



OxyFuel Combustion Technologies

Prof. Greg Jackson

Dept. of Mechanical Engineering
Colorado School of Mines
Golden, CO 80401

CCUS Student Week
October 17, 2018

Outline of presentation

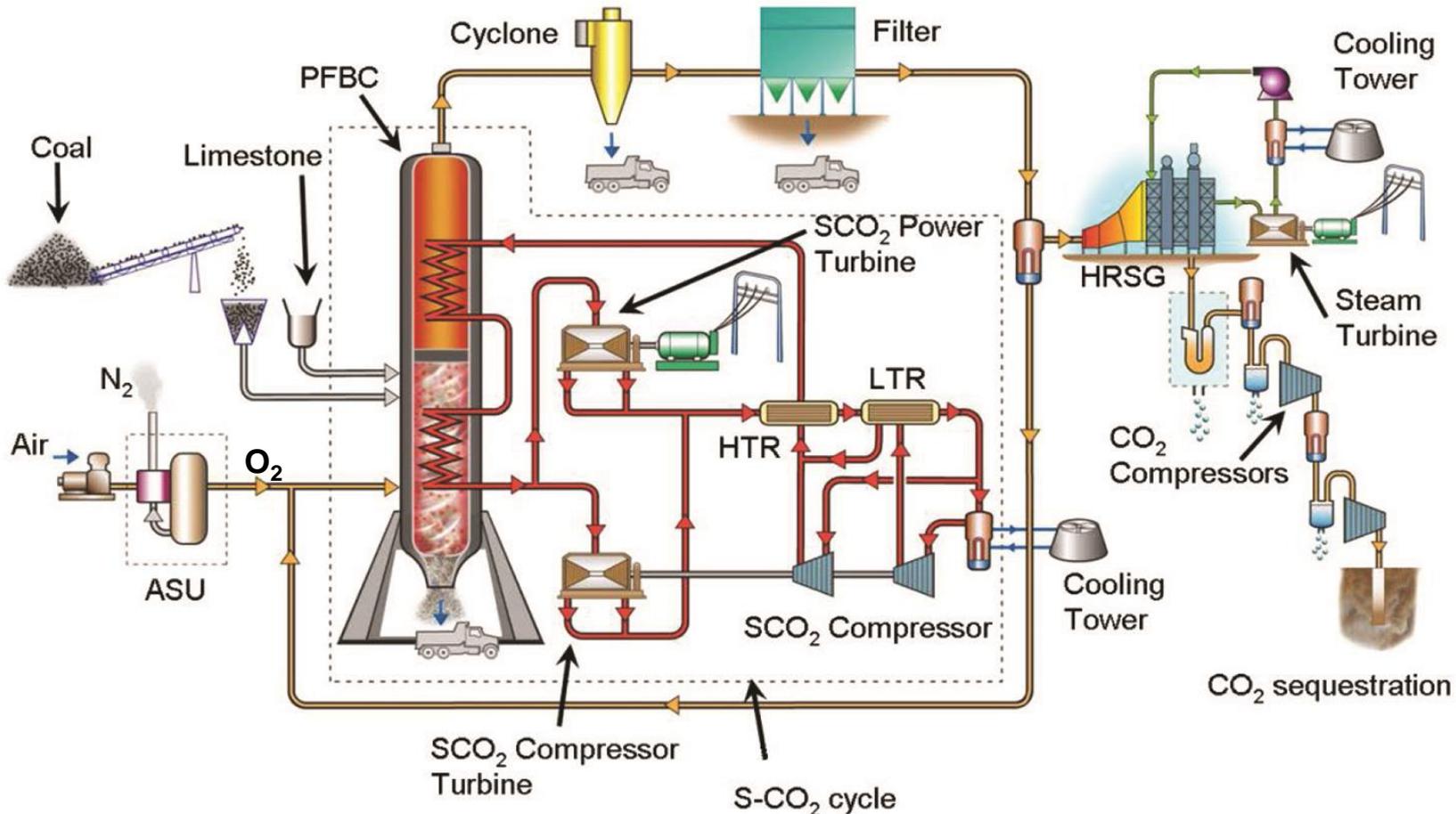


- **Current Oxyfuel Combustion**
- Oxygen-ion Transport Membranes for Air Separation
- High-Temperature Fuel Cells
- Chemical looping Combustion

Oxyfuel combustion of coal with closed Brayton supercritical-CO₂ power cycles and CO₂ capture



- U.S. DOE NETL funded efforts to design oxyfuel combustion with high efficiency s-CO₂ cycles and bottoming steam cycle and CO₂ capture



Penalty of conventional oxyfuel combustion



- Comparison of plant generating efficiency and capital expenditure of CO₂ capture technologies from Chen, Yong, & Ghoneim (2012).

PC: Conventional pulverized coal with supercritical steam cycle without CO₂ capture

Post: PC with post CO₂ capture,

A-Oxyf: Atmospheric oxy-coal with flue gas recycle and CO₂ capture

P-oxyf: Pressurized oxy-fuel combustor with flue gas recycle and CO₂ capture,

■ Capital cost
—●— Generating efficiency

Capital cost (\$/kWe)

Generating efficiency (%)
- 10

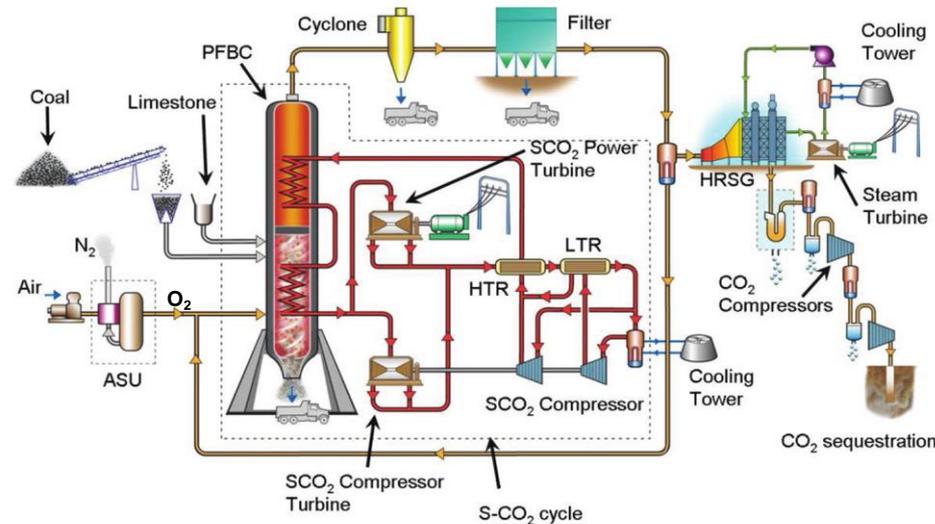
Oxyfuel combustion of coal with closed Brayton supercritical-CO₂ power cycles and CO₂ capture



Earth • Energy • Environment

Colorado School of Mines

- Advantages of oxyfuel combustion
 - Technology largely available
 - Air separation unit
 - Pulverized fluidized bed combustor
 - CO₂/H₂O flue gas separation
 - Combustor coupling to a variety of high efficiency power plants
 - Supercritical steam
 - Developing supercritical CO₂
 - Minimal emissions
 - Almost no NO_x emissions
 - Ease of flue-gas processing
 - Not sensitive to high-purity O₂



- Disadvantages
 - LCOE costs high ($\geq 10\text{¢/kWh}$)
 - Temperature distributions in the boiler

Outline of presentation



- Oxyfuel Combustion
- **Oxygen-ion Transport Membranes for Air Separation**
- High-Temperature Fuel Cells
- Chemical looping Combustion

Oxygen-transport membranes (OTMs) as efficient air separation units to replace cryogenic units



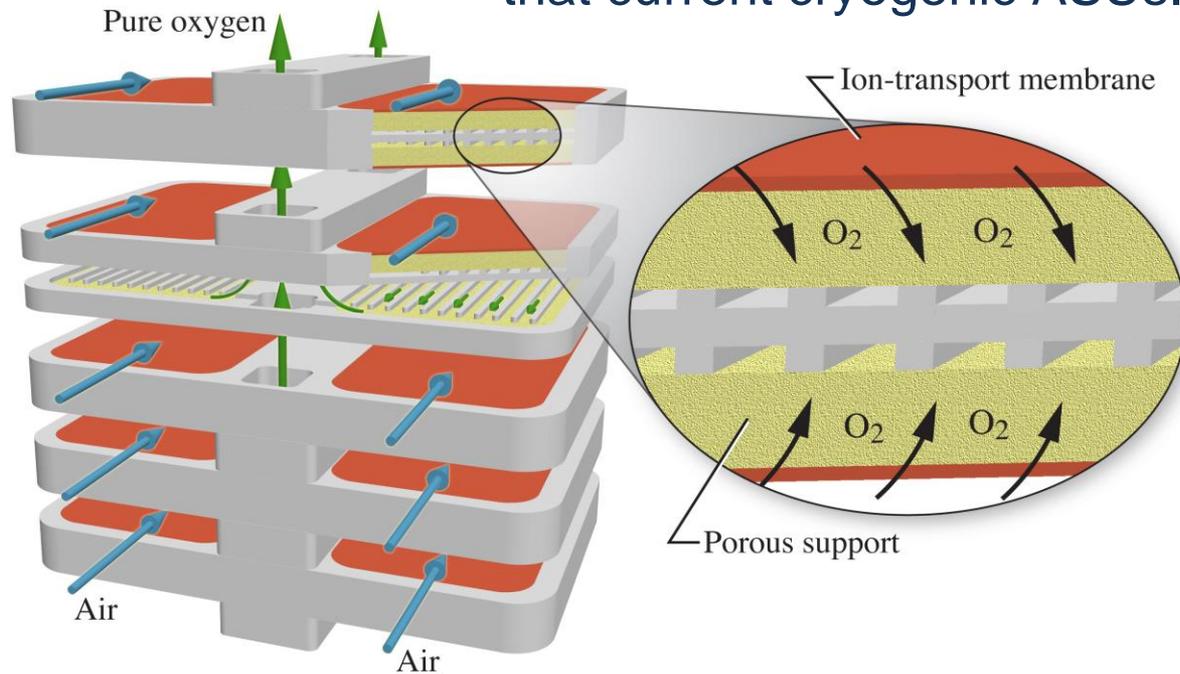
Earth • Energy • Environment

Colorado School of Mines



Air separation unit

- O_2 transport membranes that rely on P_{O_2} gradients across ceramic ion-conduction membranes offer potential for large cost reduction and faster times that current cryogenic ASUs.



