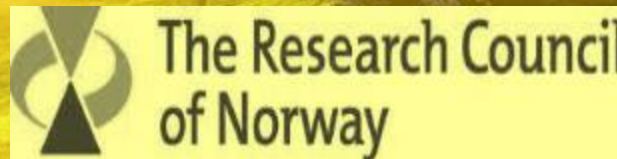


Whole Value Chain CCUS Conference Week

CO₂ Sequestration in Hydrates with Associated Carbon Neutral Methane Gas Production

ConocoPhillips



Norwegian Consulate General
San Francisco



Prof. Arne Graue
Dept. of Physics and Technology
University of Bergen, NORWAY

Whole Value Chain CCUS Conference Week, Sept. 22nd–26th, 2025, Golden, CO, USA.



18 – 20 February 2025
Kuala Lumpur, Malaysia

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Energy for all: Reflect ► Innovate ► Evolve

Host



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IPTC-25104-EA

Net Zero CO₂ Emission Utilizing Energy from Gas Hydrates - Carbon Neutral Methane Production with CO₂ Storage and Conversion to Hydrogen

Arne Graue¹ and Geir Ersland¹

¹Department of Physics and Technology, University of Bergen, Bergen Norway

Sponsoring Societies

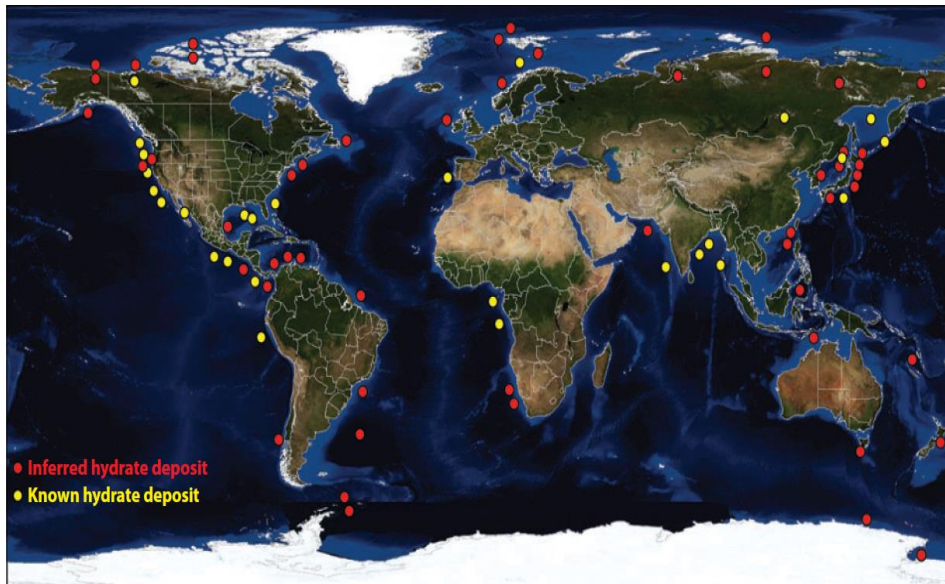


CO₂ Storage in Hydrate Reservoirs with Associated Spontaneous Natural Gas Production

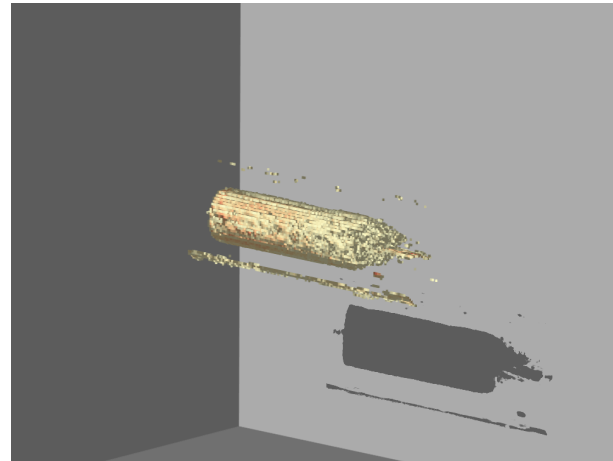
Presentation Overview:

Lab Verification Upscaled to Field Pilot Demonstration of Spontaneous Methane Production When Hydrate is Exposed to CO₂. This technology provides Net Zero CO₂ Emission when Utilizing Energy from Gas Hydrates in Carbon Neutral Methane Production with CO₂ Storage.

