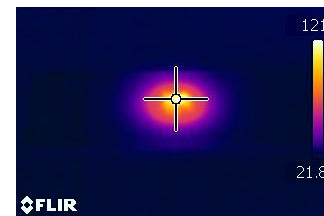


Bulk TiN particles - 10 mg sample on glass slide,
 N_2 atmosphere, 625 nm LED w/ focusing lens

Current to LED (mA)	Measured Irradiance (mW/cm ²)	Bulk Sample Temp. °C Measured @ 2 min.
23	10	26
210	100	57
700	320	120



10 mW/cm²



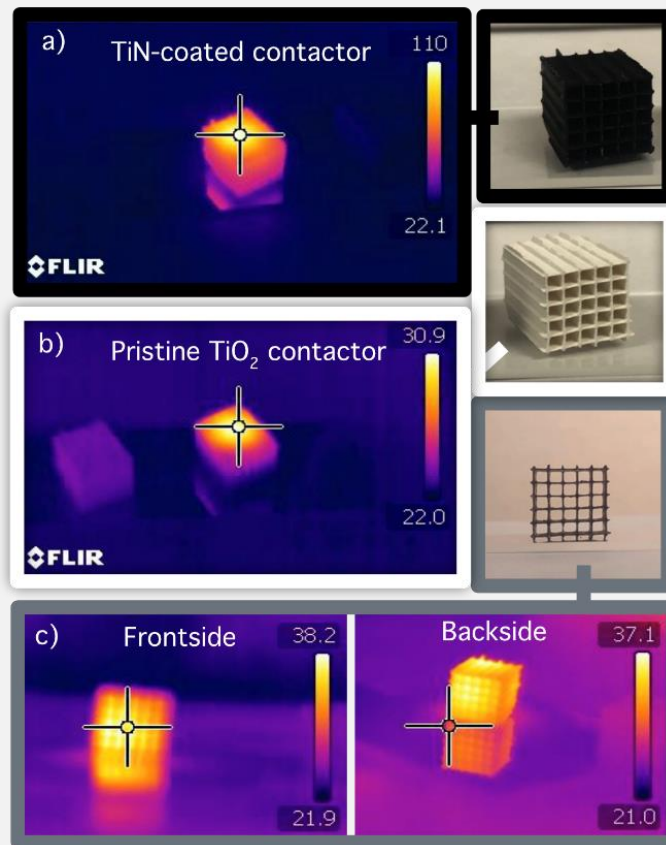
320 mW/cm²

TiN as the photo-responsive material

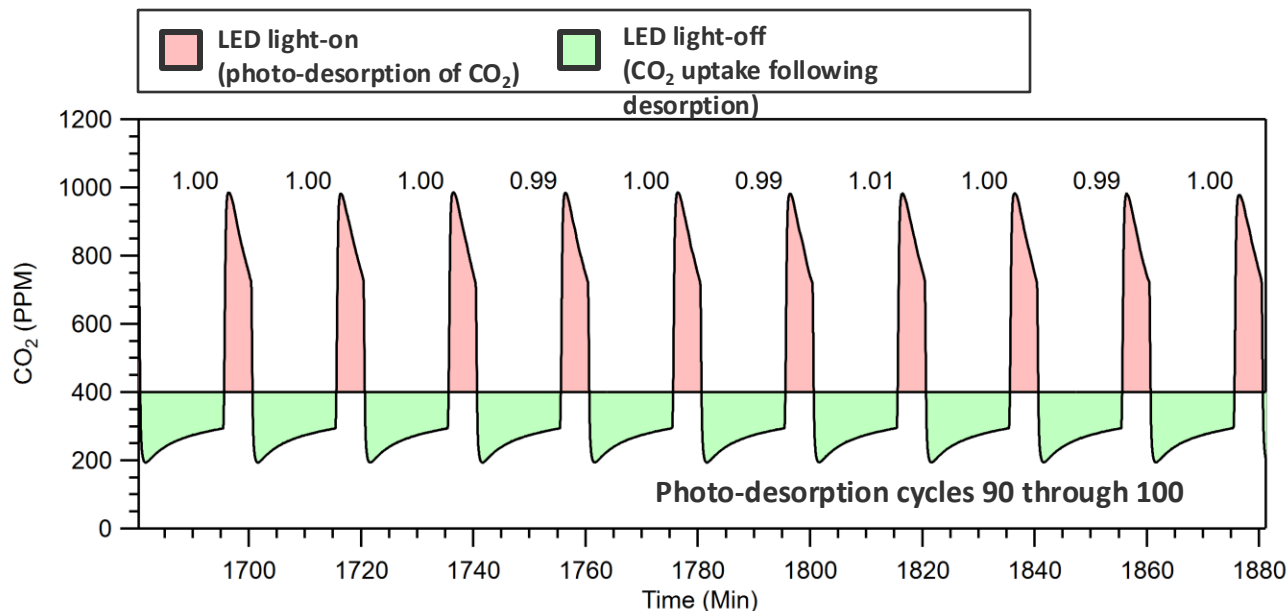
Current to LED (mA)	Measured Irradiance (mW/cm ²)	Bulk Sample Temp. °C Measured @ 2 min.
23	10	26
210	100	46
700	320	94