- Large gas companies have invested in this technology but challenges remain.

Scaled-up OTMs for air-separation with failures in commercial service environments



Earth • Energy • Environment

Colorado School of Mines



LCF-based
air-separation system

AIR
PRODUCTS 

 **CERAMATEC**
TOMORROW'S CERAMIC SYSTEMS



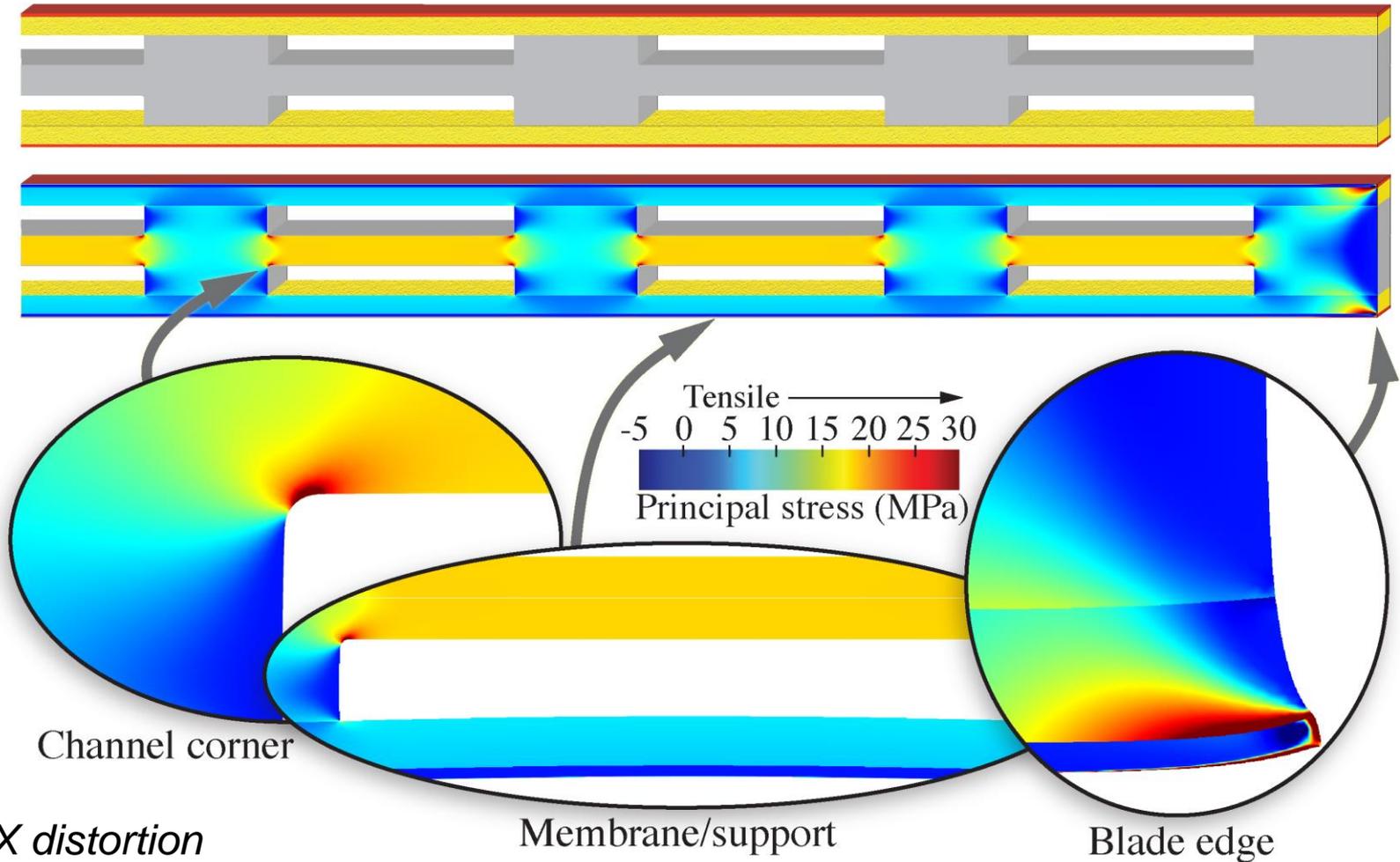
Anderson, Armstrong, Broekhuis, Carolan, Chen, Hutcheon, Lewinsohn, Miller, Repasky, Taylor, Woods,
“Advances in ion transport membrane technology for oxygen and syngas production,”
Solid State Ionics, 288:331-337 (2016)

Chemo-mechanical coupling significant local stresses leading to failures in OTMs



Earth • Energy • Environment

Colorado School of Mines



$250 \times$ distortion

Membrane/support

Blade edge

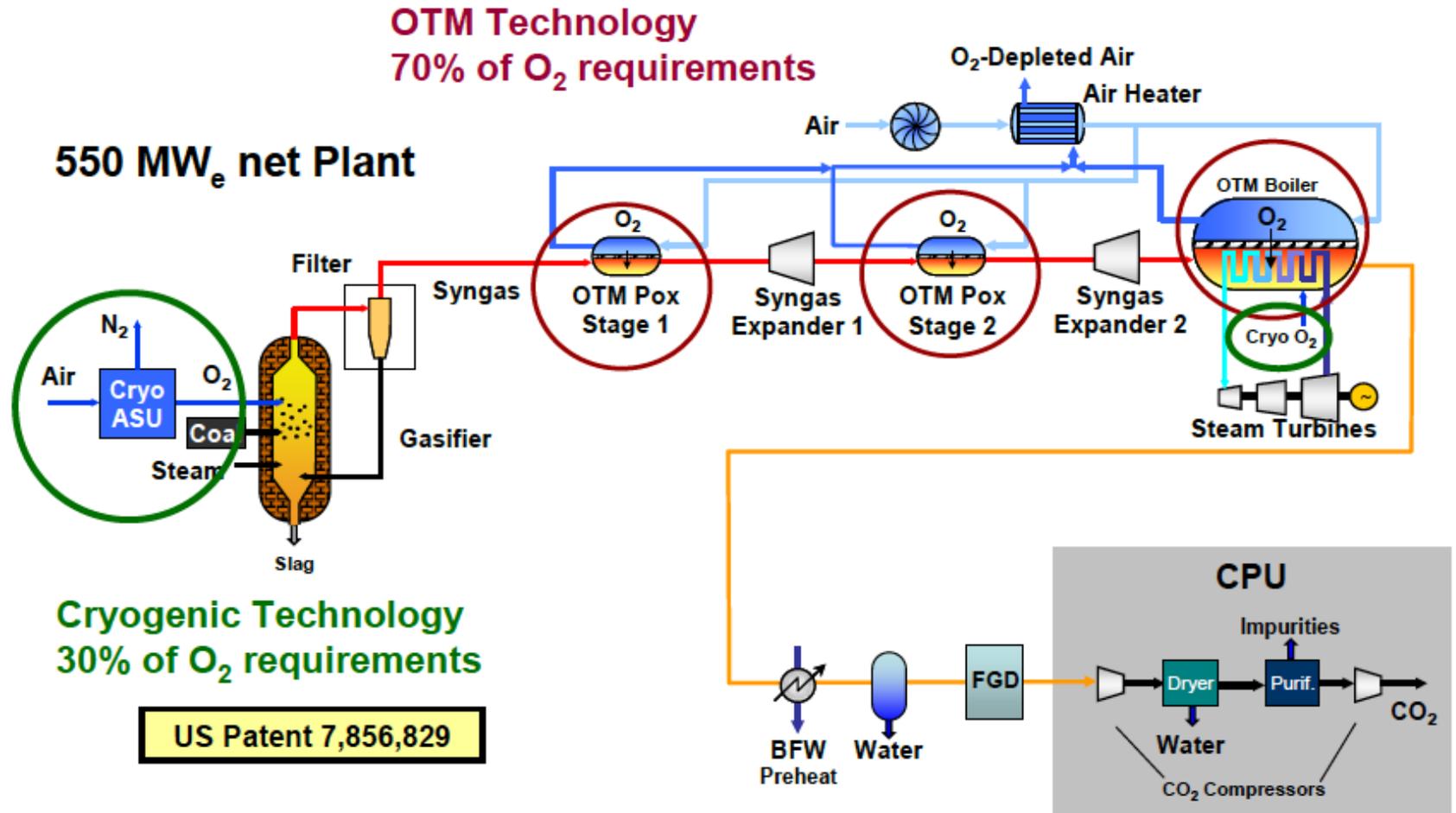
From Euser, Zhu Berger, Zhu, Lewinsohn & Kee (Mines) 2017