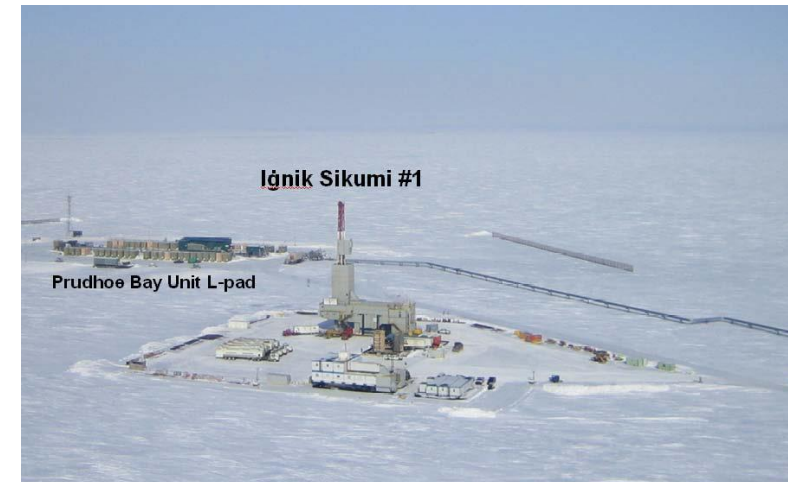
Methane hydrate reservoirs



In-Situ imaging (MRI) of hydrate formation

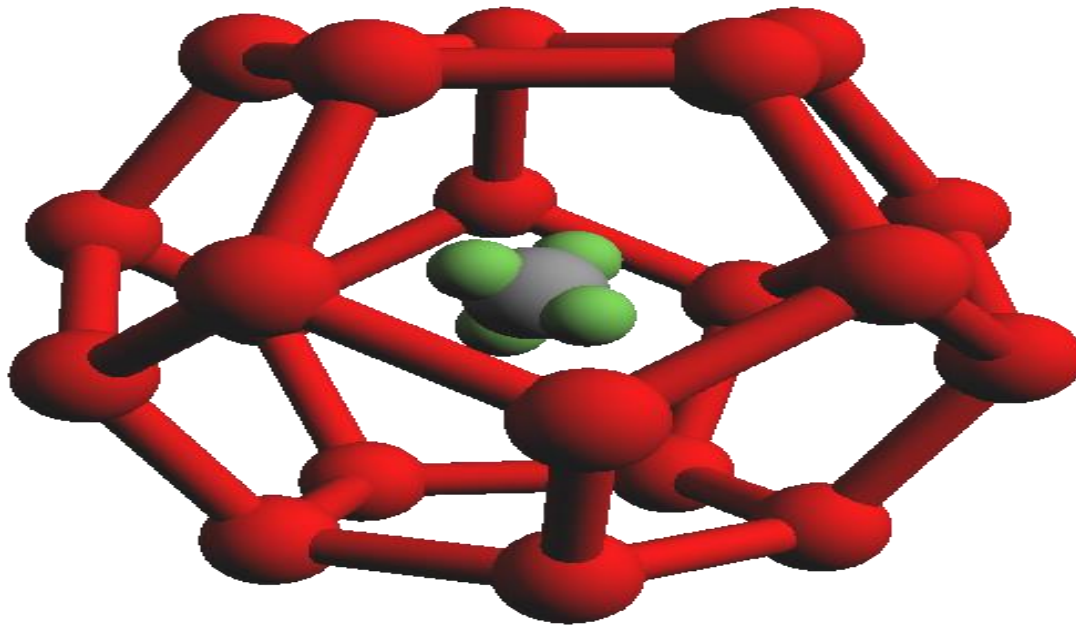


Methane production by CO₂ injection in field test in Alaska 2012



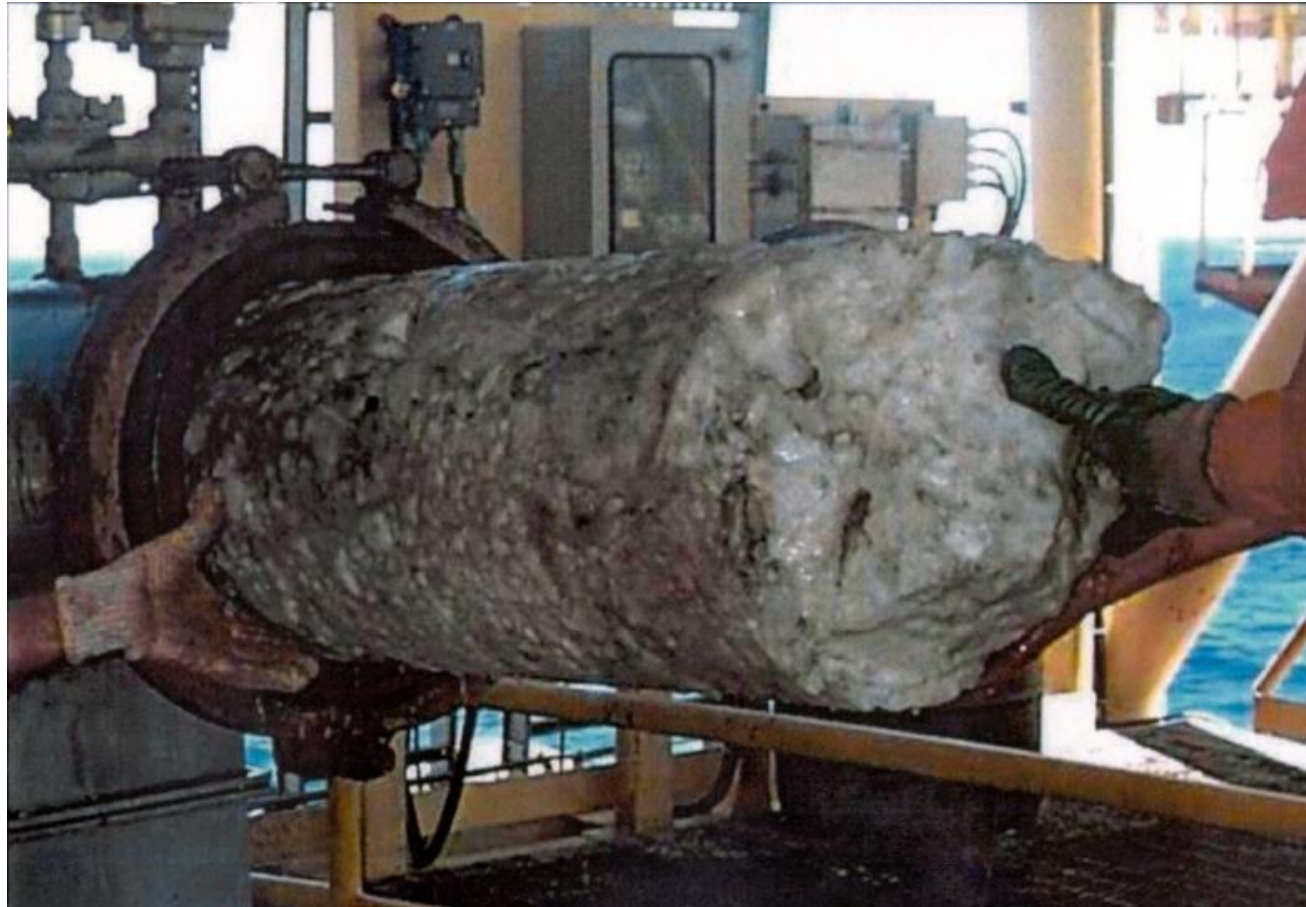
GAS HYDRATES

- Solid state of gas and water where the water molecules form a cavity that encapsulates the guest molecule.



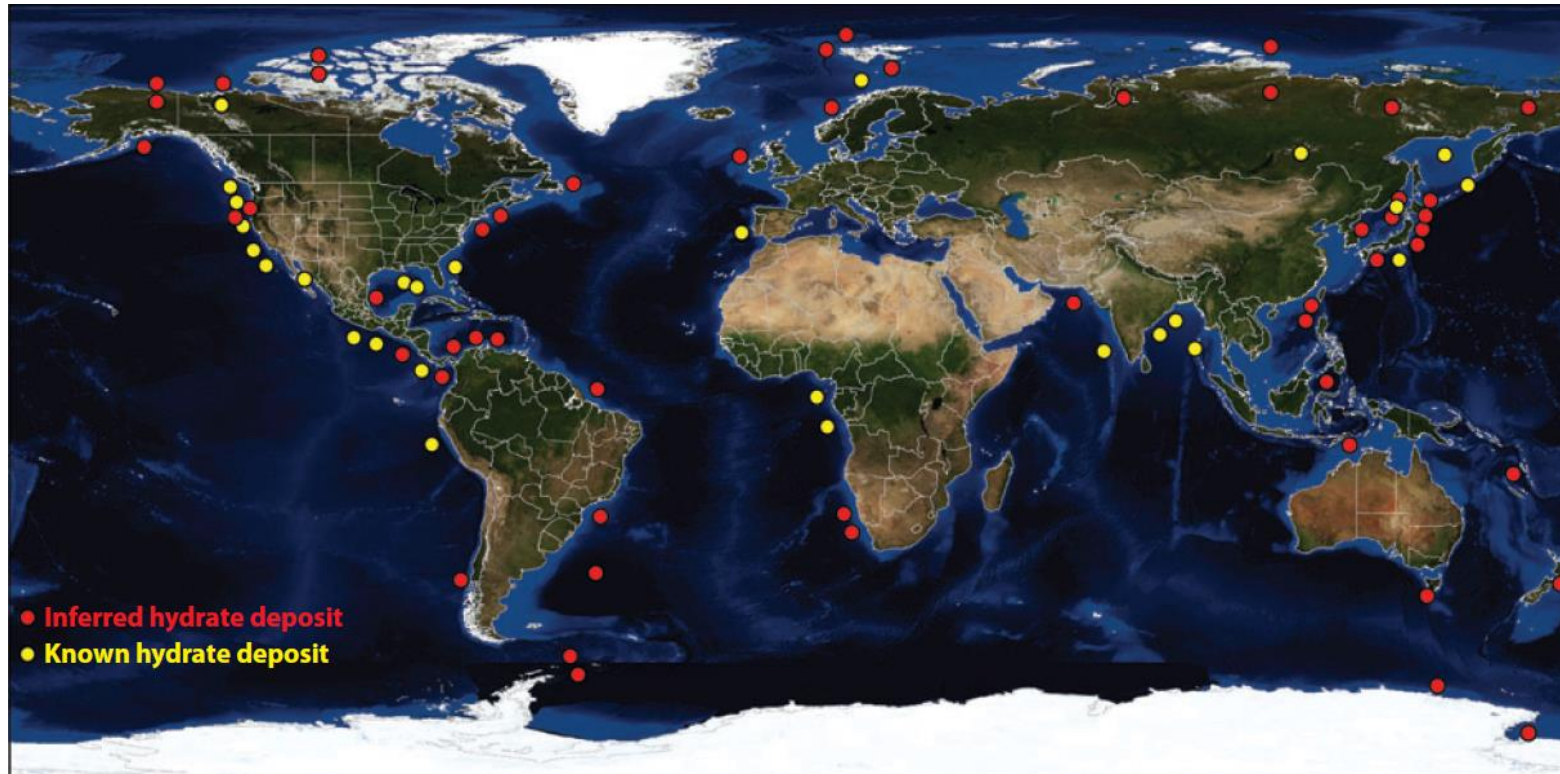
Why are hydrates of interest?

- Initial interest as a curiosity
- Plugging of production and transportation pipelines



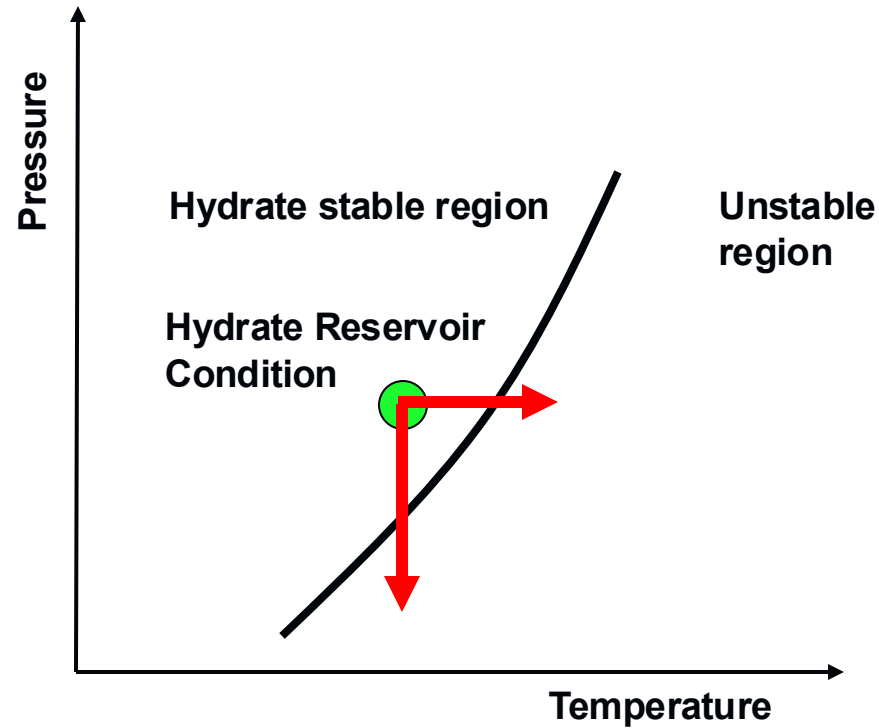
Renewed interest

- Significant amount of energy
 - Permafrost regions
 - Marine environments (high water column)



GAS HYDRATE PRODUCTION METHODS

- Move the gas hydrate outside its stability region
 - Depressurization
 - Thermal stimulation
 - Hydrate inhibitors
- **CO₂ exchange**



CO₂ Exchange: Project Motivation

- **The amount of energy bound in hydrates may be more than twice the world's total energy resources in conventional hydrocarbon reservoirs; i.e. oil-, gas- and coal reserves**
- **Simultaneous CO₂ Sequestration**
- **Win-win situation for gas production**
- **Need no hydrate melting or heat stimulation**
- **Spontaneous process**
- **No associated water production**
- **Formation integrity**

**CO₂ storage in hydrates
with associated methane
gas production**

Challenge:

**Determine exchange mechanisms during potential
sequestration of CO₂ to produce methane from hydrates**



Three component Phase Field Theory

$$F = \int d\mathbf{r} \left\{ \frac{\varepsilon^2 T}{2} (\nabla \phi)^2 + \sum_{i,j=1}^3 \frac{\varepsilon_{i,j}^2 T}{4} (c_i \nabla c_j - c_j \nabla c_i)^2 + f_{bulk}(\phi, c_1, c_2, c_3, T) \right\}$$

$$f_{bulk} = wTg(\phi) + [1 - p(\phi)]f_S(c_1, c_2, c_3, T) + p(\phi)f_L(c_1, c_2, c_3, T)$$

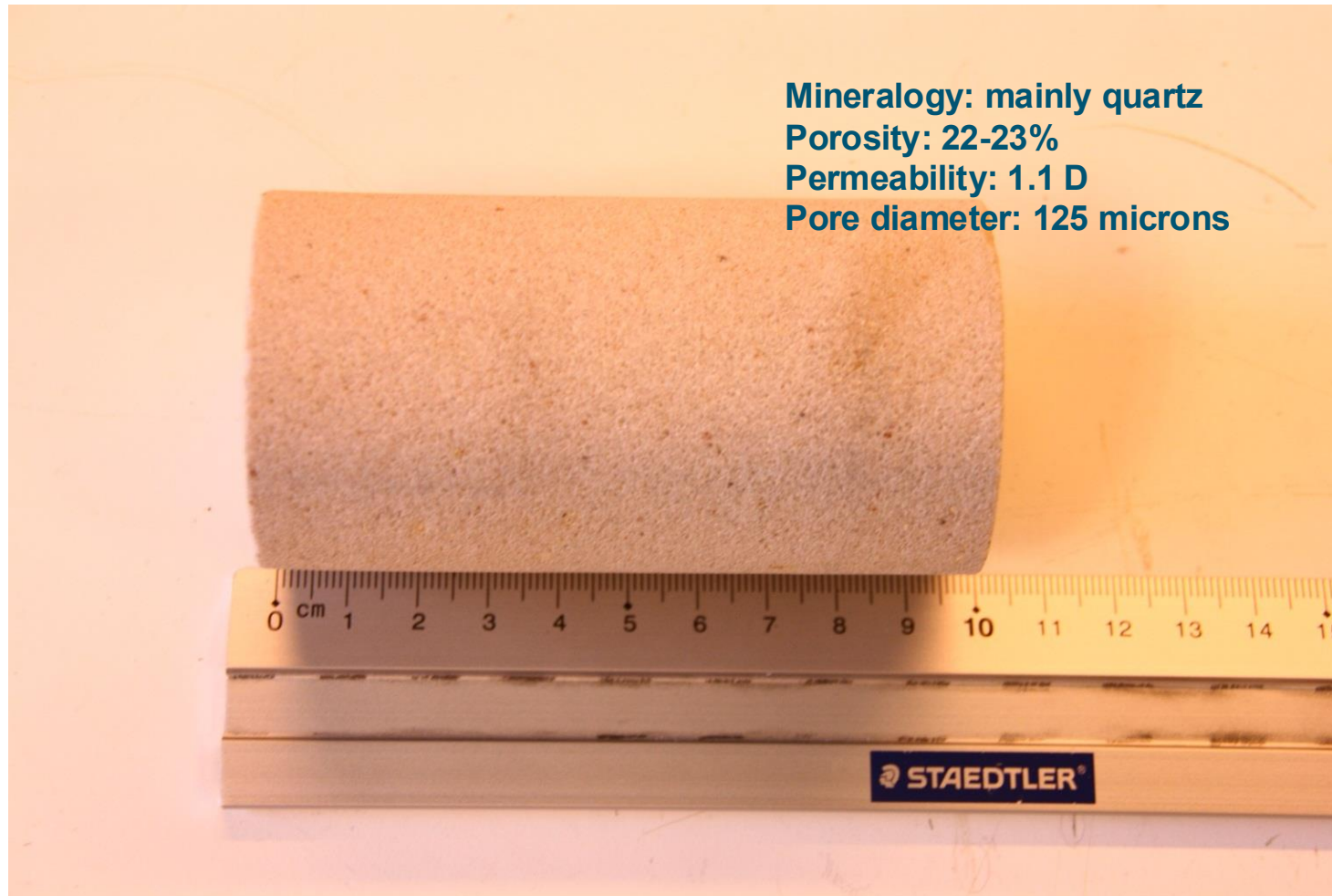
$$\dot{\phi} = -M_{\phi} \frac{\delta F}{\delta \phi} + \zeta_{\phi}$$

$$\sum_{i=1}^3 c_i = 1$$

$$\dot{c}_i = \nabla M_{ci}(c_1, c_2, c_3) \nabla \left(\frac{\delta F}{\delta c_i} - \zeta_i \right)$$

Parameters ε and w can be fixed from the interface thickness and interface free energy. ε_{ij} set equal to ε

- Hydrate reservoirs are often found in porous media
 - Sedimentary rock

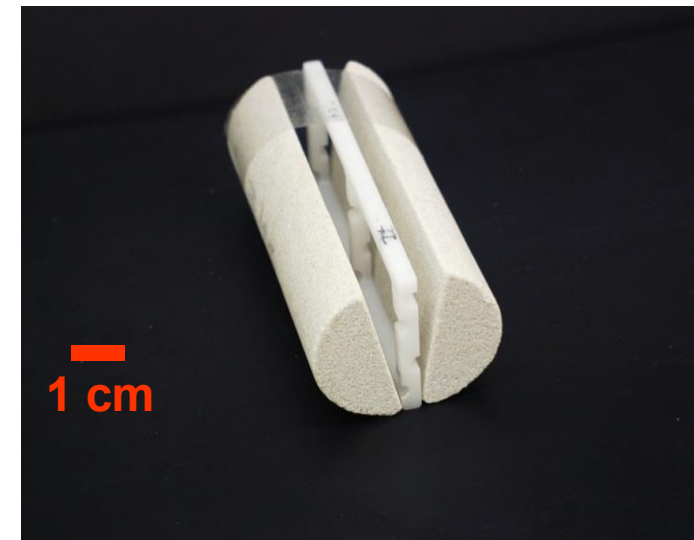
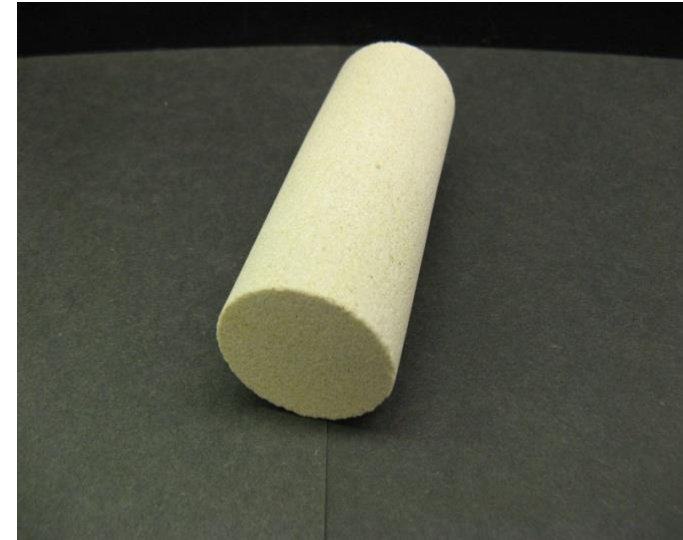


