



the center for negative carbon emissions

**ASU** IRA A. FULTON SCHOOLS OF  
**engineering**

CCUS Conference Week  
Golden, Colorado  
2025

**CO<sub>2</sub> Capture from Air**

Disclosure: Lackner advises several companies in the air capture field

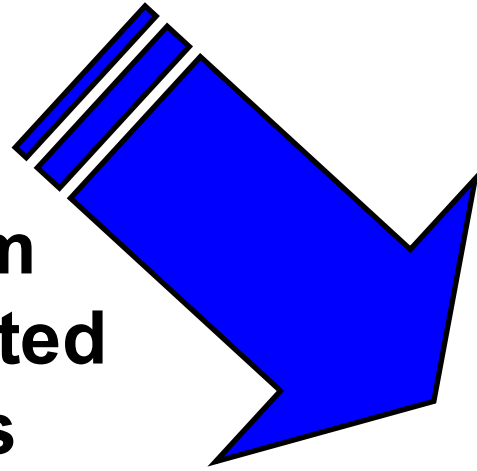
**ASU** Julie Ann Wrigley  
**Global Futures Laboratory**  
Arizona State University

**Klaus Lackner**

Founding Director, Professor at the School of  
Sustainable Engineering and the Built Environment  
September 2025

# Net Zero Carbon Economy

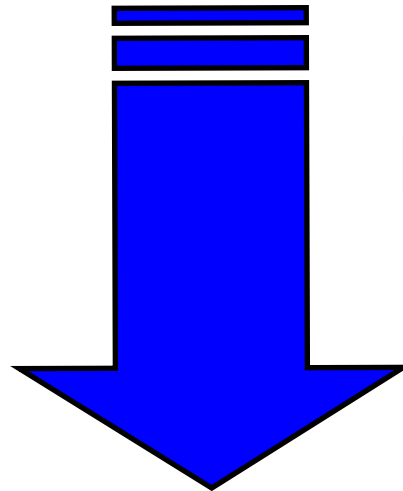
**CO<sub>2</sub> from  
concentrated  
sources**



**CO<sub>2</sub>  
extraction  
from air**



**Permanent &  
safe  
disposal**



# Three Rules for Technological Fixes

D. Sarewitz and Richard Nelson (Nature, 2008, 456, 871-872)

- I. The technology must largely embody the cause-effect relationship connecting problem to solution.**
- II. The effects of the technological fix must be assessable using relatively unambiguous or uncontroversial criteria.**
- III. Research and development is most likely to contribute decisively to solving a social problem when it focuses on improving a standardized technical core that already exists.**

“In contrast, direct removal of CO<sub>2</sub> from the atmosphere — air capture — satisfies the rules for technological fixes. Most importantly, air capture embodies the essential cause–effect relations — the basic go — of the climate change problem, by acting directly to reduce CO<sub>2</sub> concentrations, independent of the complexities of the global energy system (Rule I). There is a criterion of effectiveness that can be directly and unambiguously assessed: the amount of CO<sub>2</sub> removed (Rule II). And although air-capture technologies have been remarkably neglected in both R&D and policy discussions, they nevertheless seem technically feasible (Rule III).”

# **CDR = Carbon Dioxide Removal**

## **= Extraction from the mobile carbon pool**

- Biosphere extraction
  - Carbon removal via biomass will affect atmosphere in short order
  - Low-cost option but limited in scope
  - Siting limitations and scale limitations
- Ocean extraction
  - Thermodynamically equivalent to air capture
  - Immediate impact on the atmosphere
  - Carbon dilution in the ocean 1:26,000
- Air extraction (Direct Air Capture or DAC)
  - Fastest mixing times
  - Dilution 1:2,500
  - No siting limitations – no scale limitation
  - Cost must come down tenfold!!

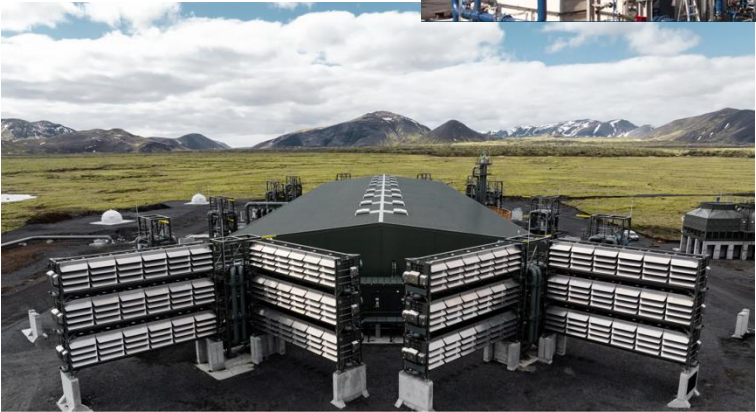
# DAC Systems



1 pointfive website



Carbon Engineering Website



Climeworks Website

# Direct Air Capture

- Chemical process technology for extracting CO<sub>2</sub> from ambient air
  - Distinct from photosynthetic (biological) systems
  - It collects carbon dioxide for use, disposal or both
  - CO<sub>2</sub> is available everywhere
- DAC separates sources from sinks
  - Allows for capture from diffuse sources
    - Cars, trucks, ships, airplanes
    - Boilers etc.
    - Past emissions



# Direct Air Capture: Closing the Carbon Cycle on the Teratonne-Scale

- Air capture can produce feedstock for fuels and chemicals (DACCU)
  - Current rate of oil consumption generates 1.5 Teraton CO<sub>2</sub> in the 21<sup>st</sup> century
  - DAC can promote solar energy to become the dominant primary energy source
- Air capture can collect waste from past and future emissions (DACCS)
  - Collecting 100 ppm from the atmosphere requires 1.5 Teraton of CO<sub>2</sub> capture
  - Sequestration cannot be avoided anymore
- What else can reach this scale? (Trillion-dollar annual revenue industry)
  - Without competing with food production
  - Without large environmental footprints