Photo-induced heating is compatible with amines

2.5 wt% TiN in $\sim 1:1$ PEI:Al₂O₃
N₂ atmosphere, 625 nm set to 700 mA LED w/ focusing lens

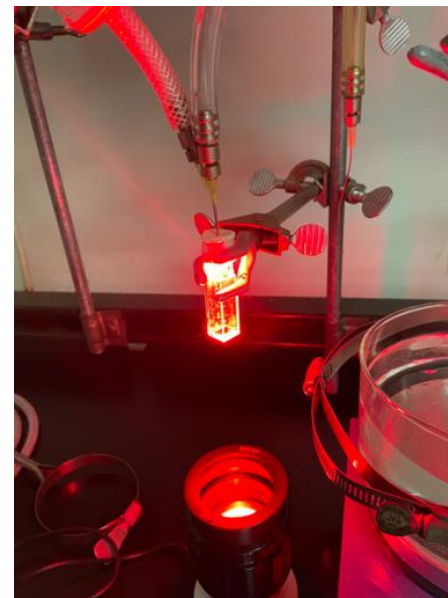


Demonstrated Photo-thermal Swing Adsorption



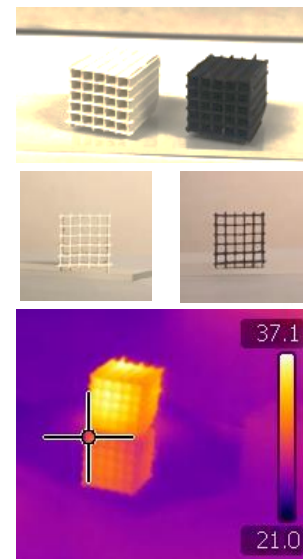
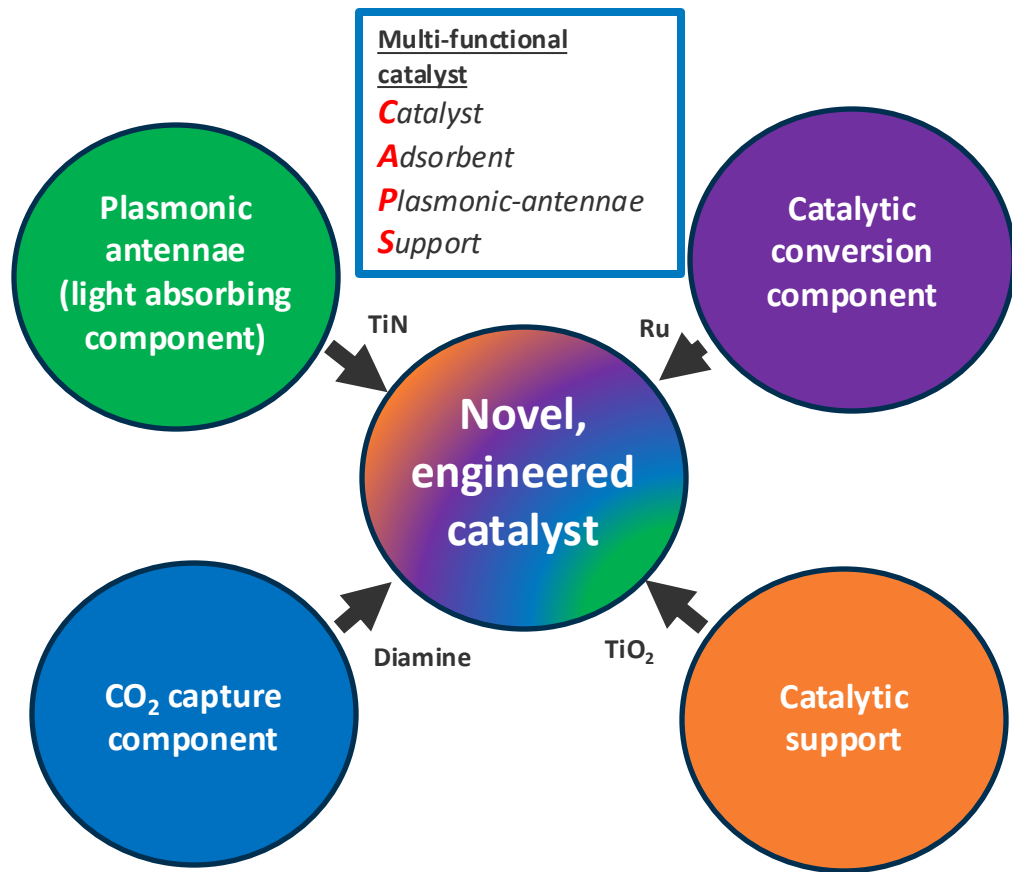
- Efficient photo-desorption without amine degradation