Core Sample Design

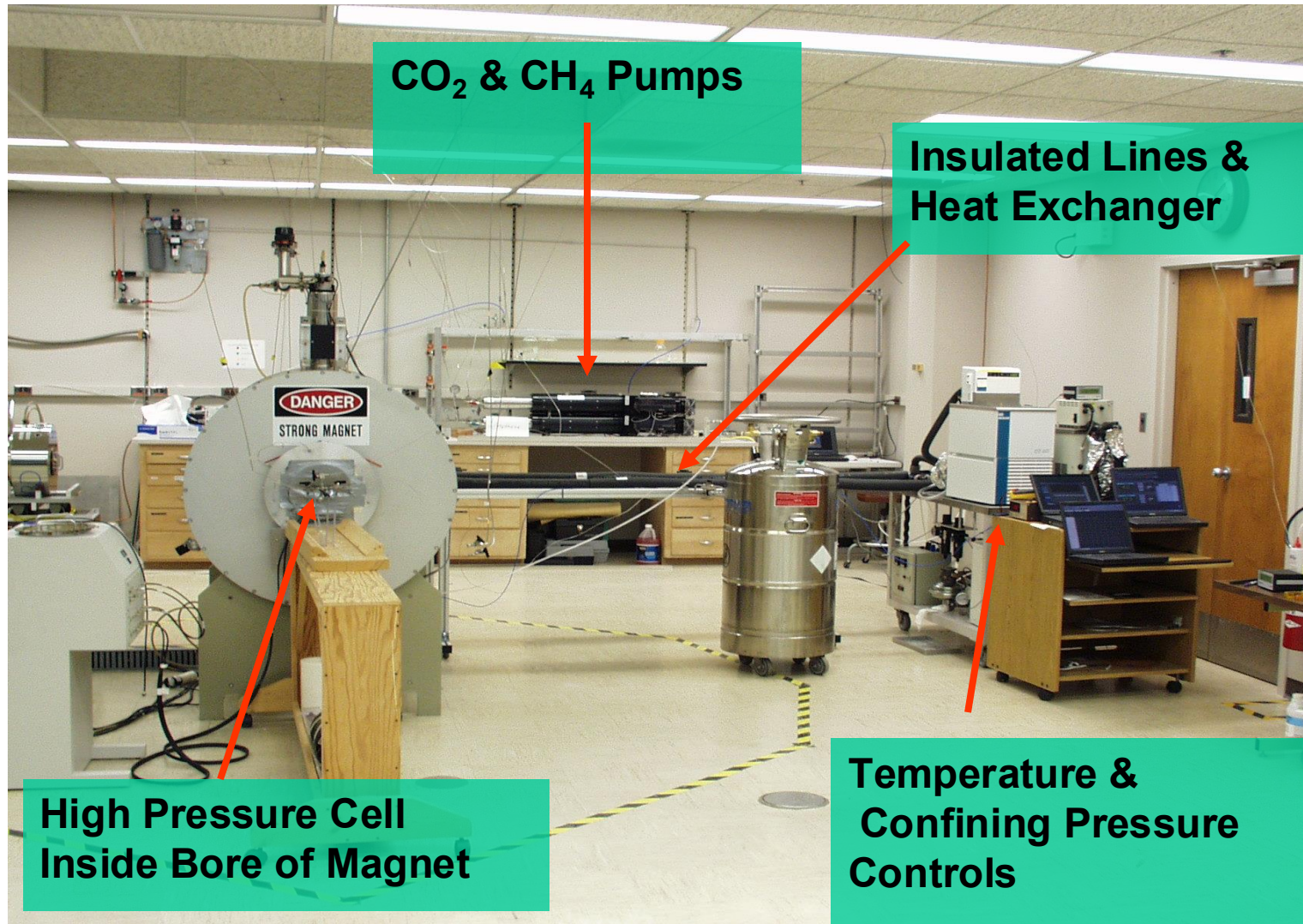
Bentheim Sandstone

20-25% porosity, ~1.1 D Perm

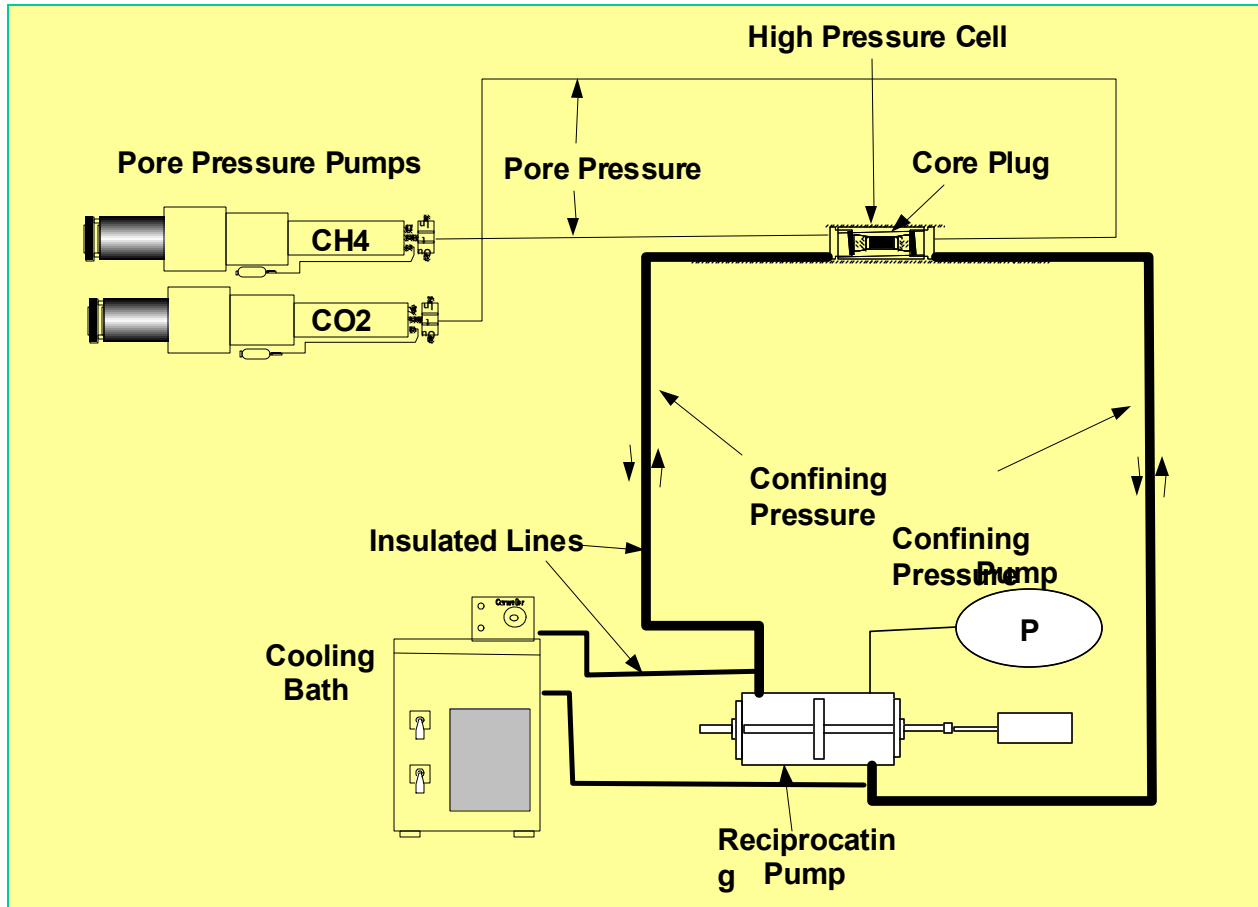
- Whole Core
- Longitudinal Cut With Machined Spacer to Simulate Open Fracture.