**DAC is exceptionally well positioned**

# Technical Feasibility of DAC

- Plenty analogs that demonstrate technical feasibility
  - Carbon dioxide removal on submarines and space craft
  - Stripping CO<sub>2</sub> and H<sub>2</sub>O from air for cryogenic air capture
  - Flue gas scrubbing for scale, but much higher concentrations
- DAC requires very different optimization
  - Extract CO<sub>2</sub> from air rather than clean it out
  - 100 to 300 times higher dilution than in flue gas capture



# What is taking so long?

- Too different from established technologies
  - Heavier-than-air flight and direct air capture are (nearly) impossible with off-the-shelf technology



Wikipedia images



- Lack of market pull

# Feasibility & Affordability?

CO<sub>2</sub> in air is dilute, and air is full of water



- Sherwood's relates costs scale linearly to dilution
- The air carries 10 to 100 times as much H<sub>2</sub>O as CO<sub>2</sub>
- First-of-a-kind apparatus is expensive (~\$1000/tonne)

**Not a conventional separation technology**

# First: CO<sub>2</sub> in the air is not too dilute!

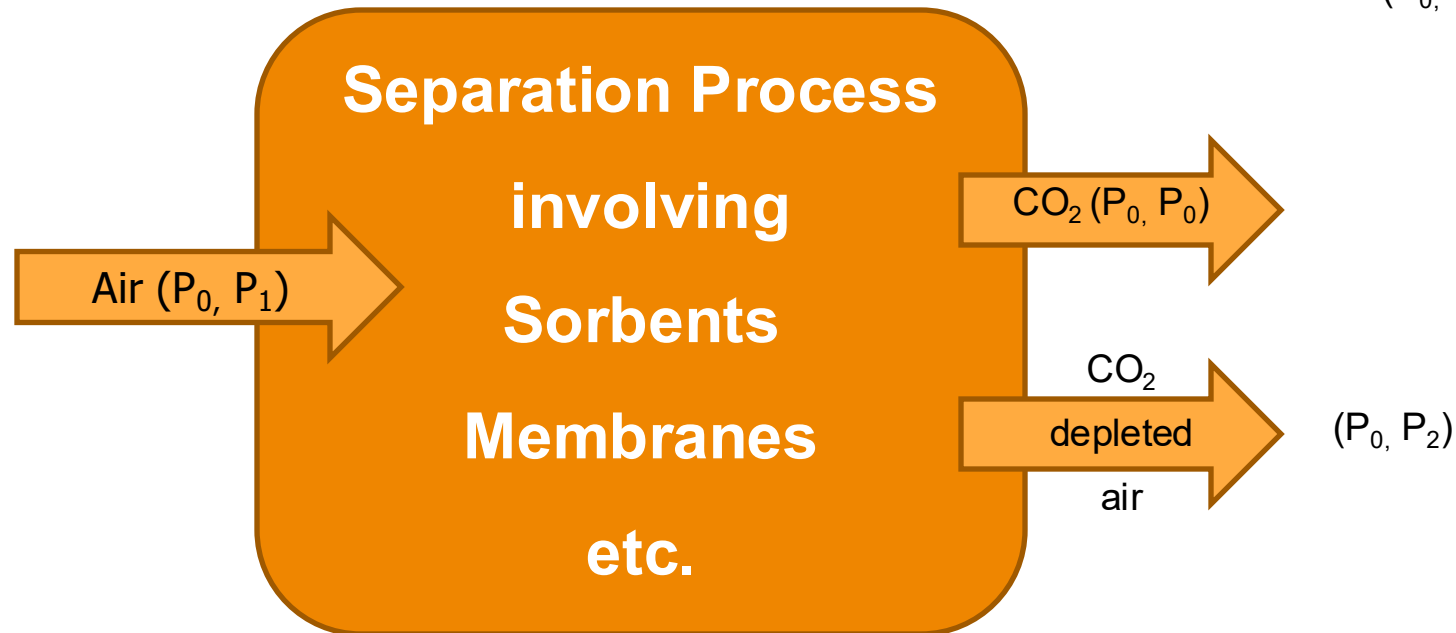
- One cubic kilometer of air
  - Passes through a windmill in the course of an afternoon
  - Carries \$300 of kinetic energy
    - assuming a wind speed of 6m/s and a value of 5¢/kWh
  - Carries \$21,000 of CO<sub>2</sub>
    - assuming a tipping fee or commodity value of \$30/ton

**As a source of CO<sub>2</sub>, the air is 70 times more valuable than as a source for wind energy.  
Wind energy is routinely harvested**

# Second: Thermodynamics is not limiting

Theoretical minimum free energy requirement for the regeneration is the free energy of mixing

Total gas pressure  $P_0$   
CO<sub>2</sub> partial pressure  $P_x$   
Denoted as  $(P_0, P_x)$



$$\Delta G = RT \left( \left( \frac{P_0 - P_2}{P_1 - P_2} \right) \frac{P_1}{P_0} \ln \frac{P_1}{P_0} - \left( \frac{P_0 - P_1}{P_1 - P_2} \right) \frac{P_2}{P_0} \ln \frac{P_2}{P_0} + \left( \frac{P_0 - P_1}{P_0} \right) \left( \frac{P_0 - P_2}{P_0} \right) \frac{P_0}{P_1 - P_2} \ln \frac{P_0 - P_1}{P_0 - P_2} \right)$$

**22 kJ/mol CO<sub>2</sub> vs. heat of combustion 400 – 900 kJ/mol CO<sub>2</sub>**  
*(Specific irreversible processes will always require more energy)*