OTMs have been proposed for many aspects of a zero emission combustion plant



Earth • Energy • Environment

Colorado School of Mines



From Shah, Jamal, Drenvich et al (Praxair) 2010

Outline of presentation

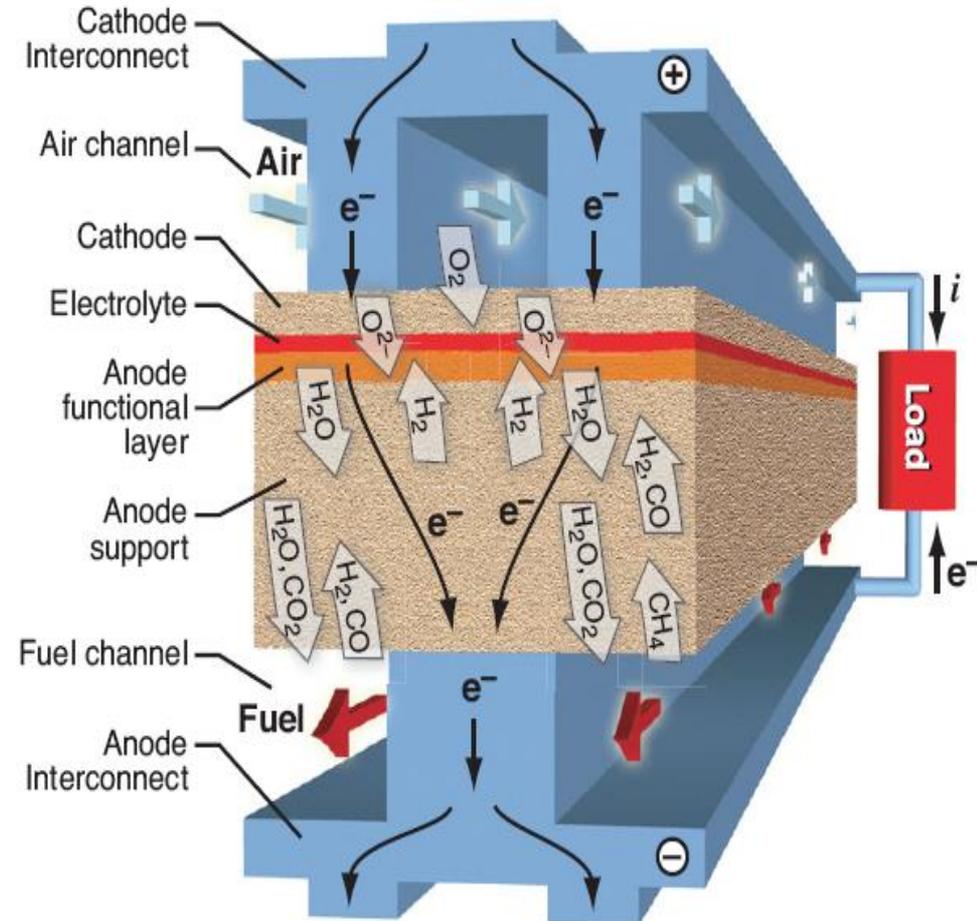


- Oxyfuel Combustion
- Oxygen-ion Transport Membranes for Air Separation
- **High-Temperature Fuel Cells**
- Chemical looping Combustion

Solid oxide fuel cells (SOFCs) as an oxy-fuel combustion with direct electrochemical conversion



- Solid oxide fuel cells (SOFCs) provide a lot of benefits for fossil fuel conversion.
- High potential for CO₂ capture
- Thin (10-20 μm) O²⁻ conducting electrolytes provide oxidizer from air-fed cathode to fuel-fed anode
- To date, distributed power SOFCs have costs >>\$2000 kW_{elec} due in part to low production volumes but also to high energy ceramic processing.



from Kee, Zhu, Sukeshini, and Jackson, Combust. Sci. Tech. 2008

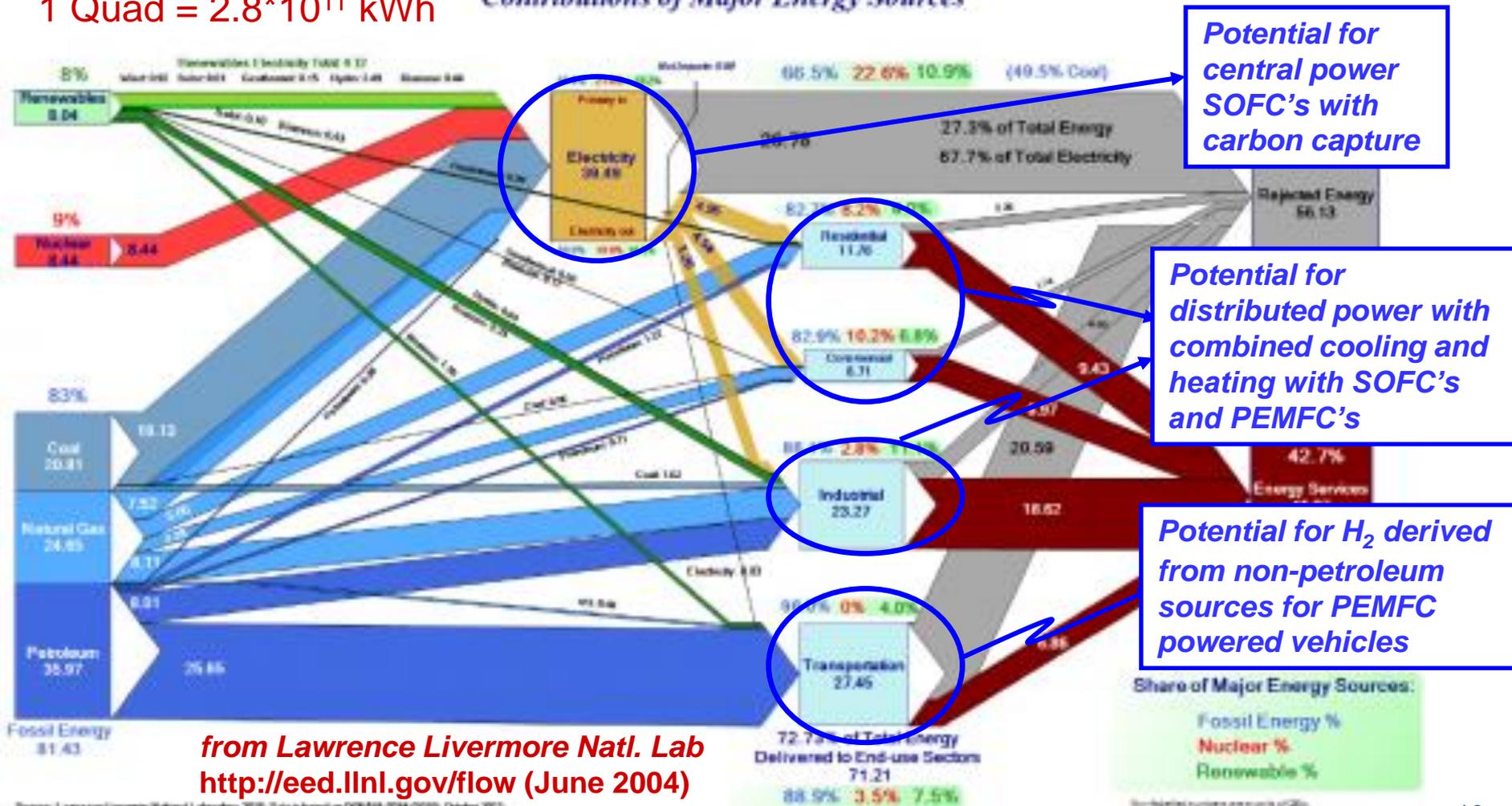
Fuel cells as a potential energy conversion technology to assist in carbon capture



Estimated U.S. Energy Use in 2010: 98.0 Quads

Contributions of Major Energy Sources

1 Quad = 2.8×10^{11} kWh



Potential for central power SOFC's with carbon capture

Potential for distributed power with combined cooling and heating with SOFC's and PEMFC's

Potential for H₂ derived from non-petroleum sources for PEMFC powered vehicles

from Lawrence Livermore Natl. Lab
<http://eed.llnl.gov/flow> (June 2004)

Source: Lawrence Livermore National Laboratory, 2010. Data is based on DOE/EIA 2010 (October 2010). Reorganized to separate and associate flows to major energy sources.