Material: Amine + Titanium Nitride + Monolith Contactor
Conditions: 400 ppm CO₂ constant flow
Irradiance with 625 nm LED light



Use electrified LED lighting to replace thermal (steam) heating for CO₂ desorption

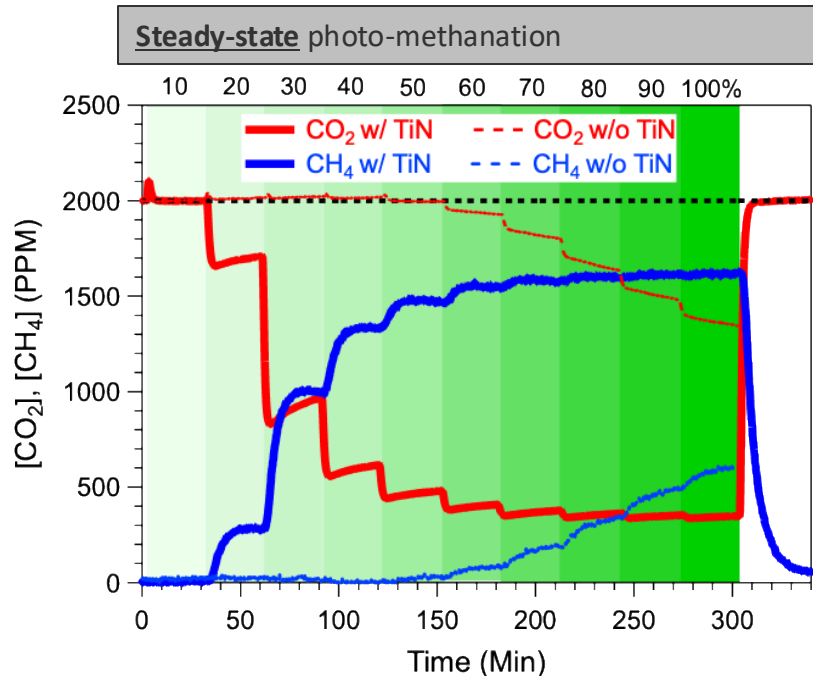
Multi-function “CAPS” material for photo-induced reactive DAC



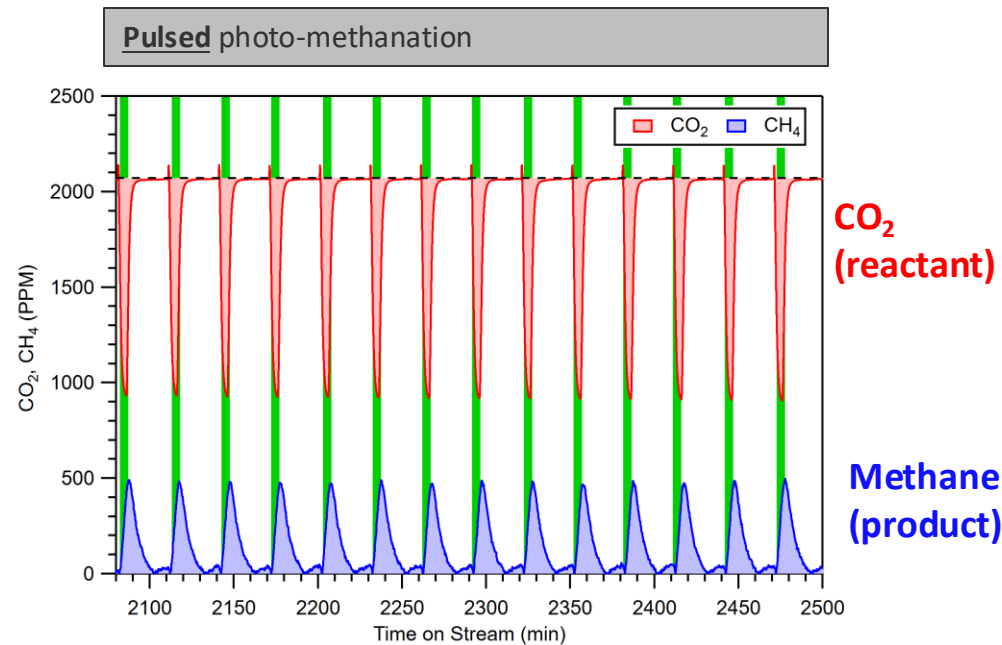
Monolith impregnated with TiN for light absorption