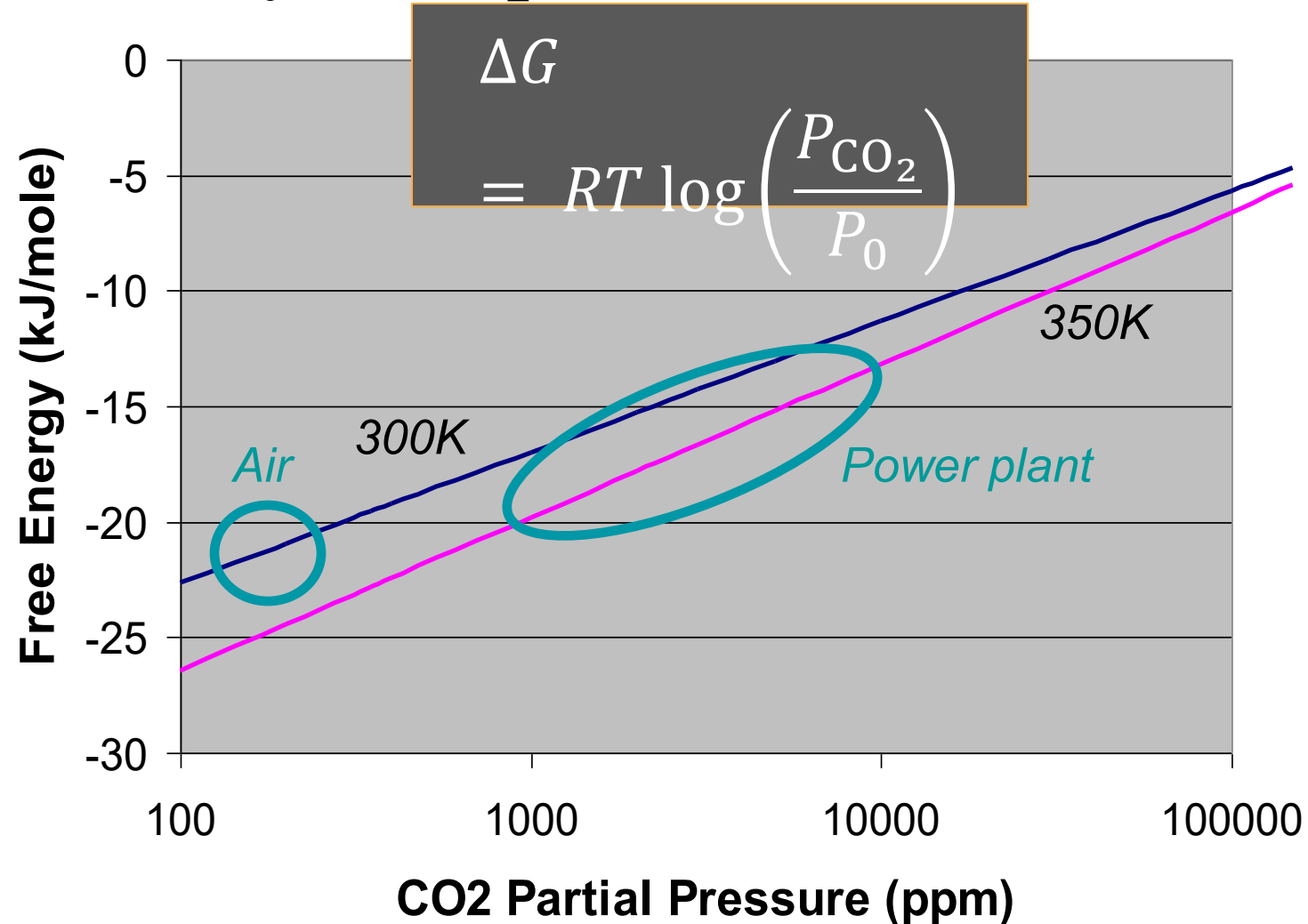
# Air capture is sorbent based\*

- Sorbents bind  $\text{CO}_2$  without need for spending energy on the air
  - Concentration ratio is 1 : 2500
    - Sorbents postpone work to the regeneration step, only do work on  $\text{CO}_2$
- All air capture sorbents are chemical sorbents
  - At 400 ppm only chemical bonds are strong enough,  $|\Delta G| > 22 \text{ kJ/mol}$
- Today's air capture sorbents exploit carbonate chemistry
  - Alkali hydroxides
  - Weak and strong based amines
  - Thermal, vacuum and reaction-based recovery
  - Humidity swing takes advantage of  $\text{H}_2\text{O} - \text{CO}_2$  – sorbent reactions
- Solid sorbents deliver better kinetics

\*The exception are active membranes. We are working on a moisture gradient to pump  $\text{CO}_2$  against a concentration gradient