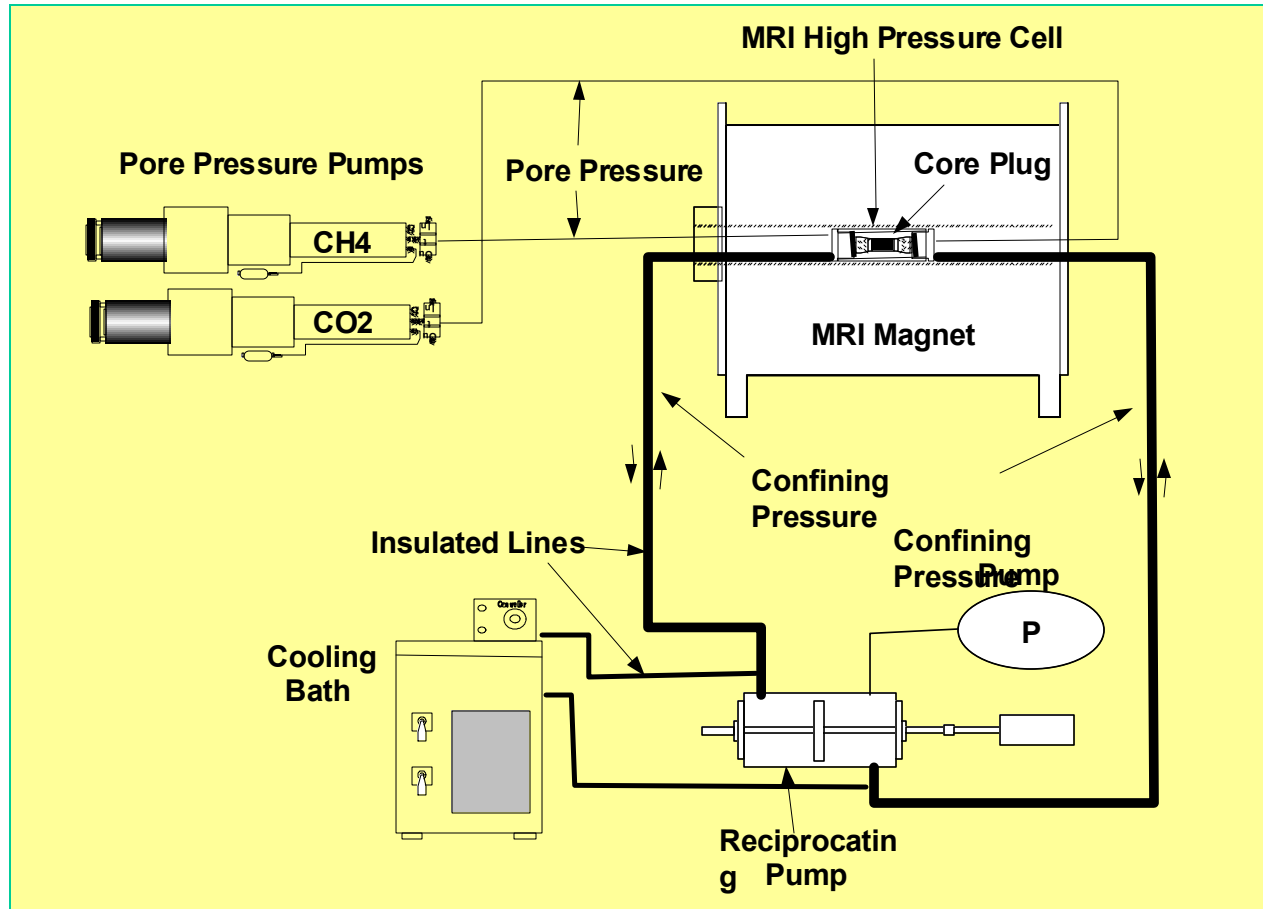
Experimental Setup



Experimental Setup



Experimental Setup



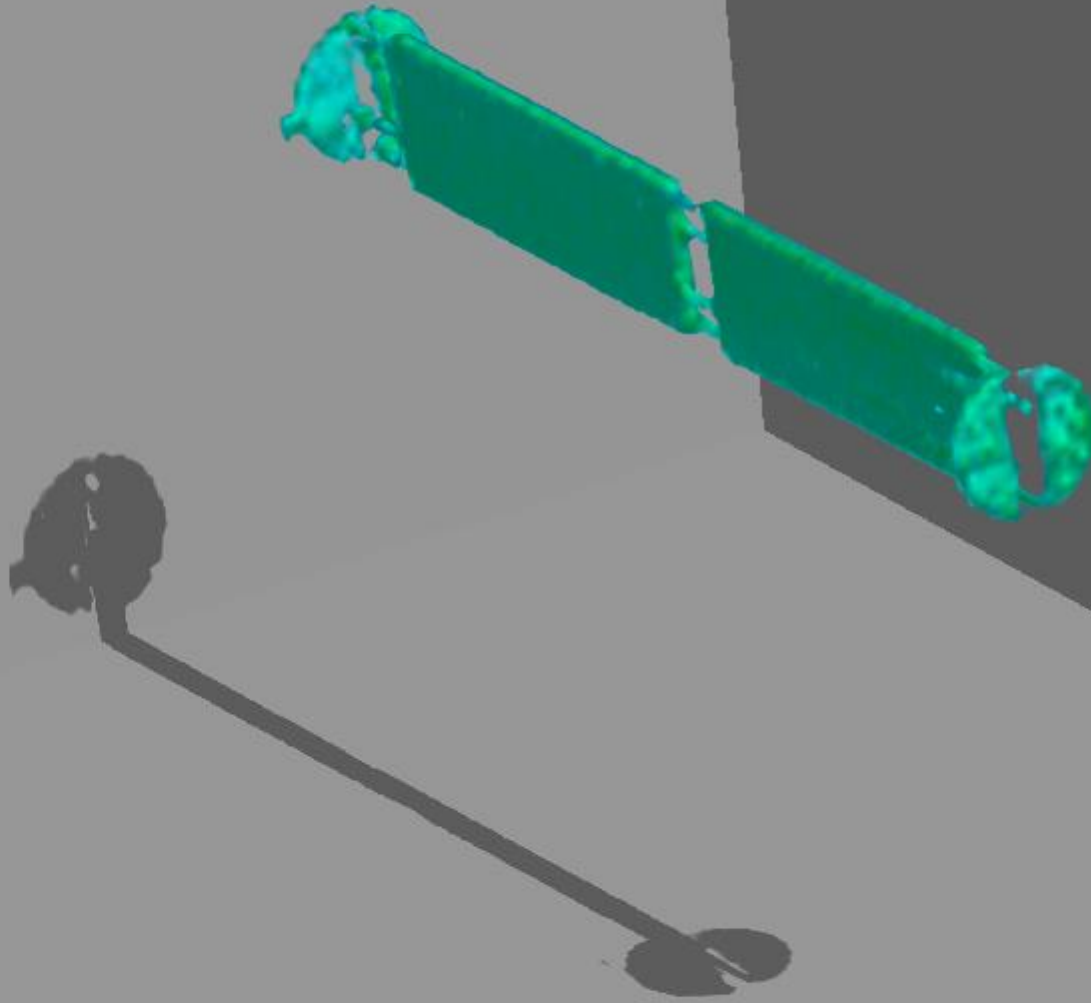
**Monitor P-V-T and MRI
Intensity**

**During Hydrate
Formation**

33-03

0.0 hrs

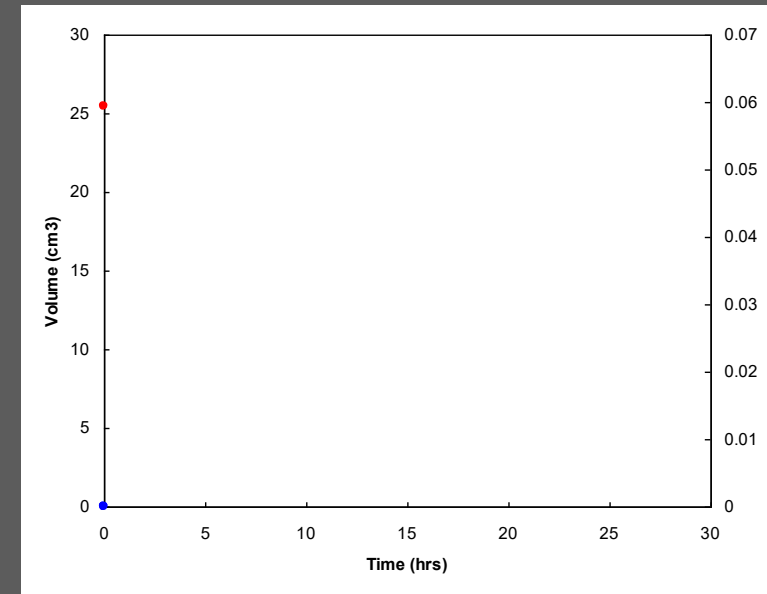
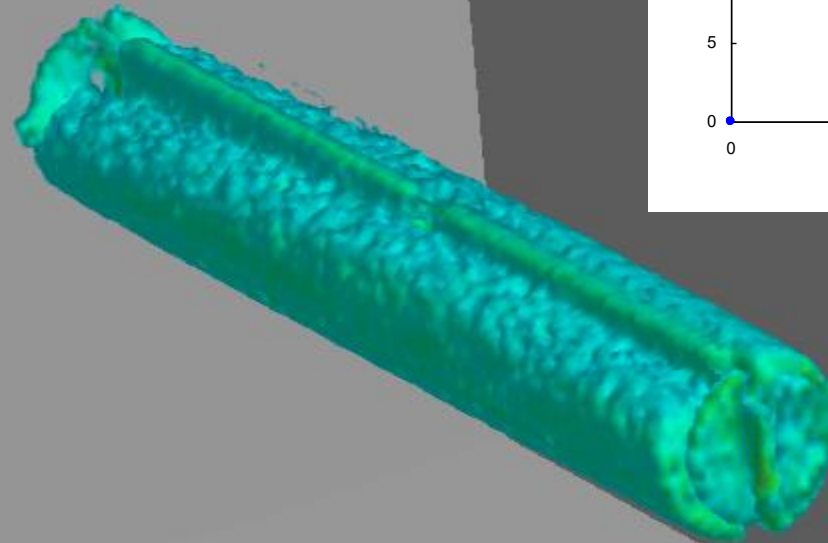
Methane in Spacer



33-07

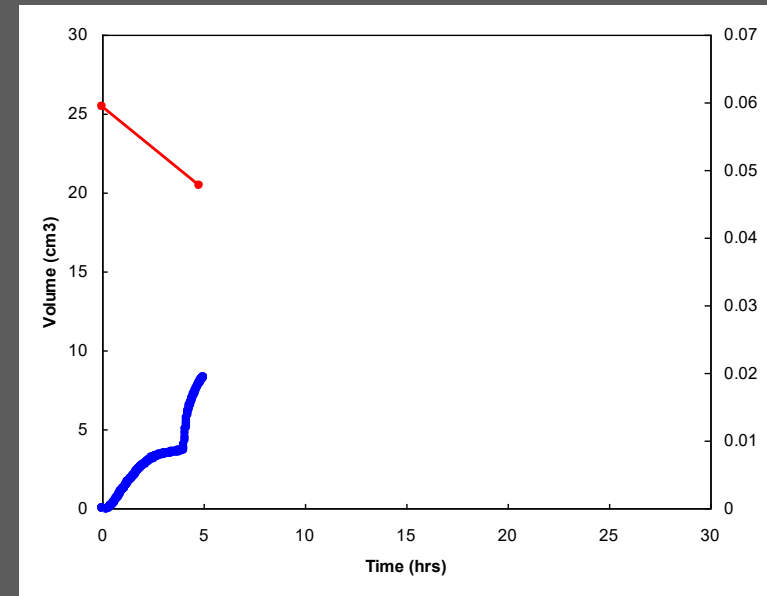
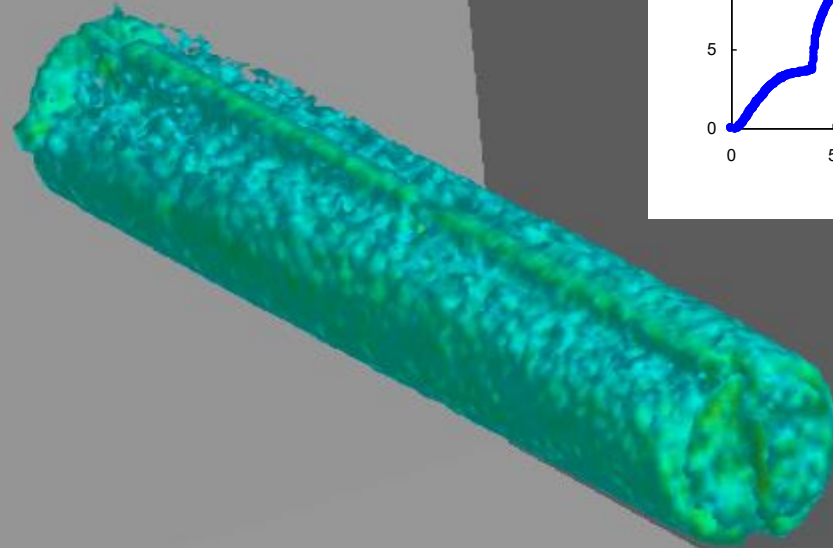
0.0 hrs