SOFCs for distributed power applications



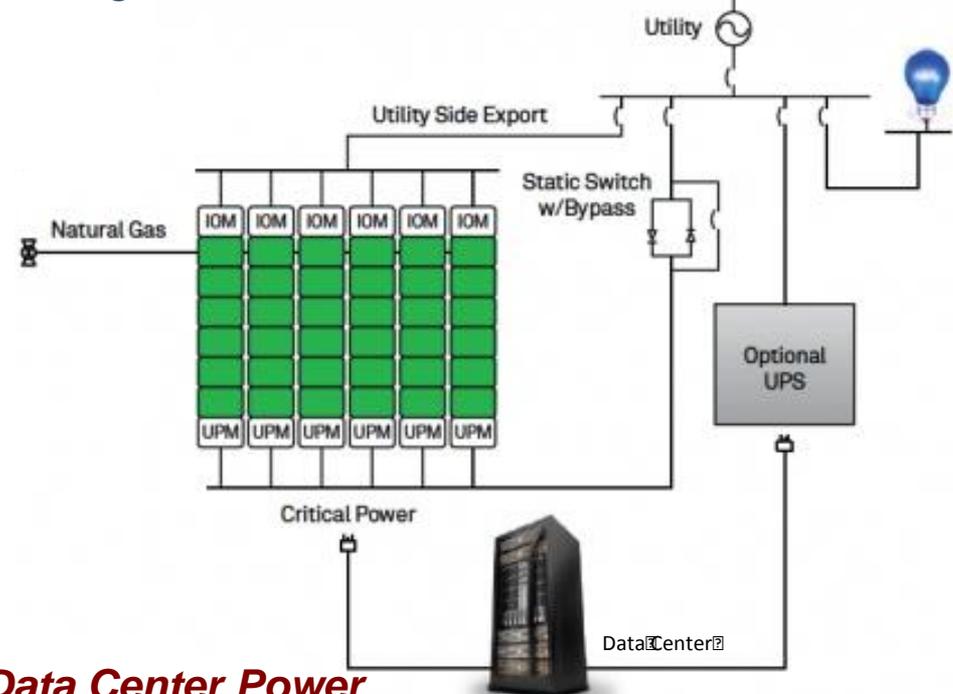
Earth • Energy • Environment

Colorado School of Mines



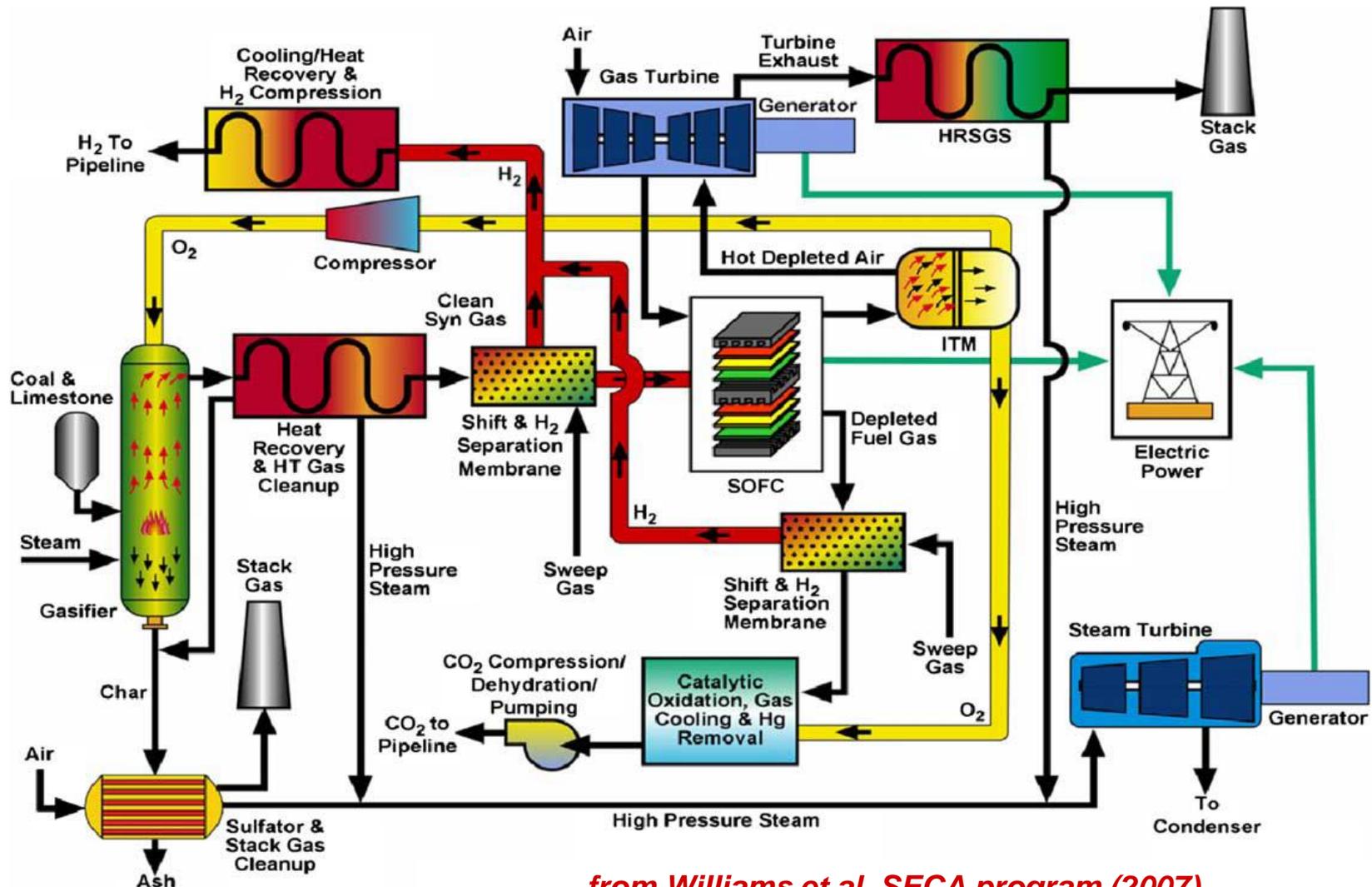
- If NG costs $\sim \$0.035/\text{kW}_{\text{th}}$ and electricity costs $\sim \$0.10/\text{kW}_{\text{elec}}$, a payback period of 20 years would exist for a 55% efficient SOFC at $\$8000 \text{ kW}_{\text{elec}}$.
- Economics are highly sensitive to gas and electricity costs and make value proposition region specific.

- Bloom Energy supplies 125 & 250 kW_{elec} natural-gas fueled SOFC systems that run at 50 – 60% efficiency but at costs of $> \$7000 \text{ kW}_{\text{elec}}$ (in comparison to $\$1000 \text{ kW}_{\text{elec}}$ for diesel with 40-50% efficiency).
- High maintenance costs raise concerns.



**Data Center Power
from Bloom Energy**

Early proposals for SOFCs on central coal power with gas-turbine pressurization and bottoming cycle

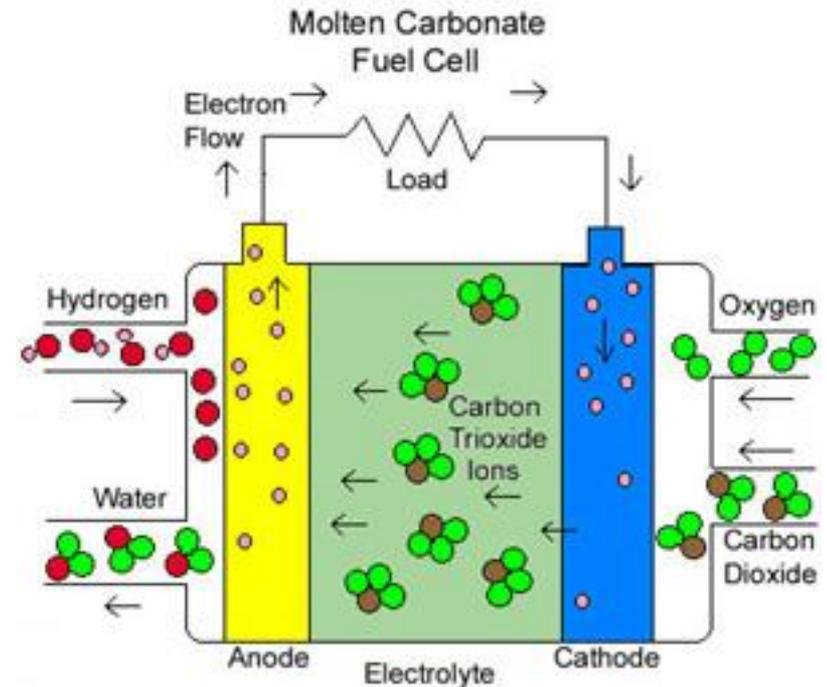


from Williams et al. SECA program (2007)

Molten carbonate fuel cells as a means of carbon capture



- High-temperature fuel cells such as molten carbonate fuel cells (MCFC) require large capital investments for both development and plant assembly.
- Thus, systems are relatively large for fuel cells ($> 300 \text{ kW}$) and typically not cheap ($\sim \$1400 \text{ kW}_{\text{elec}}$).
- World leaders Fuel Cell Energy are showing, high-T fuel cells like MCFC can be more than power generation.
- Opportunities for combined heat, power, and hydrogen show the value in fuel cells converting low-cost gas into value-added products.