# Required Sorbent Strength

depends logarithmically on CO<sub>2</sub> concentration at collector exit



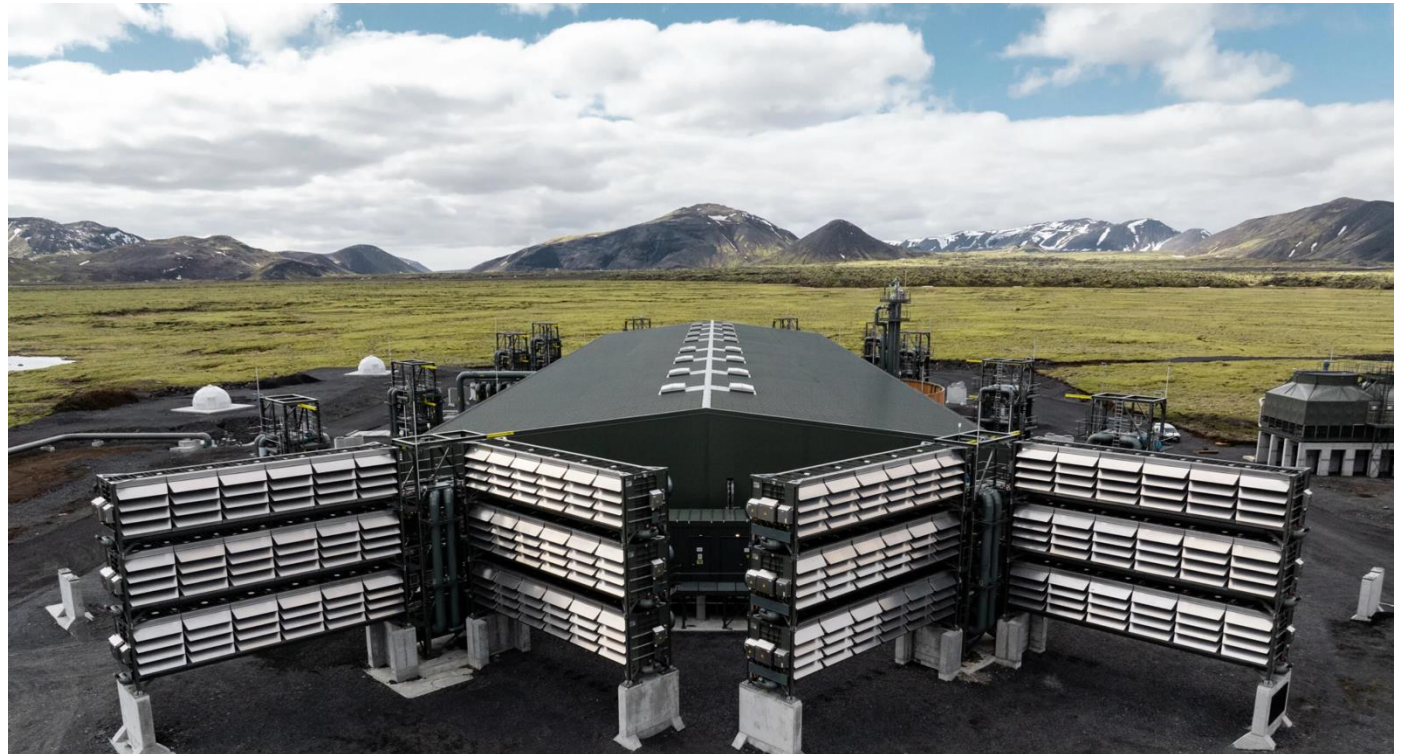
Disagrees with  
Sherwood's Rule



# Third: Direct Air Capture is too expensive

- Tax incentives (45Q, LCFS in the US and California) provide about \$200/tonne
- No DAC company is currently selling actual credits on this basis