Sw=0.5 + Methane



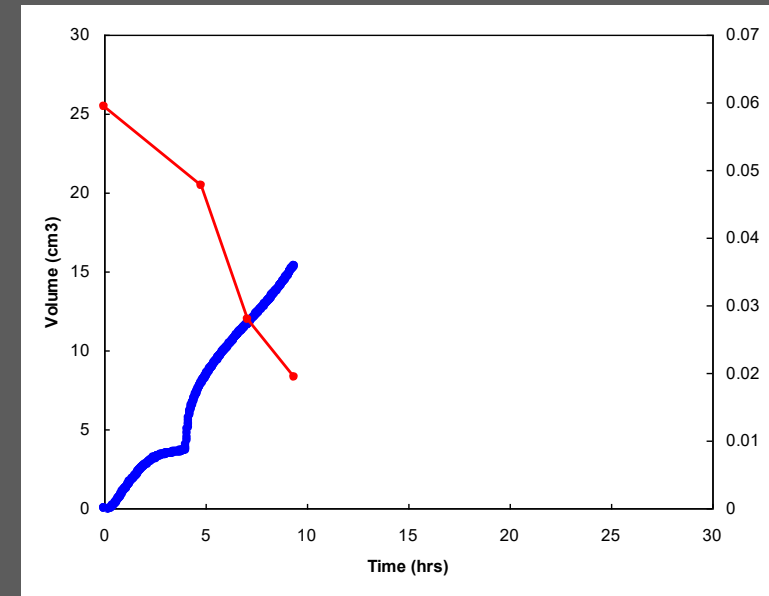
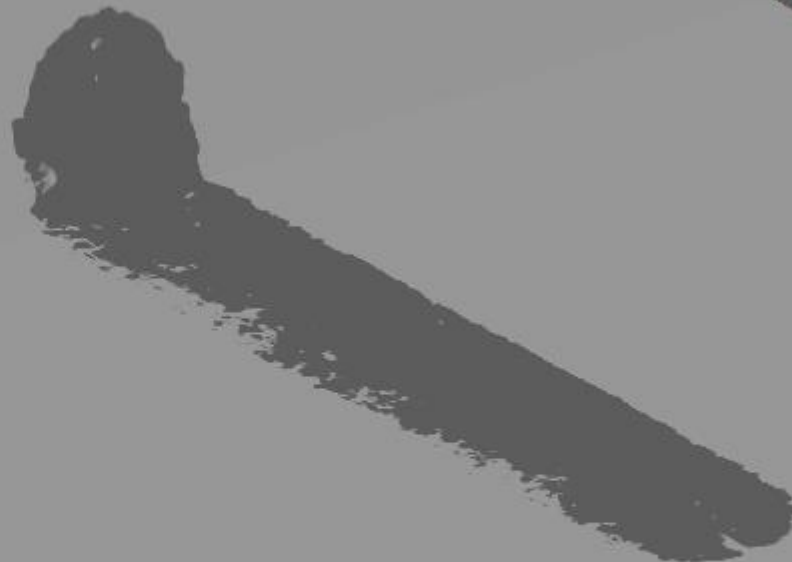
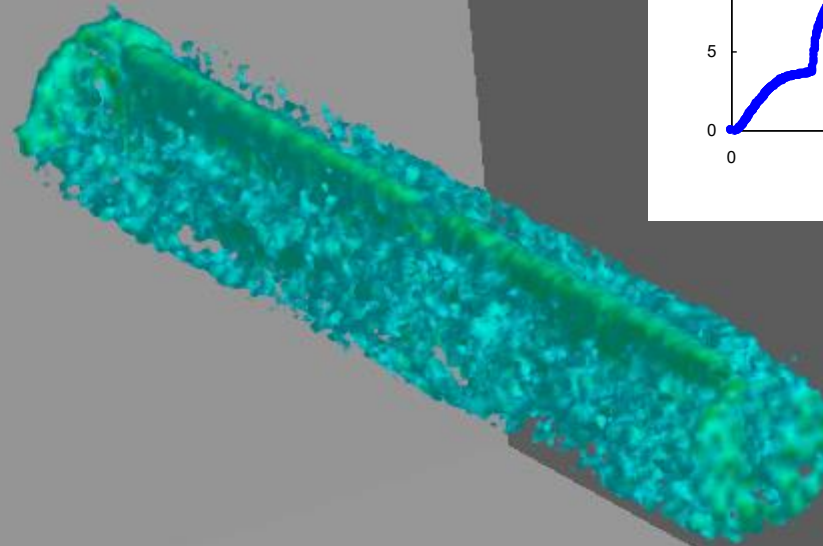
33a-01

5.0 hrs
Cooling Starts



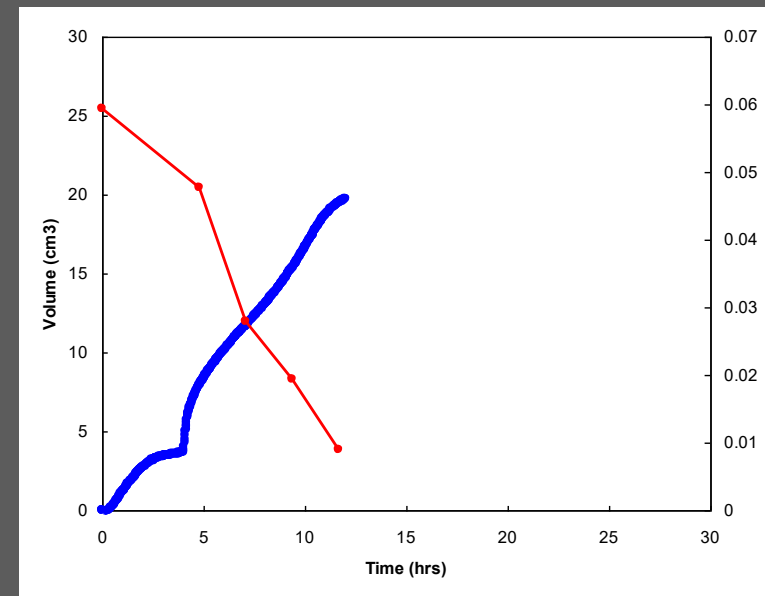
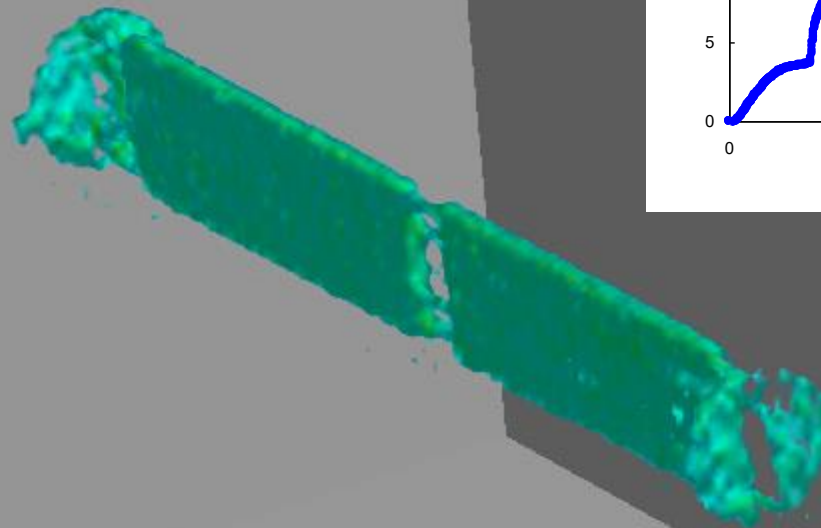
33c-02

9.4 hrs



33c-03

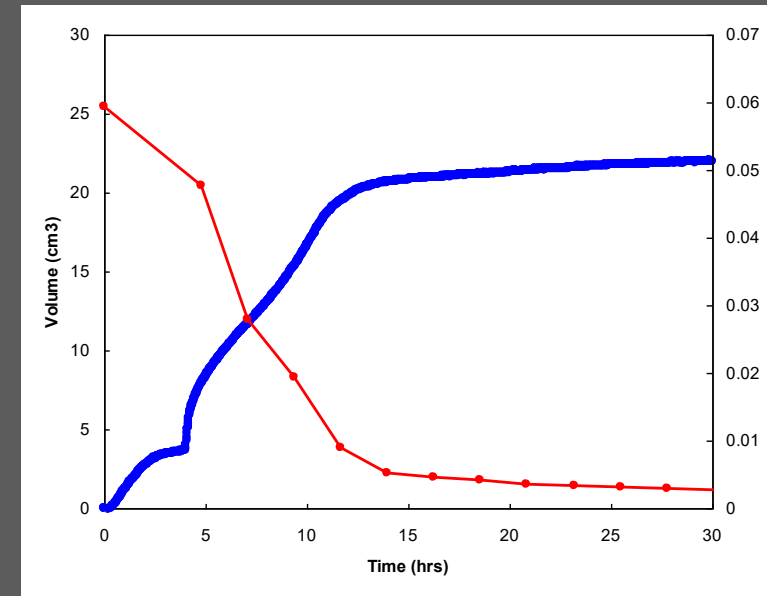
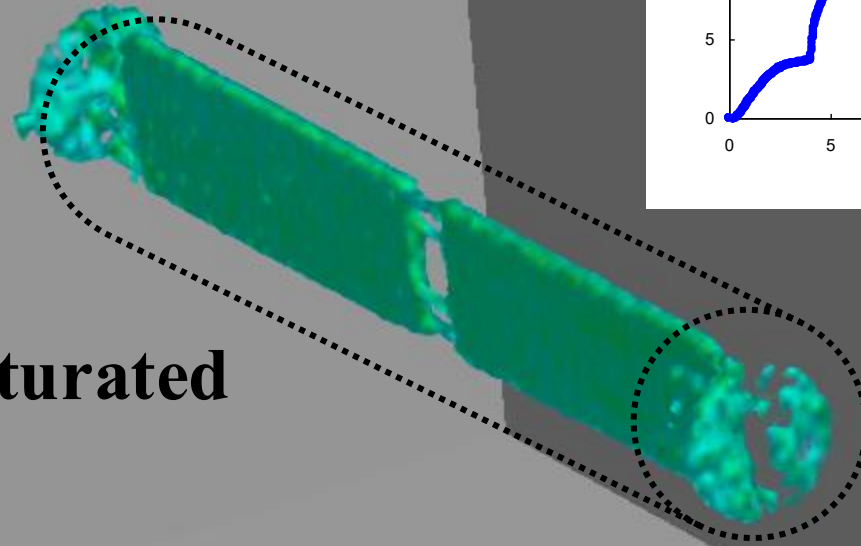
12.0 hrs