Photo-induced methanation (steady-state and pulsed)

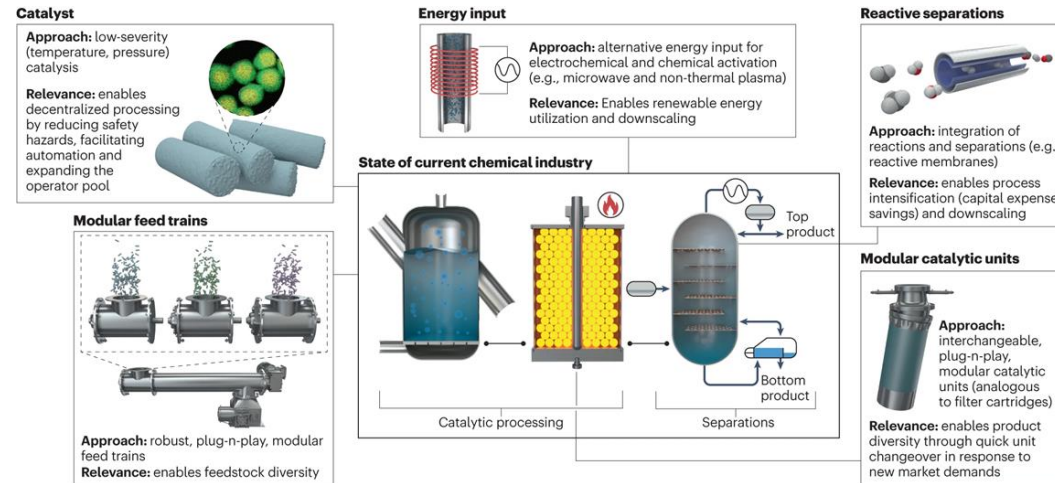


Ru-TiN-diamine-TiO₂



Summary and Conclusions

- The industrial sector is **complex, diverse, and energy intensive**
- **Innovation** can be a differentiator
- **CO₂ utilization** presents a significant future opportunity, but is expensive today
- **Reactive CO₂ capture, coupled with alternative energy input strategies**, is at an early-stage of development and provides a means to reduce cost and energy intensity



Key Contributors



Noemi
Leick



Mat
Rasmussen



Sawyer
Halingstad



Isabel
Shim



Mia
Martinsen



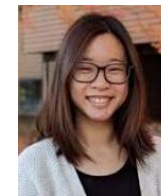
Wade
Braunecker



Simon
Pang



Nathan
Ellebracht



Alvina
Aui



Sneha
Akhade



Michael
Griffin



James
Crawford



Brittney
Petel



Chae
Jeong-Potter



Matt
Yung



Martha
Arellano-Treviño



Melinda
Jue



Wenqin
Li



Thomas
Ludwig



Hannah
Goldstein



Wilson
McNeary



Alexander
Hill



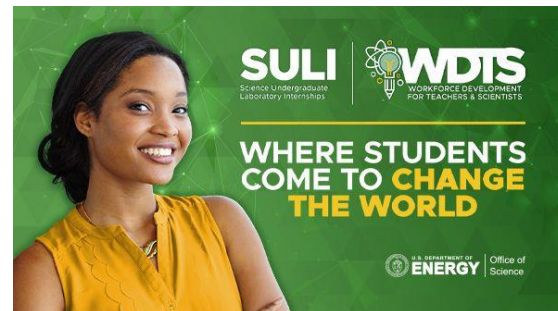
Daniel
Ruddy



Anh To



Acknowledgements



**NREL Laboratory Directed Research
and Development (LDRD) Program**

Let's Discuss!

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