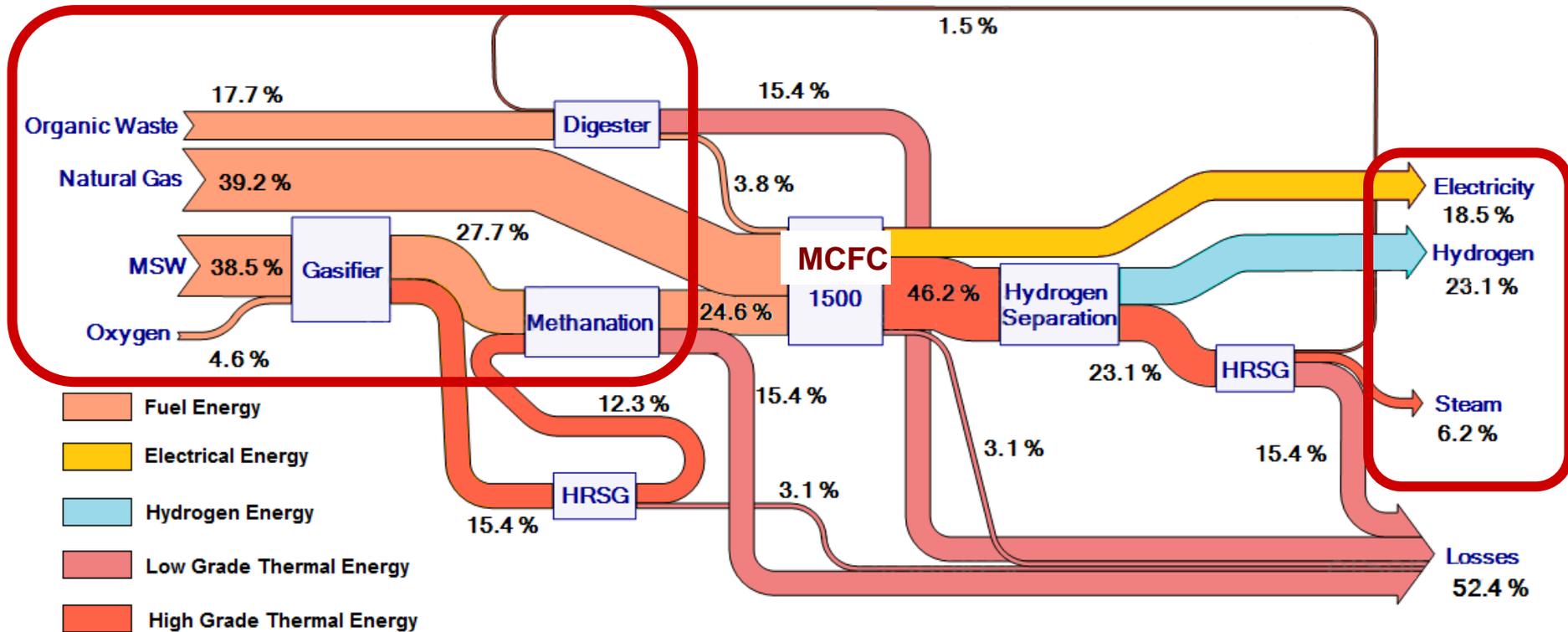


from NPTEL Fuel Cell Course
<http://nptel.ac.in/courses/103102015/3>

Molten carbonate fuel cell opportunities for multi-functional distributed power plants



- MCFC can add much more value such as conversion of waste to relatively clean energy, heat and H₂ (CHHP).



from Spencer, Moton, Gibbons, et. al 2013

Outline of presentation



- Oxyfuel Combustion
- Oxygen-ion Transport Membranes for Air Separation
- High-Temperature Fuel Cells
- **Chemical looping Combustion**

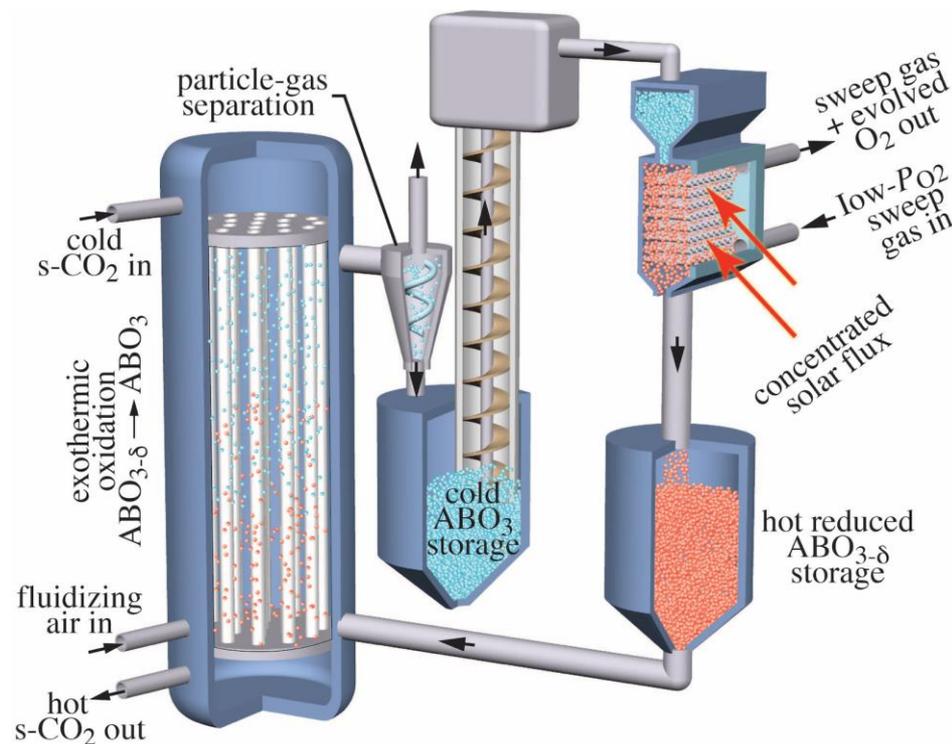
Chemical looping can be coupled to thermochemical energy storage



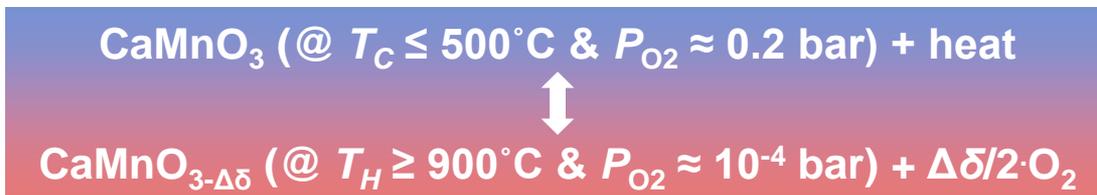
Earth • Energy • Environment

Colorado School of Mines

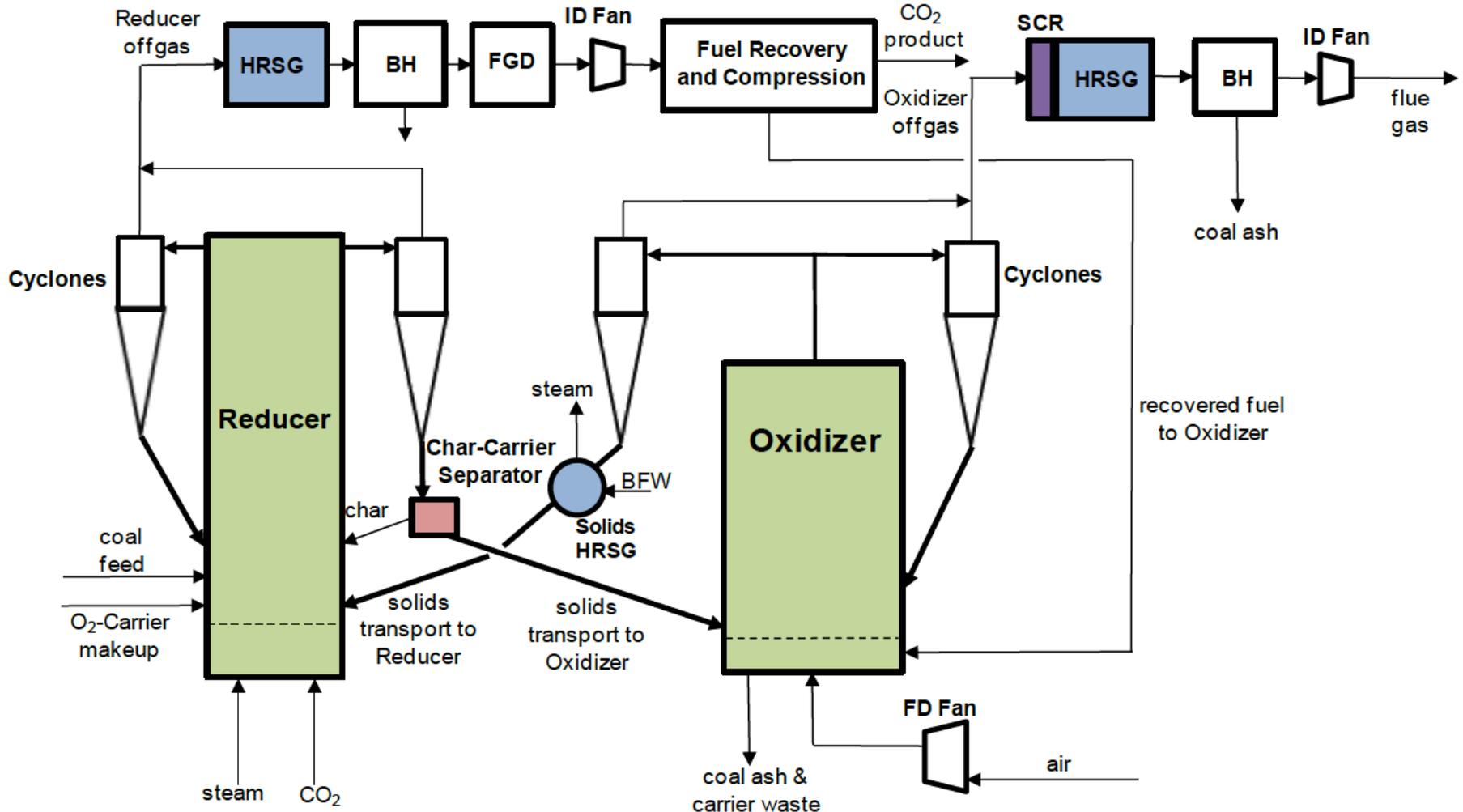
- Reducible perovskite oxides such as CaMnO_3 can provide thermochemical energy storage (TCES) and controlled heat release at temperatures T above 800°C .
- High chemical reduction occurs at O_2 partial pressures, $P_{\text{O}_2} \geq 10^{-4}$ bar.
- Re-oxidation and cooling can be sustained at T relevant for driving supercritical CO_2 power cycles.
- The challenge is how to accelerate kinetics for high specific TCES.