\$250 to \$1,000/tonne





# Sherwood's Rule is avoidable

Cost of separation scales linearly with dilution  $D$

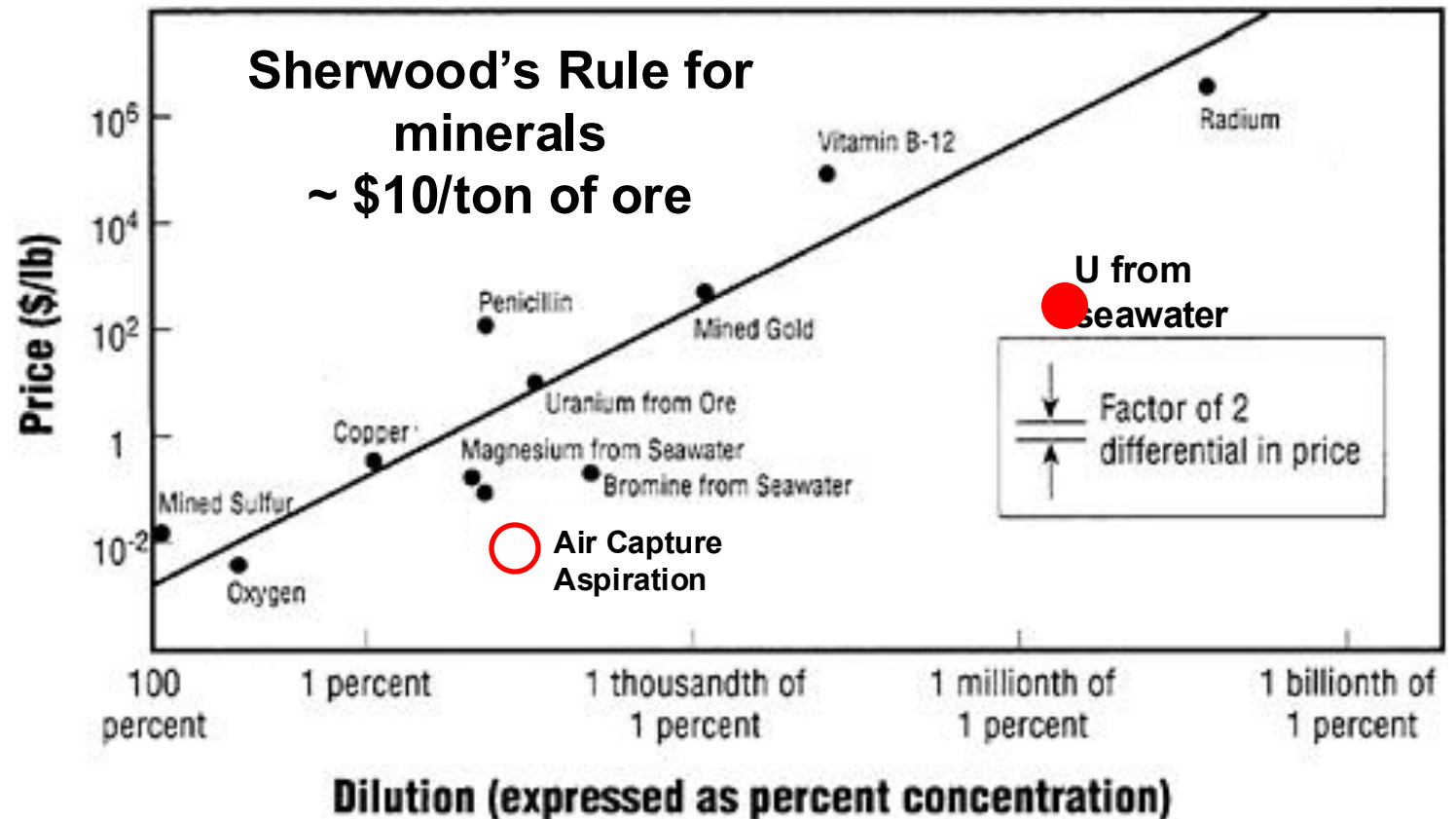
## Sherwood's Rule

- The cost of the first step in the separation dominates

- $Cost = aD + b + c \log D$

Bulk  
processing

Thermodynamic  
separation



SOURCE: National Research Council (1987)

# Artificial kelp to absorb uranium from seawater

- Passive, long-term exposure to water
  - Braids of sorbent covered buoyant plastic
  - Anchored to the floor
  - Replaced initially active systems
- Low energy sorbent
  - Laminar flow over sorbent
  - Uptake is limited by boundary layer transport
- Regeneration
  - After harvesting the strings
- Gross violation of Sherwood's Law
  - Cost estimates range from \$200 to \$1200/kg
  - Sherwood \$3 million/kg



Artificial kelp to absorb uranium from seawater

$$Cost = aD + b + c \log(D)$$

- Passive, long-term exposure to water
  - Braids of sorbent covered buoyant plastic
  - Anchored to the floor
  - Replaced initially active systems

**must make parameter  $a$  small**

- Low energy sorbent
  - Laminar flow over sorbent
  - Uptake is limited by boundary layer transport

**passive systems!**

- Regeneration
  - After harvesting the strings

- Gross violation of Sherwood's Law

- Cost estimates range from \$200/kg to \$2000/kg
- Sherwood \$3 million/kg

**Wind driven air capture circumvents  
Sherwood's Rule**



# Irreducible Cost

- Raw inputs and outputs
  - provide lower bounds on costs that exclude inefficiencies, friction, and dissipation
- Lower bound
  - The difference between initial costs and the frictionless limit can be large and may never go to zero.
- Irreducible cost per tonne
  - Thermodynamic requirement (separation and compression) 215 kJ → \$2.15 ... \$10.75
    - Sherwood's rule does not change it
  - Equipment cost \$10-20/kg
    - Sorbent captures 3 mol/kg, cycle time 1000 sec, sorbent 1/3 of total mass
    - 10%/yr discount rate
    - \$0.70 to \$1.50/tonne
- Land use cost is insignificant

**\$10 - \$20/tonne**  
**This is what learning could aim for**



# Air Capture can avoid Sherwood's Rule

**DAC need not  
crush or grind  
air**

**Dominant  
cost is  
sorbent  
regeneration**

**somewhat  
more  
energetic than  
flue gas  
sorbent  
recovery**



**Air collector reduces net CO<sub>2</sub> emissions  
much more than equally sized windmill**

**Extracting kinetic energy from air at 20  
J/m<sup>3</sup> is feasible**



**Wind energy  
~20 J/m<sup>3</sup>**

**CO<sub>2</sub> combustion  
equivalent in air  
10,000 J/m<sup>3</sup>**

**Contacting of air can  
be inexpensive**

**Regeneration cost are  
slightly larger than for  
flue gas scrubbing**

# Cost of DAC is important

- Policy and consumers ultimately care about price
  - Difference between price and cost can be huge
- It is future costs not today's costs that matter
  - Huge cost reductions are common
- Unfortunately, cost is a slippery concept
  - Costs today are very different from costs tomorrow
  - Supply chain costs are other producers' prices
  - Supply chain costs respond to demand

**How to maximize the odds**



# Cost under mass production – The learning curve

- Doubling cumulative output lowers the reducible part of the cost by a factor  $\varepsilon \sim 0.8$
- $L = 1 - \varepsilon$  is known as the learning rate

$$k(n) = c(n) + r$$

$$c(n) = c_1 \varepsilon^{\log_2 n} = c_1 n^{\log_2 \varepsilon}$$

$$\log_2 \varepsilon = \alpha - 1$$

$k(n)$  is the cost of the  $n^{\text{th}}$  unit ( $k(1) = \$500/\text{t}$ )