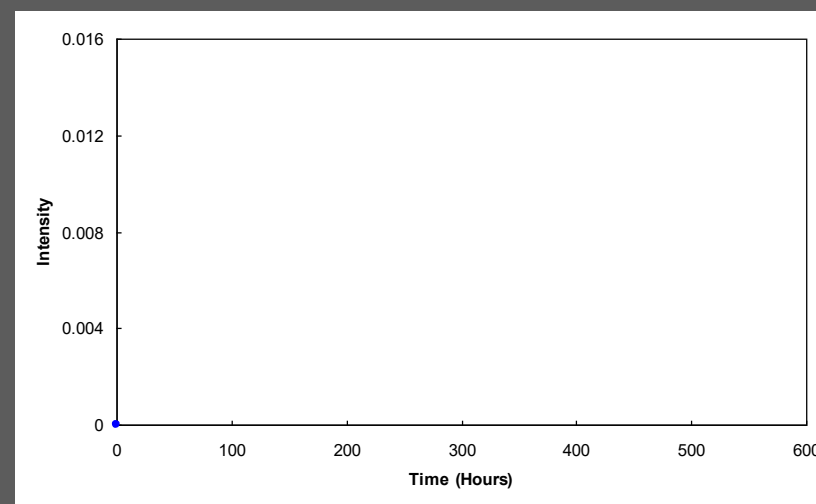
33c-11

30.0 hrs

**Core Halves Saturated
with hydrate**

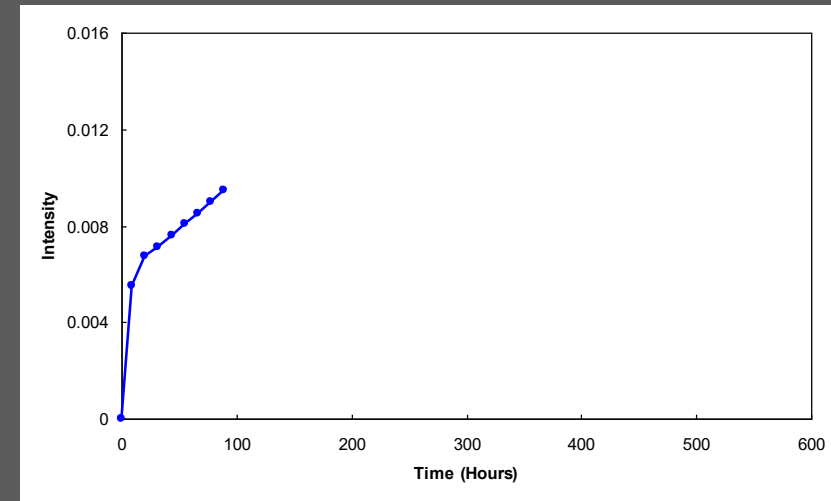


0.0 hrs



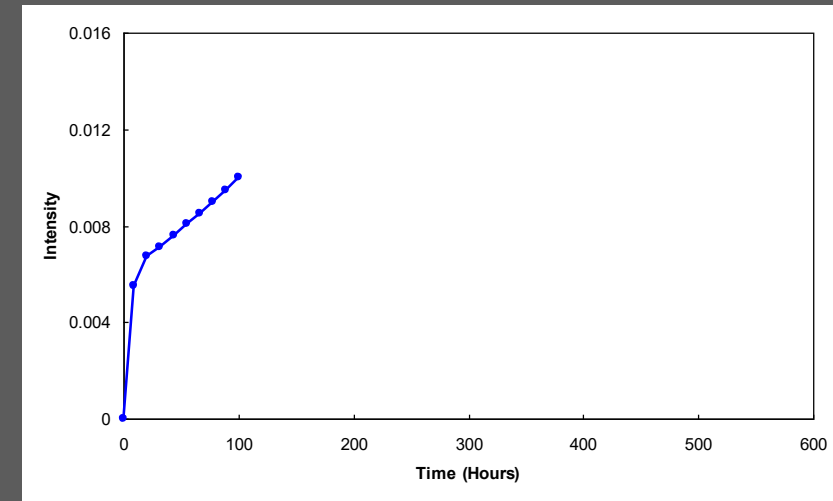
34b-07

89.2 hrs



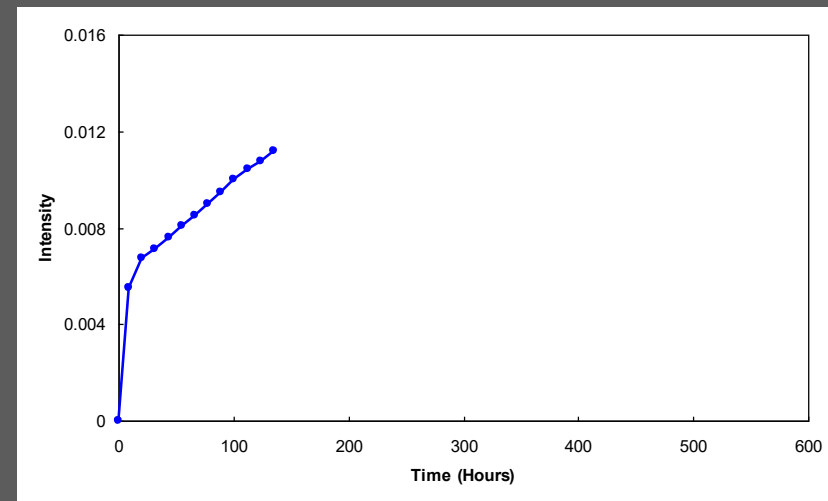
34b-08

100.6 hrs



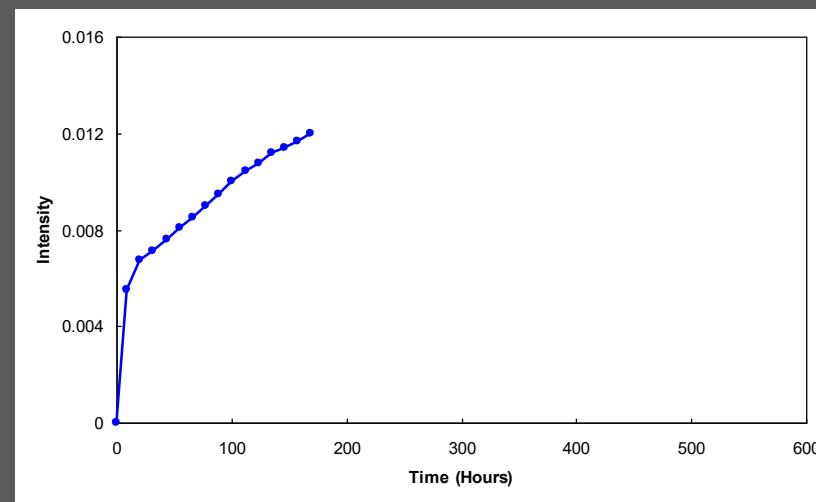
34b-11

135.0 hrs



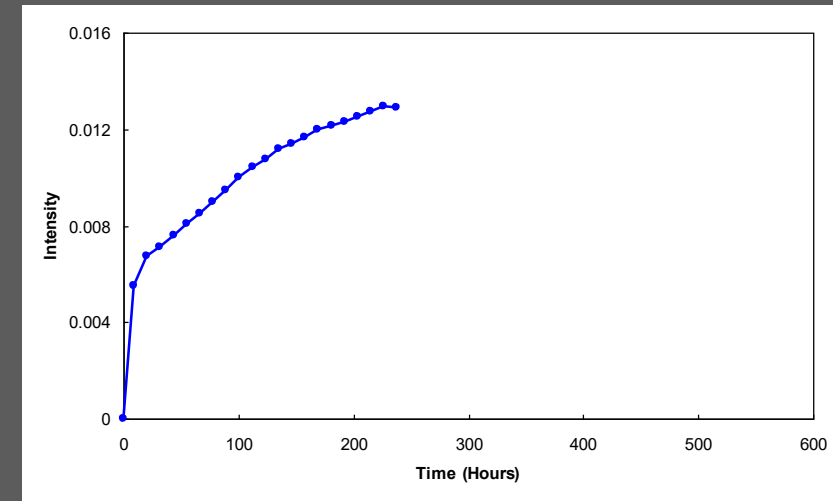
34b-14

169.3 hrs



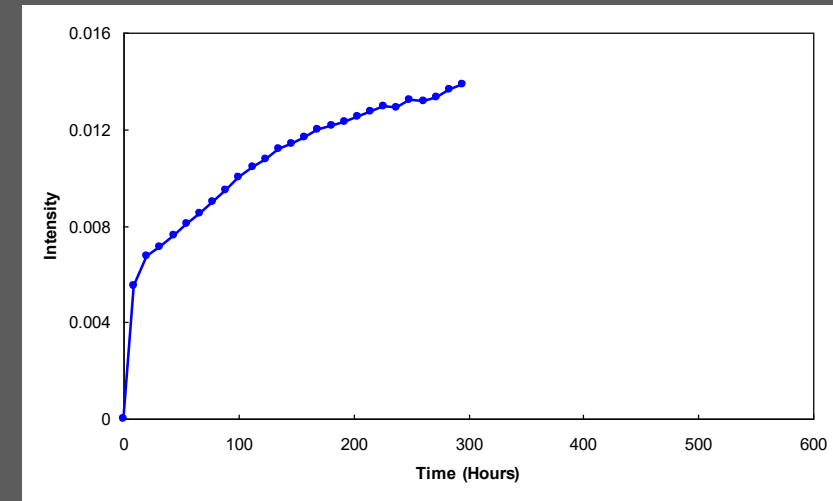
34b-20

237.9 hrs



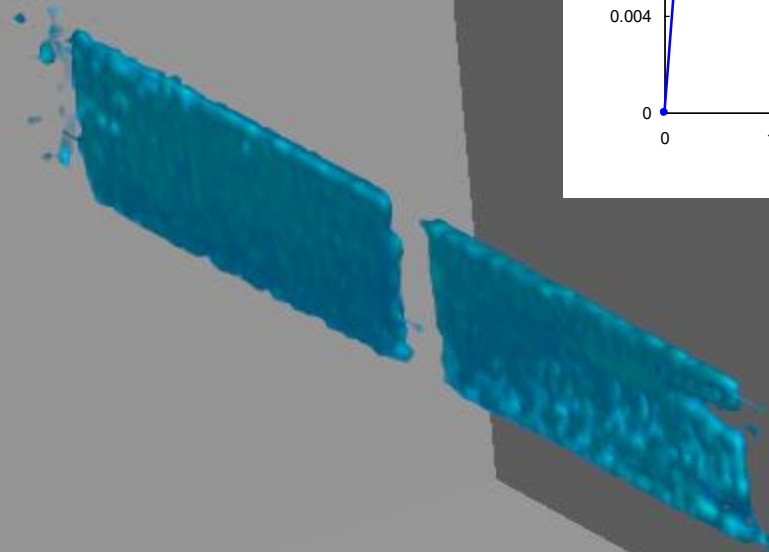
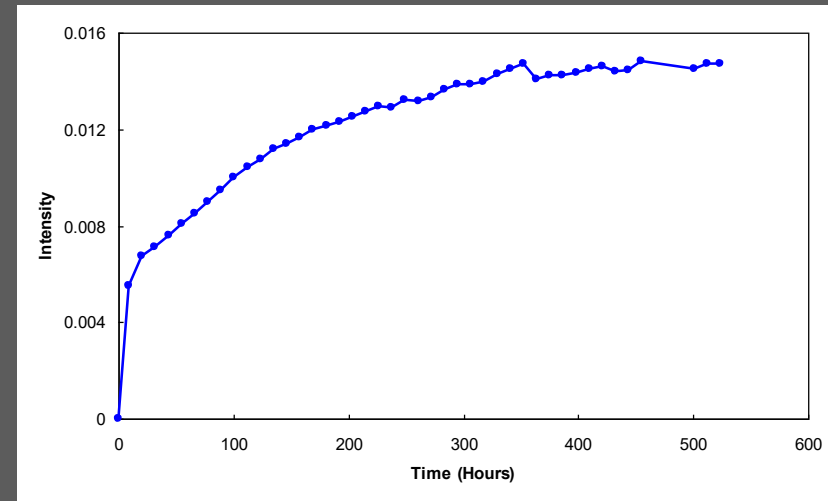
34b-25