*from Imponenti et al. 2017,
Albrecht et al. 2018*



Chemical looping combustion integrated with gas processing for CO₂ sequestration



Chemical looping combustion materials and potential costs



- NETL-funded studies on two common chemical looping materials have shown cost performance superior to conventional pulverized bubbling bed reactor with oxyfuel combustion

Oxygen Carrier Type	Fe ₂ O ₃	CaSO ₄	Conventional PC BBR Case 12
Plant Net Capacity (MW)	550	550	550
Plant Efficiency (% HHV)	35.1	32.6	28.4
Carbon Capture Efficiency (%)	95.8	91.4	90
CO ₂ Product Purity (mole% CO ₂)	98.9	99.7	100
Total Plant Cost (\$/kW; \$2011)	2,379	2,597	3,563
Cost of Electricity (\$/MWh; 1 st -year w/o T&S)	115.2	104.7	137.3
Reduction in COE (%) [Reference IGCC w CCS @ 133 \$/MWh]	13.4	21.3	-3.2
Cost of Captured CO ₂ (\$/tonne) [Reference SC PC plant @ 81.0 \$/MWh]	40.1	26.8	56.5

Particle materials for chemical looping



Earth • Energy • Environment

Colorado School of Mines

- Multivalent cations such as Fe^{x+} and Mn^{x+} provide promising potential for chemical looping with $\text{CO}_2/\text{H}_2\text{O}$ effluent for ease of carbon capture.

Material	ρ [kg/m ³]	$C_{p,avg}$ [kJ/kg-K]	$\Delta h_{500-800}$ [kJ/kg]	$\Delta h_{500-900}$ [kJ/kg]	Cost [\$/kg]	Cost ^b [\$/kWh]
Co_3O_4	6110	0.96	1125 ^b	1226	≈ 27.0	≈ 86
$\text{Mn}_{1.8}\text{Fe}_{0.2}\text{O}_3$	4630	0.88	219	620	< 1.5	< 12
$\text{Ca}_{0.9}\text{Sr}_{0.1}\text{MnO}_{3-\delta}$	4530	0.86	524 ^b	706 ^a	<2.0	<21

a) equilibrated at $P_{\text{O}_2} = 0.0001$ bar

b) Based on thermal energy stored from 500 to 750°C

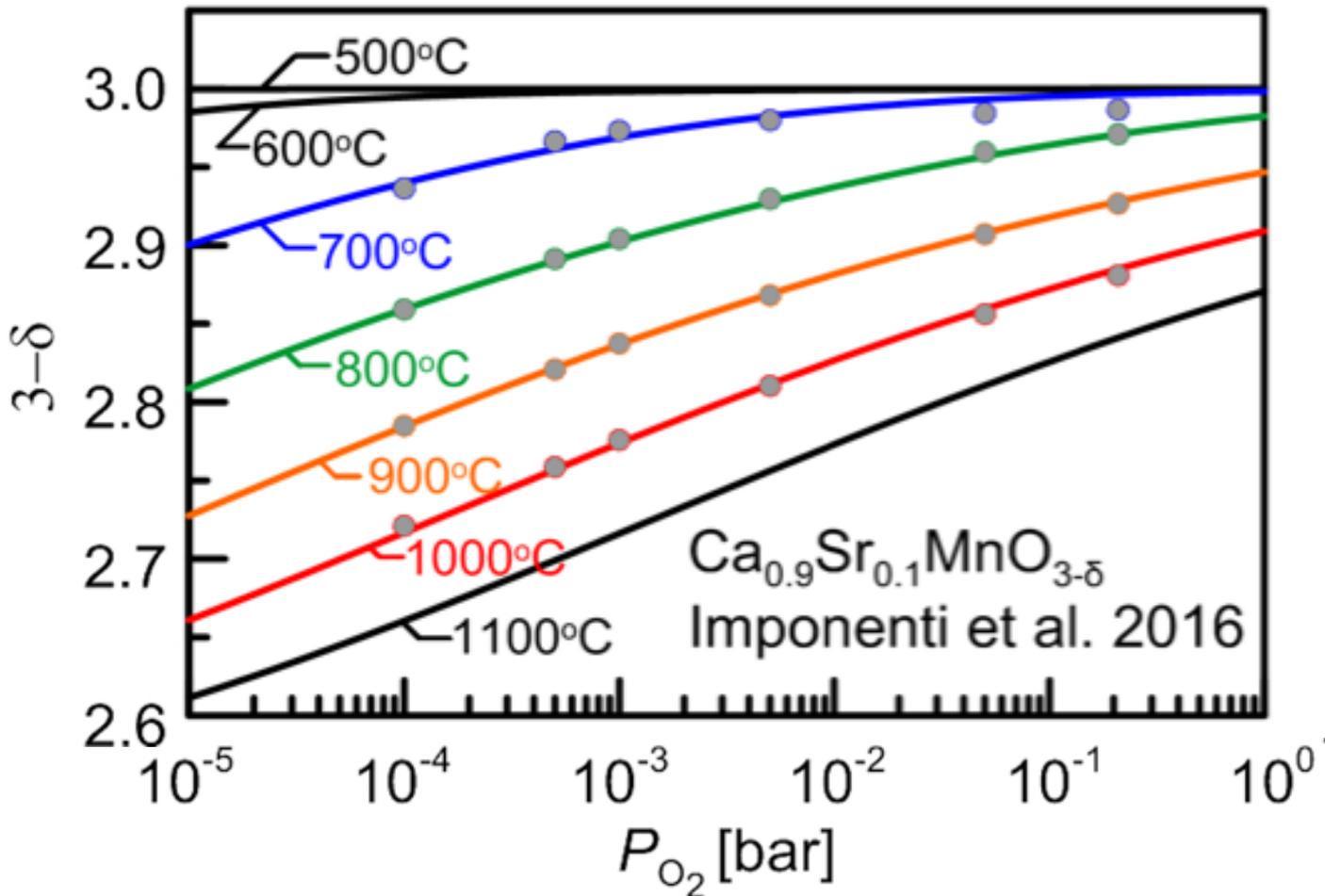
- European researchers have identified doped $\text{CaMnO}_{3-\delta}$ as a promising chemical looping oxygen carrier.

Compositions	kg oxide / kg fuel]	T_H [°C]	P_{O_2} [bar]	Δh_{R2} [kJ/kg]
$\text{Ca}_{0.90}\text{Sr}_{0.10}\text{MnO}_{3-\delta}$ + C_nH_{2n} fuel	52.2	920	5e-6	835
	63.2	830	3e-6	690
	79.0	750	706	552

Cost-effective doped $\text{CaMnO}_{3-\delta}$ for reversible release and uptake oxygen in chemical looping systems



- Thermochemistry modeled by two-step point-defect reaction mechanism:



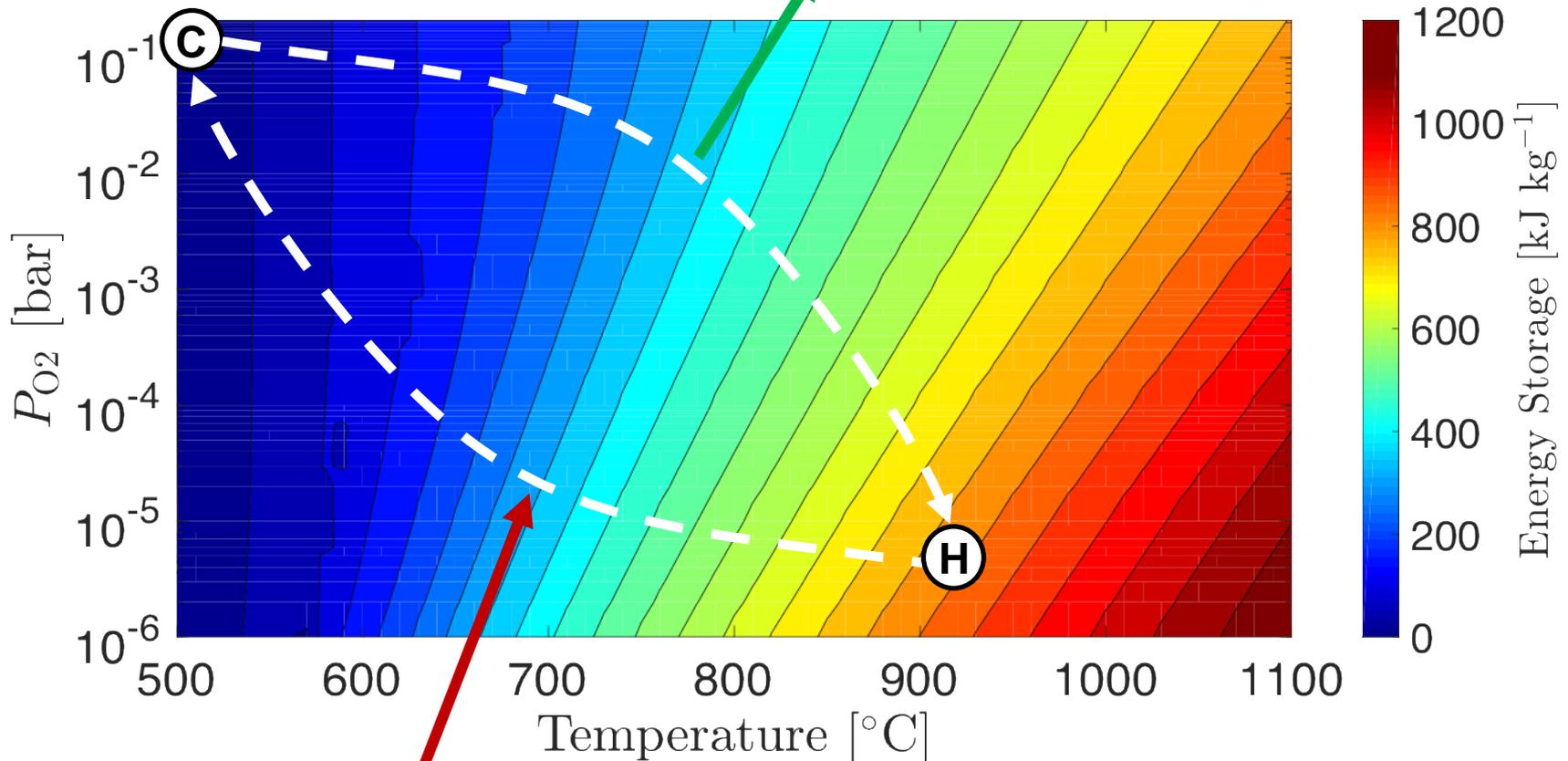
Thermodynamics of chemical looping combustion with Sr-doped $\text{CaMnO}_{3-\delta}$