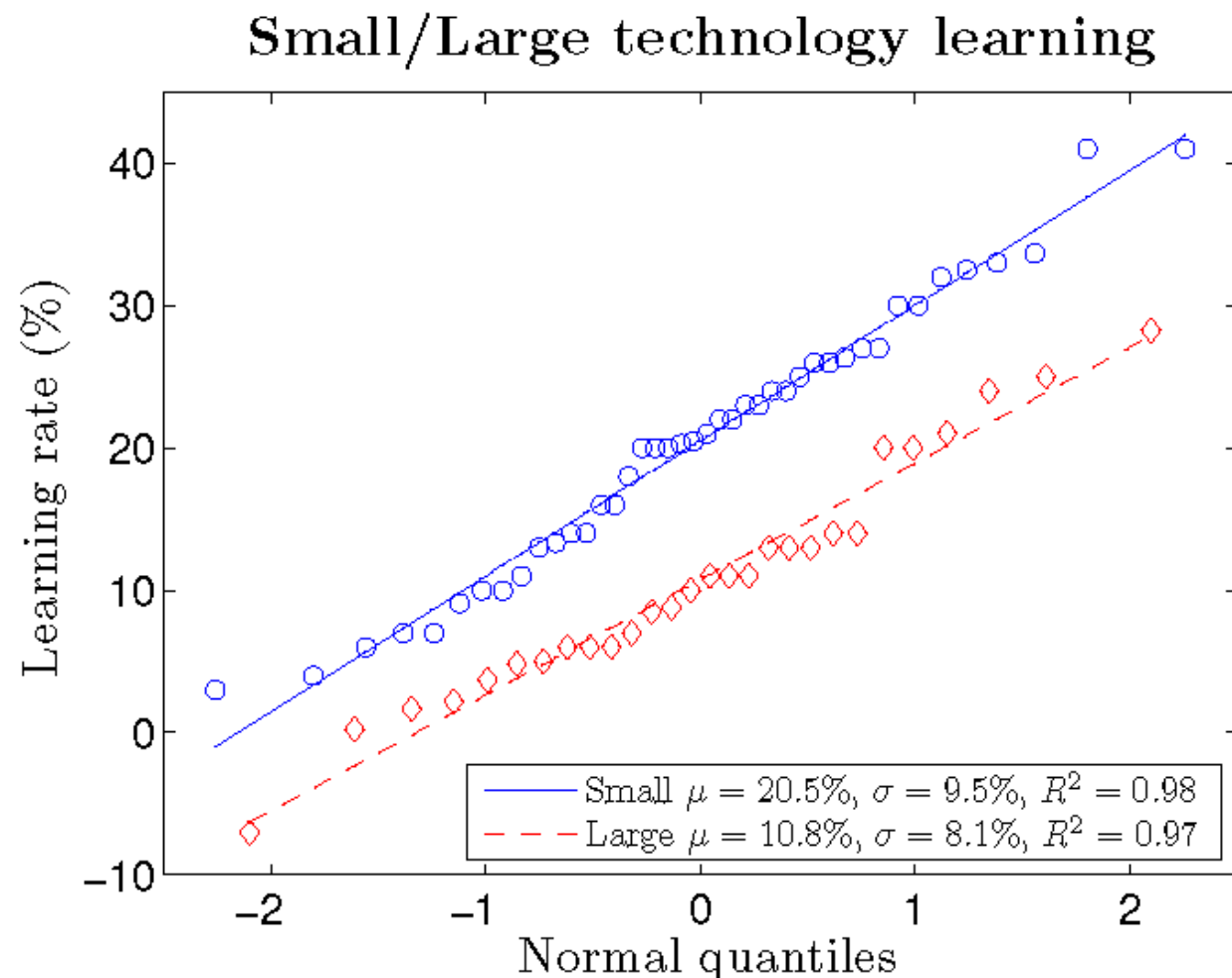
$r$  is the irreducible cost ( $r = \$30/\text{t}$ )

$c(n)$  follows a power law  $c(n) = c_1 n^{\alpha-1}$

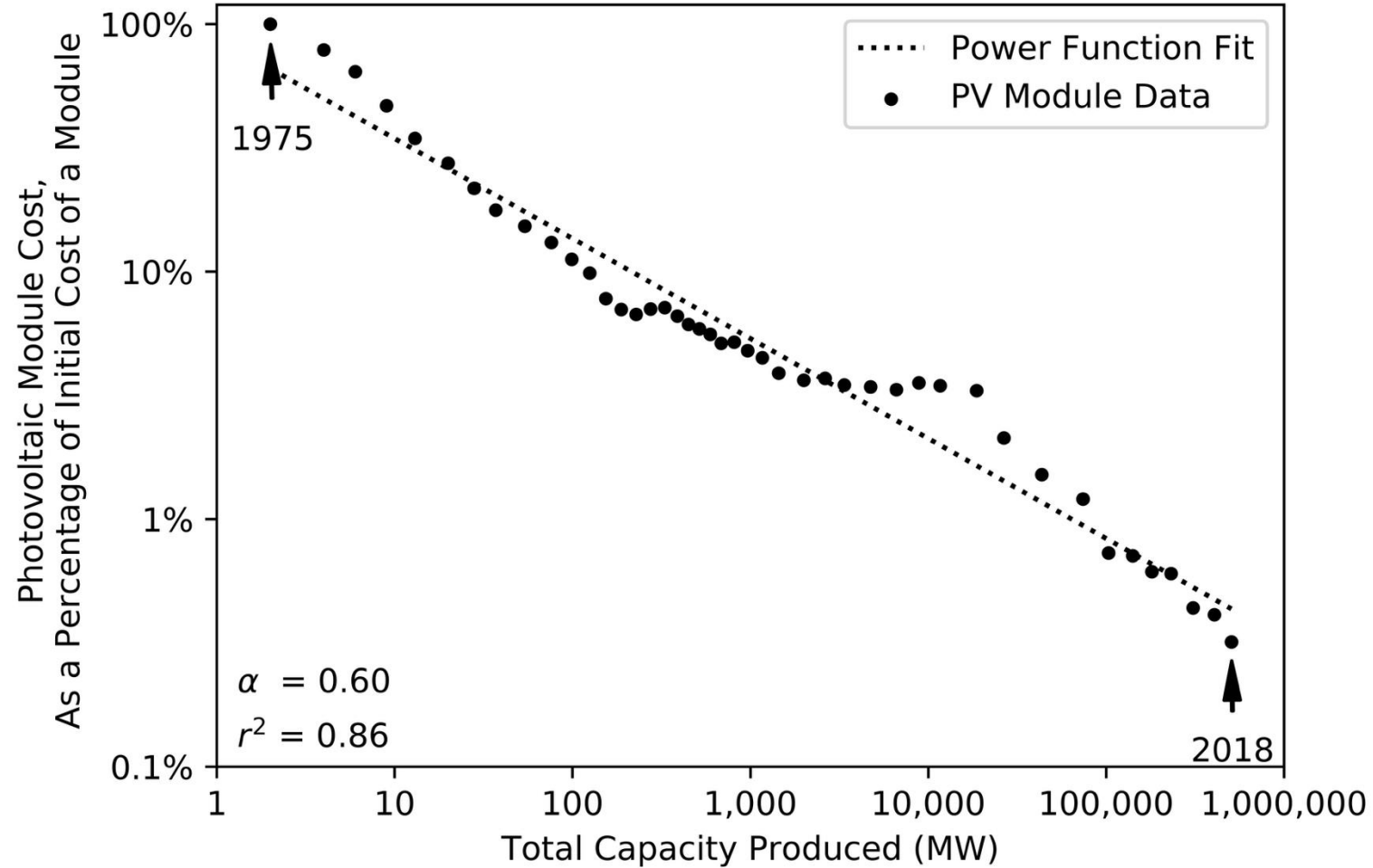
$\alpha$  is the power coefficient that relates total cost to cumulative size of production