295.1 hrs



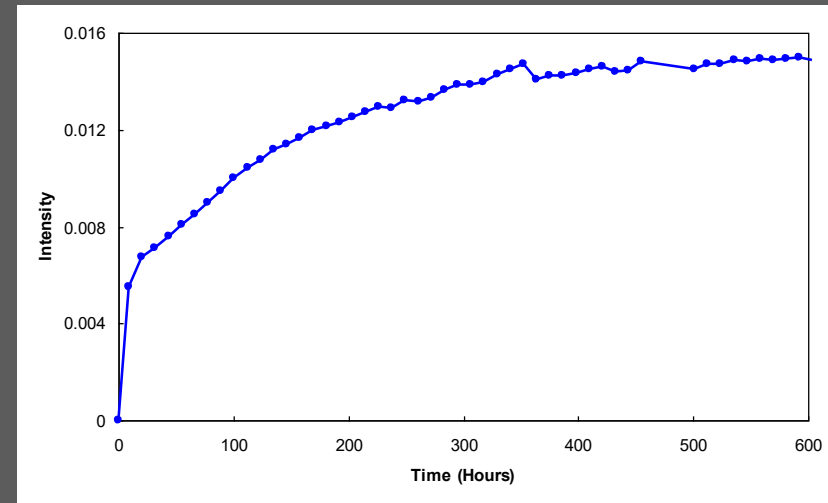
34b-45

523.8 hrs

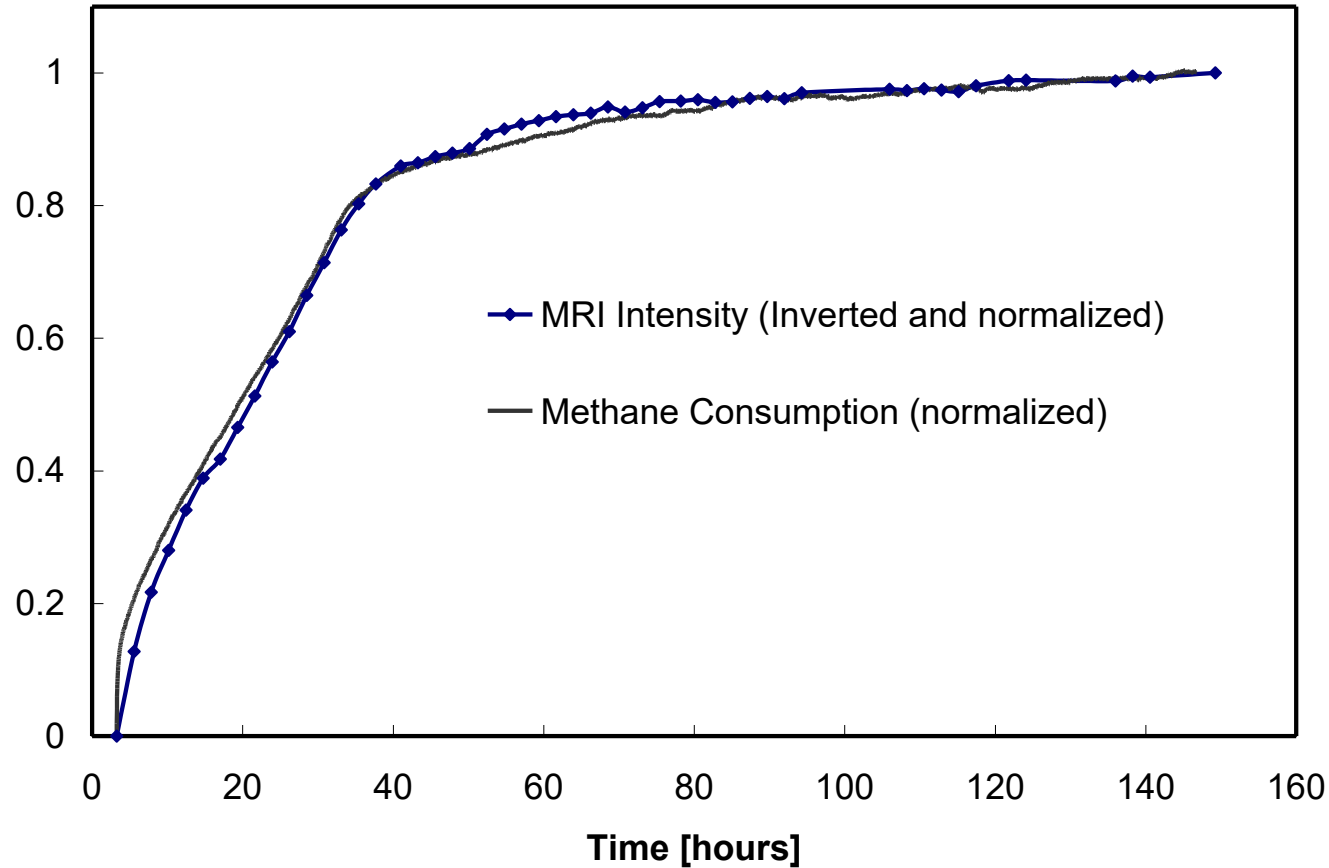


34b-52

603.9 hrs



Volumetrics and MRI Results



MRI Intensity in Core and CH₄ Volume Consumption

Scientific Conclusions

- **MRI Provides Unique Dynamic Data of Hydrate Formation and Production Consistent with Conventional Results.**
- **CO₂ Exchange for CH₄ in Hydrates Is Rapid and Efficient.**
- **No Free Water Observed During Exchange Process.**
- **Sufficient Permeability Remains During Hydrate Formation and Subsequent Production.**