Earth • Energy • Environment

Colorado School of Mines

➤ Specific energy stored during reduction and heating from $T_C = 500^\circ\text{C}$ in air



Particle materials for chemical looping



Earth • Energy • Environment

Colorado School of Mines

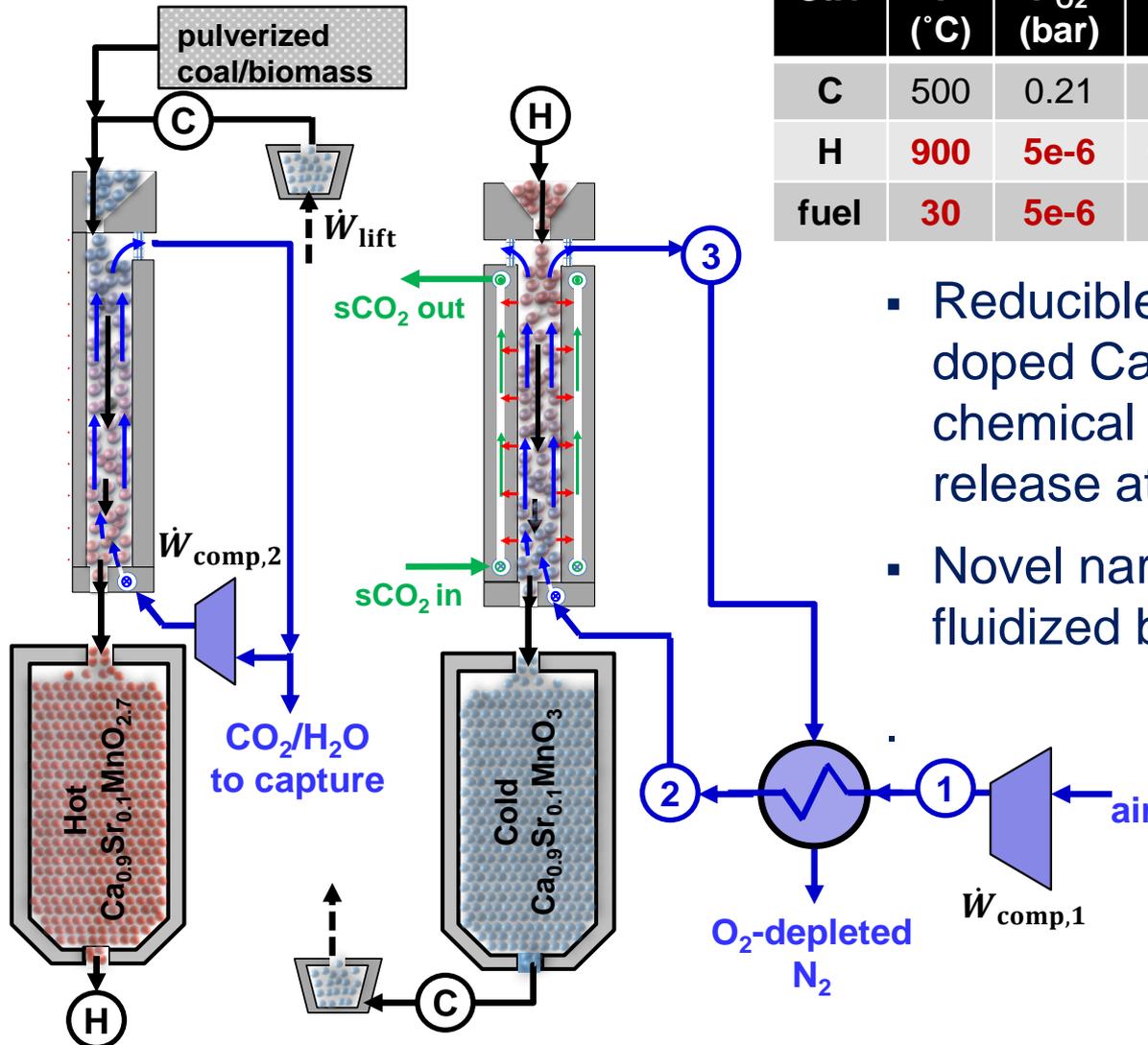
Compositions	kg oxide / kg fuel]	T_H [°C]	P_{O_2} [bar]	Δh_{R2} [kJ/kg]
$Ca_{0.90}Sr_{0.10}MnO_{3-\delta}$ + C_nH_{2n} fuel	52.2	920	5e-6	835
	63.2	830	3e-6	690
	79.0	750	706	552

Proposed Perovskite Chemical Looping Combustion Subsystem for 100 MW_{elec} plant



Earth • Energy • Environment

Colorado School of Mines



St.1	T (°C)	P _{O₂} (bar)	Stoich.	Δh _{tot} (kJ/kg)	Flow (kg/s)
C	500	0.21	Ca _{0.9} Sr _{0.1} MnO ₃	0.0	239.5
H	900	5e-6	Ca _{0.9} Sr _{0.1} MnO _{2.7}	835	232.1
fuel	30	5e-6	C _n H _{2n}		4.6

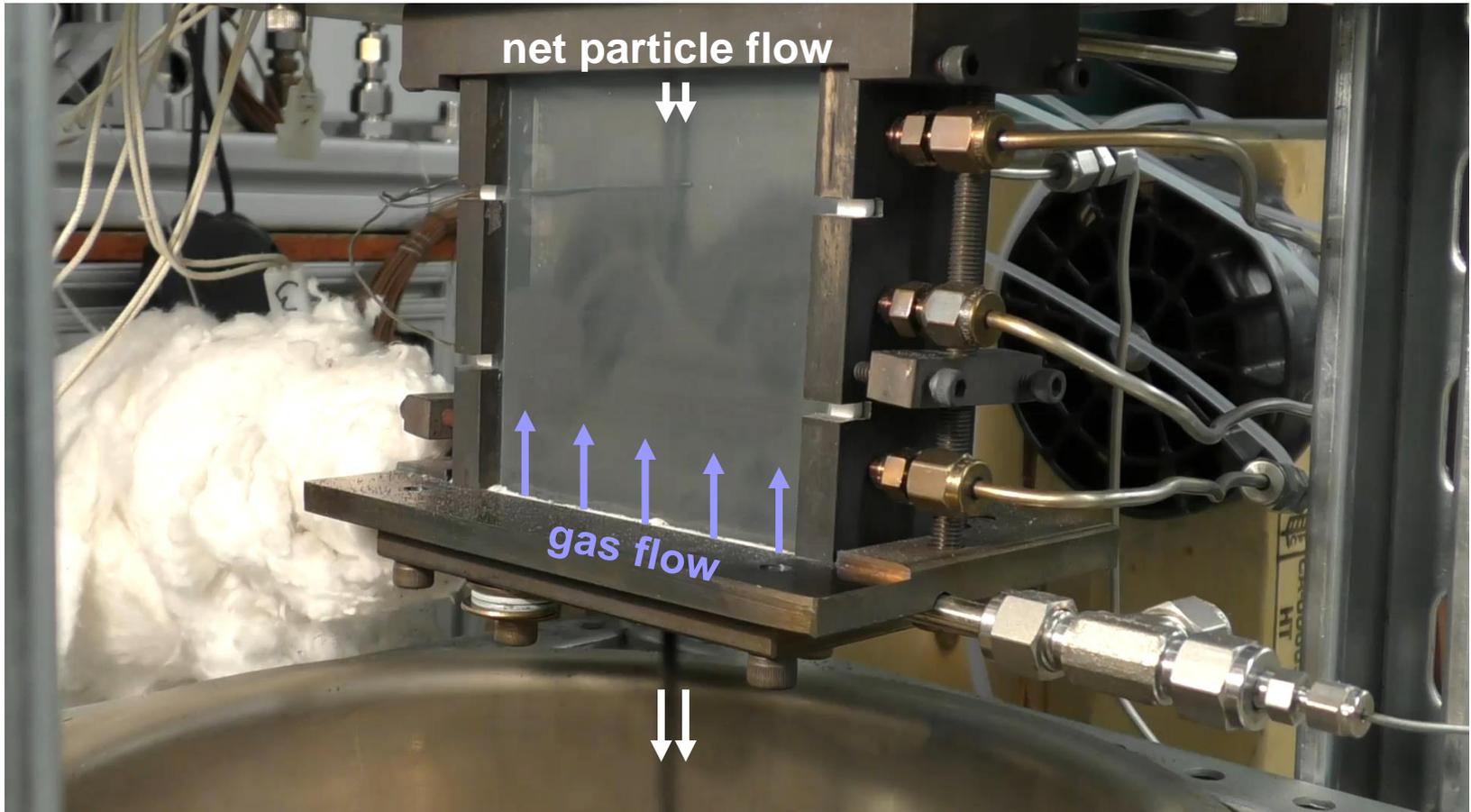
- Reducible perovskite oxides such as doped CaMnO₃ can provide thermochemical energy storage for heat release at off peak hours.
- Novel narrow-channel, counterflow fluidized beds offer modular system.