# Learning works better for small units

- Range of learning coefficients ( $1 - \varepsilon$ ) changes with size



# Solar Energy Costs are Dropping Fast



Habib Azarabadi, ASU

# Industries forced into modularity often do well

- Automobile industry
- Smart phone industry
- Solar photovoltaic industry
- Wind industry (?)

**Cost reductions are faster in numbering up**

**Risks are reduced**

**Response to market changes is flexible**

**Multiple approaches can be tested**

# Demand Curves

- Market avoids stalling if

$$k(n) < P(n) \quad \text{for all } n$$

With  $n_d(P)$ , the demand curve and  $n_c(P)$  the inverse cost curve,  
The equivalent statement is

$$n_d(P) \geq n_c(P) \quad \text{for all } P$$

$$n_d(P) \geq \left( \frac{P - R}{c_1} \right)^{-\frac{1}{\alpha}}$$

# Attributes of a good DAC technology

- Low starting cost
  - Low capital and maintenance
  - Low energy inputs
- High learning rate
  - Rates vary across industries and technologies
- Low irreducible cost
  - For normal scaling costs should be acceptable at a 1000-fold increase
  - Climate action requires million-fold increase which means operation near the limit
  - Winners are impossible to predict

**Experiment with many different approaches**  
**Many small startups probe the technology space**  
**Enable scaling by a modular approach**

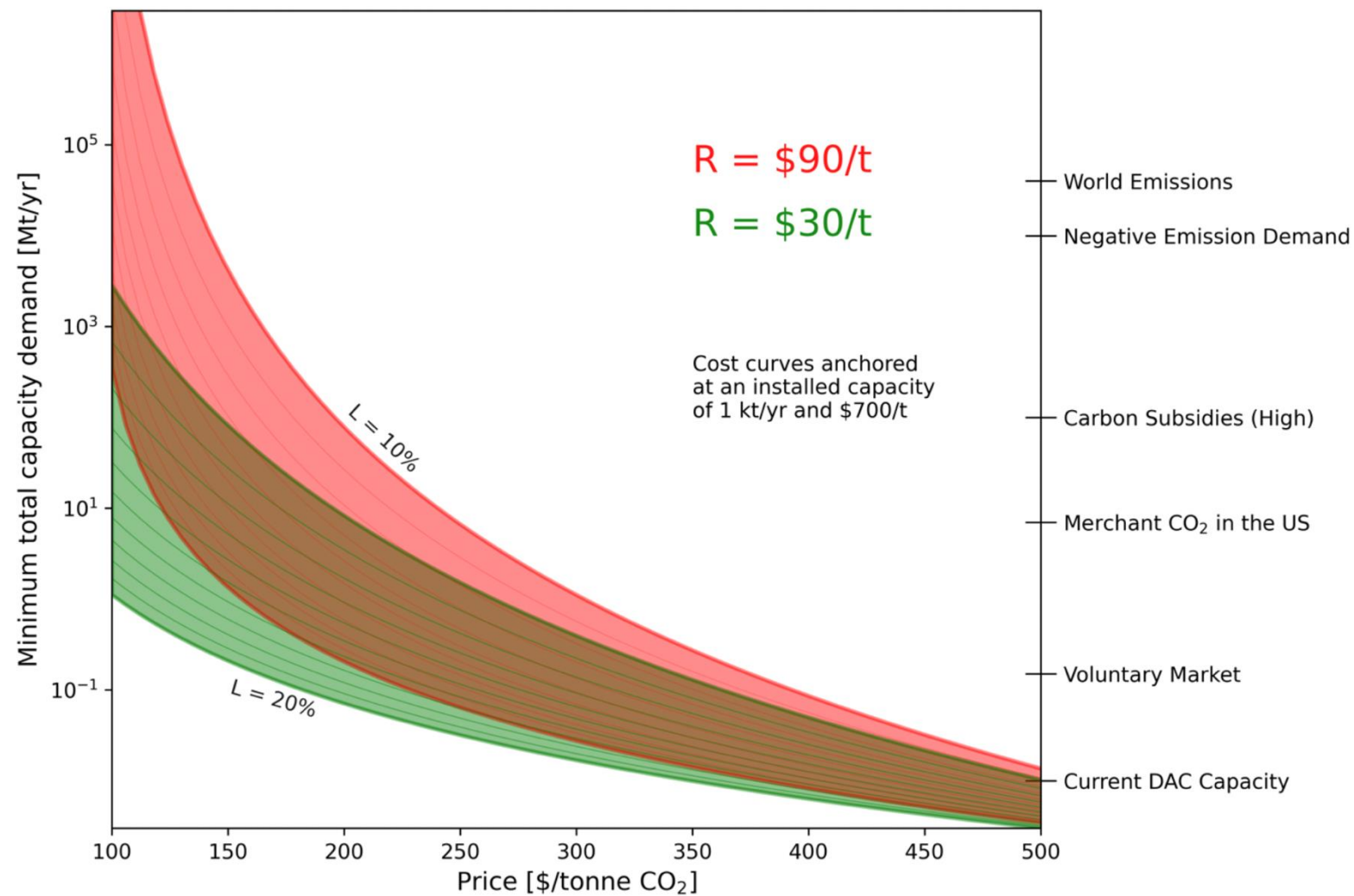
# What scale do we need?