Visualization of narrow-channel, counterflow fluidized bed



Earth • Energy • Environment

Colorado School of Mines



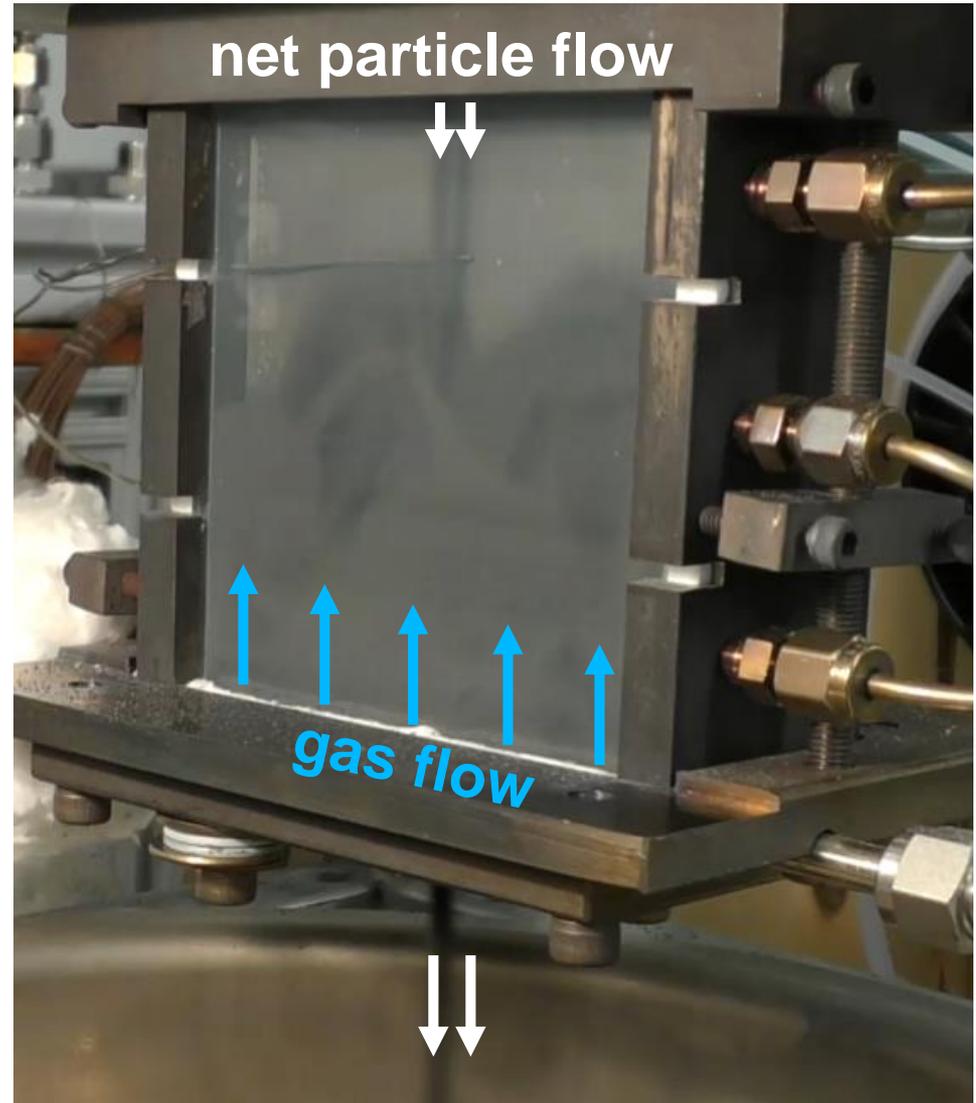
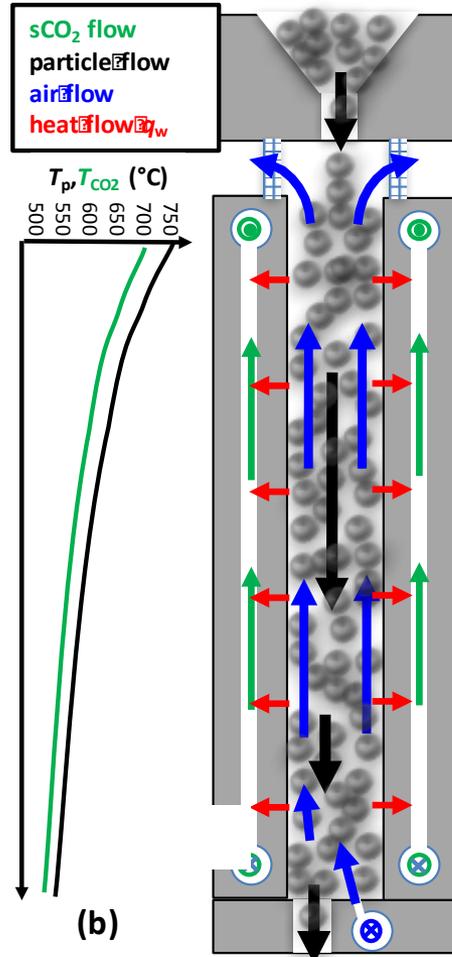
from Miller, Pfutzner, & Jackson 2018

Novel modular heat recovery devices to make coal, waste and biomass combustion more modular



Earth • Energy • Environment

Colorado School of Mines



Thoughts for further exploration



- Current oxyfuel combustion technologies with traditional air-separation units and steam power plants do not make a compelling technical or economic cases.
 - Game changers include supercritical-CO₂ power cycles and oxygen transport membranes for lower cost and more flexible O₂ separation.
- Solid oxide fuel cells provide an novel approach to achieving very high efficiencies with carbon capture, but their costs do not scale and applications seem currently limited to distributed power applications.
- Chemical looping seems like a promising approach to combust pulverized coal, biomass, and waste.
 - High mass of reduced oxide can serve as thermochemical energy storage
 - Novel fluidized bed designs for energy transfer to the power cycle and fluidized combustor here may enable modular designs.

Acknowledgements



- Solid oxide fuel cell work and oxygen transport membrane work was supported by Prof. Robert Kee, Prof. John Berger Dr. Huayang Zhu, and former student Bryan Euser.
- Perovskite redox cycles based on particle fluidized beds have been supported by students Luca Imponenti, Dan Miller, and Rounak Kharait, and former student Dr. Kevin Albrecht, and Profs. Rob Braun and Ryan O'Hayre.
- Collaborations with National Renewable Lab (Dr. Zhiwen Ma, Dr. Janna Martinek, and Judy Netter) have been helpful in modeling and demonstrating a test receiver design.
- Our team acknowledges the support of the Department of Energy through the SunShot Initiative under the CSP:ELEMENTS program.
- Thanks to Brent Kinson, Devin Clay, Steven Landin, and Frank Anderson at CoorsTek for the engineering efforts and time dedicated to perovskite synthesis for this project.



References (highlighted from Mines)



- K.J. Albrecht, G.S. Jackson, R.J. Braun, (2018), *Solar Energy*, **167**, 179–193.
- K.J. Albrecht, G.S. Jackson, R.J. Braun (2016), *Applied Energy*, **165**, 285-296.
- E. Bakken, T. Norby, S. Stølen (2005), *Solid State Ionics*, 176(1-2), 217-223.
- L. Imponenti, K.J. Albrecht, J.W. Wands, M.D. Sanders, G.S. Jackson (2017), *Solar Energy*, **151**, 1-13.
- L. Imponenti, K.J. Albrecht, R.J. Braun, G.S. Jackson (2016), *ECS Transactions*, **72(7)**, 11-22.
- L. Imponenti, K.J. Albrecht, R. Kharait, M.D. Sanders, G.S. Jackson (2018), *Applied Energy*, in review for publication.
- E.I. Leodinova, I.A. Leodinov, M.V. Patrakeev, and V.L. Kozhevnikov (2011). *J. Solid State Electrochem.*, **15**, 1071-1075.
- D.C. Miller, C.J. Pfutzner, G.S. Jackson (2018), *Intl. J. Heat and Mass Transfer*, **126**, 730-745.
- C.K. Ho, J.M. Christian, D. Romano, J. Yellowhair, N. Siegel, L. Savodi, R. Zanino (2016), *J. of Solar Energy Engineering – Trans. ASME*, **139** #021011-1.
- H. Zhang, H. Benoit, D. Gauthier, J. Degève, J. Baeyens, I. Pérez López, M, Hemati, G. Flamant (2016), *Applied Energy*, **161**, 206-224.
- J. Martinek, Z. Ma (2015), *J. of Solar Energy Engineering – Trans. ASME*, **137** # 051008-1.