- 40 Gt/yr three different ways
  - Roughly current rate of emissions
  - Emissions at the end of the century of carbon intensity is reduced fivefold
  - Lowering CO<sub>2</sub> in the atmosphere by 100 ppm over 40 years
- How much time do we have?
  - 10 years?
  - 50 years?
  - 100 years?

**At this scale even “natural” solutions turn technical**



# Learning rates matter



# Waste problems have been solved before



[https://commons.wikimedia.org/wiki/File:The\\_main\\_drainage\\_of\\_the\\_Metropolis\\_Wellcome\\_M0011720.jpg](https://commons.wikimedia.org/wiki/File:The_main_drainage_of_the_Metropolis_Wellcome_M0011720.jpg)

- Waste management is a lucrative service industry that need to be built
- CO<sub>2</sub> waste is global and can be addressed globally
- Carbon recycling will be driven by the cost of waste disposal

# DAC provides Carbon Capture of last resort

- Its advantage is it can be done anywhere and for any emissions
  - Likely operates where CO<sub>2</sub> is needed or energy is available
- Unless there is no need for DAC, it will set the price of carbon
  - Harsh commodity business, but so are many other necessities
- Net negative carbon economies require CDR technologies
  - Can balance out past emissions and emissions difficult to capture at the source
  - DAC offers the possibility of returning carbon to the developing countries

# Photovoltaic power will challenge fossil energy even without carbon constraints!

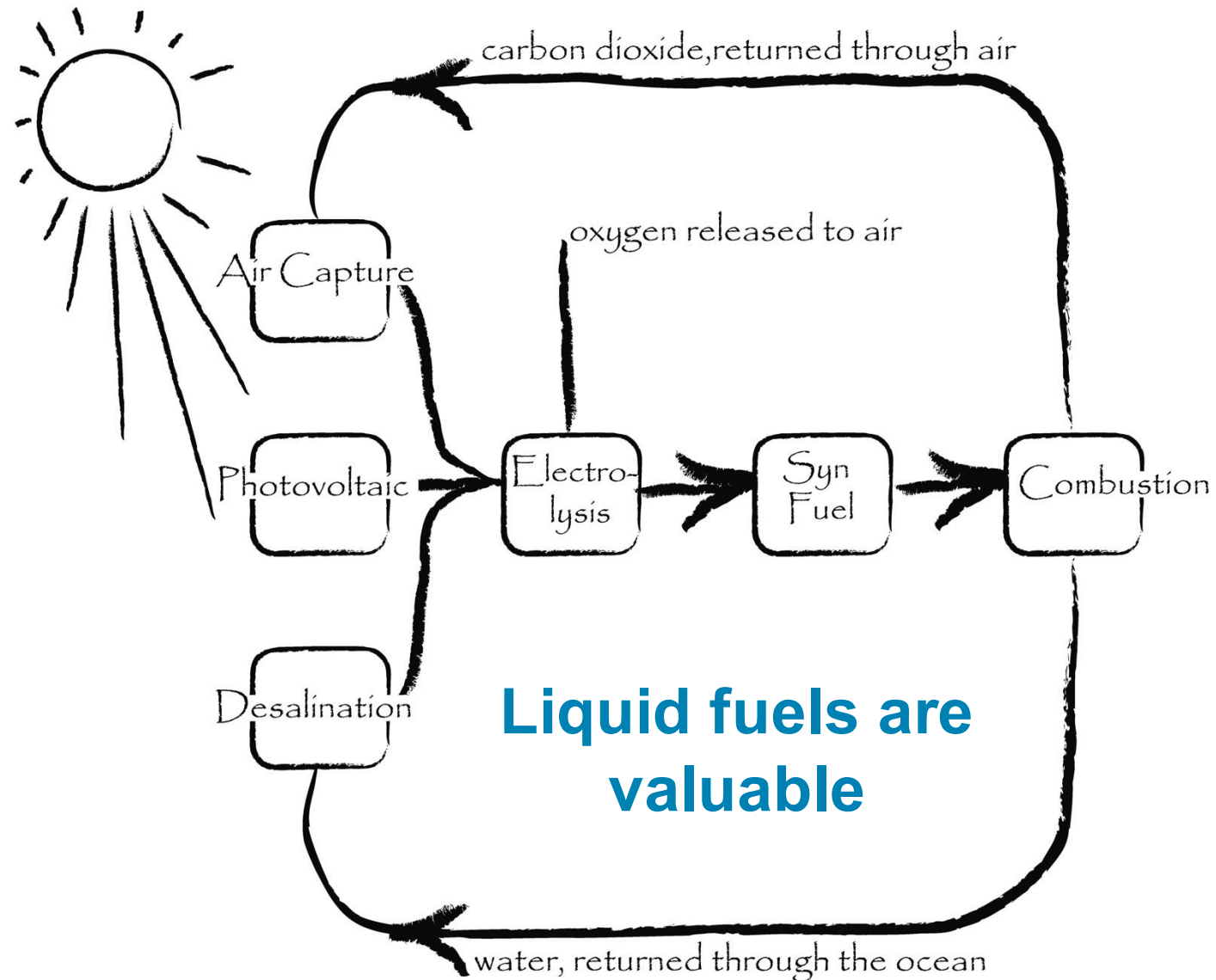


Pixabay stock

- It is the cheapest source of electricity
- Its energy can be stored in liquid fuels
- Cheap storage and easy transport



# Carbon emissions and fossil fuels can separate



# Technology usually succeeds if asked to deliver



- The first aircela machine
  - Taking CO<sub>2</sub> from air, water and electricity to make gasoline
  - Designed for Photovoltaic Power